AD-A129 031	REFINING VOLUME I H R MOORI	OF MILITA 1 (IN SITU E ET AL. M	ARY JET FU J(U) ASH MAR 82 AFW	ELS FROI LAND PE AL-TR-8	M SHALĒ TROLEUM 1-2056-1	OIL PA CO KY VOL-2-P	RT 11 T-2	1/ 5		
UNCLASSIFIED	F 33615-78	8-C-2080				F/G	21/4 .	NL		
		= =	- H							
			=							



MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

AFWAL-TR-81-2056 Part II, Vol II





Part II, Vol II
In Situ Shale Oil Process Data

H. R. Moore

L. M. Henton

C. A. Johnson

D. A. Fabry

3

AD

ASHLAND PETROLEUM COMPANY
ASHLAND RESEARCH AND DEVELOPMENT
ASHLAND, KENTUCKY 41101

MARCH 1982

INTERIM TECHNICAL REPORT FOR PERIOD June 1979 - October 1980

Approved for Public Release; Distribution Unlimited

AERO PROPULSION LABORATORY
AIR FORCE WRIGHT AERONAUTICAL LABORATORIES
AIR FORCE SYSTEMS COMMAND
WRIGHT-PATTERSON AIR FORCE BASE, OHIO 45433



DIC FILE CO

#### NOTICE

When Government drawings, specifications, or other data are used for any purpose other than in connection with a definitely related Government procurement operation, the United States Government thereby incurs no responsibility nor any obligation whatsoever; and the fact that the government may have formulated, furnished, or in any way supplied the said drawings, specifications, or other data, is not to be regarded by implication or otherwise as in any manner licensing the holder or any other person or corporation, or conveying any rights or permission to manufacture use, or sell any patented invention that may in any way be related thereto.

This report has been reviewed by the Office of Public Affairs (ASD/PA) and is releaseable to the National Technical Information Service (NTIS). At NTIS, it will be available to the general public, including foreign nations.

This technical report has been reviewed and is approved for publication.

Charlotte R. Eigel CHARLOTTE R. EIGEL

Fuels Branch, Fuels and Lubrication Division Aero Propulsion Laboratory ARTHUR V. CHURCHILL

Chief, Fuels Branch Fuels and Lubrication Division Aero Propulsion Laboratory

ROBERT D. SHERRILL

Chief, Fuels and Lubrication Division

Aero Propulsion Laboratory

If your address has changed, if you wish to be removed from our mailing list, or if the addressee is no longer employed by your organization please notify <u>AFWAL/POSF</u>, W-PAFB OH 45433 to help us maintain a current mailing list.

Copies of this report should not be returned unless return is required by security considerations, contractual obligations, or notice on a specific document.

SECURITY CLASSIFICATION OF THIS PAGE (When Date Entered)

REPORT DOCUMENTATION PAGE	READ INSTRUCTIONS BEFORE COMPLETING FORM
	3. RECIPIENT'S CATALOG NUMBER
AFWAL-TR-81-2056, Part II, Vol II AD AIQC	
4. TITLE (and Subtitle)	Interim Report for Period
REFINING OF MILITARY JET FUELS FROM SHALE OIL PART II, VOL II	June 1979 - October 1980
(In Situ Shale Oil Process Data)	6. PERFORMING ORG. REPORT NUMBER
7. AUTHOR(*) H. R. Moore D. A. Fabry	S. CONTRACT OR GRANT NUMBER(s)
H. R. Moore D. A. Fabry L. M. Henton	F33615-78-C-2080
C. A. Johnson	
9. PERFORMING ORGANIZATION NAME AND ADDRESS	10. PROGRAM ELEMENT, PROJECT, TASK
Ashland Petroleum Company	PE 63215F
Ashland Research and Development	WU 24800004
P.O. Box 391, Ashland KY 41101	
11. CONTROLLING OFFICE NAME AND ADDRESS	12. REPORT DATE
Aero Propulsion Laboratory (AFWAL/POSF)	March 1982
Air Force Wright Aeronautical Labs AFSC Wright-Patterson AFB. Ohio 45433	13. NUMBER OF PAGES 287
14. MONITORING AGENCY NAME & ADDRESS(If different from Controlling Office)	15. SECURITY CLASS, (of this report)
	Unclassified
, ·	15a. DECLASSIFICATION/DOWNGRADING
	SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report)	<u></u>
Approved for public release; distribution unlimited.	•
$\{$	
. 17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from	en Report)
19. SUPPLEMENTARY NOTES	
{	·
19. KEY WORDS (Continue on reverse side if necessary and identify by block number)	
Shale Oil Refining Basic Nitrogen	Extraction
JP-4 Jet Fuel Total Nitrogen	Reforming
JP-8 Jet Fuel Fluid Catalytic Cracking	
The second secon	
Hydrotreating Guardcase Hydrotreater	
Denitrogenation Automatic Saturation	
Denttrogenation Automatic Saturation  20. ABSTRACT (Continue on reverse side if necessary and identify by block number)	plation of two primary tasks
Denttrogenation  20. Asstract (Continue on reverse side it necessary and identify by block number)  Phase II work performed was directed toward the com	
Denttrogenation  Automatic Saturation  20. ABSTRACT (Continue on reverse side it necessary and identity by block number)  Phase II work performed was directed toward the comparable I under this phase was directed at evaluating	the effect of operating
Denitrogenation  20. ABSTRACT (Continue on reverse side it necessary and identify by block number) Phase II work performed was directed toward the comparable I under this phase was directed at evaluating conditions on material quality, energy balances, pro	the effect of operating oduct composition and
Denitrogenation  20. ABSTRACT (Continue on reverse side it necessary and identify by block number) Phase II work performed was directed toward the compart of the conditions on material quality, energy balances, presconomics. Task II was directed at obtaining scale-	the effect of operating oduct composition and -up data for those unit
Denitrogenation  20. ABSTRACT (Continue on reverse side it necessary and identify by block number) Phase II work performed was directed toward the comparable I under this phase was directed at evaluating conditions on material quality, energy balances, pro	the effect of operating oduct composition and -up data for those unit ory data to confirm prelim-

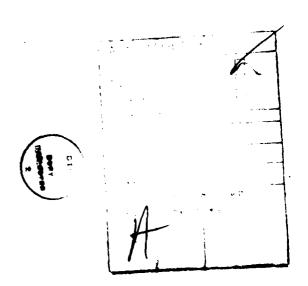
DD 1 JAN 73 1473 EDITION OF 1 NOV 65 IS OBSOLETE

Unclassified

#### **FOREWORD**

This project was sponsored by the U. S. Air Force
Wright Aeronautical Laboratories (AFWAL), Air Force Systems
Command, under Contract No. F33615-78-C-2080. The work
herein was performed during the period June 15, 1979 to
October 30, 1980 under Program Elements 62203F and 63215F,
Work Units 30480504 and 24800004, respectively. A discussion
of the entire Ashland effort under this contract can be found
in a series of six reports (AFWAL-TR-81-2056) entitled "Refining
of Military Jet Fuels from Shale Oil", Parts I, II, III, and IV
(Part II consists of three volumes). This report is Volume II
of Part II which describes the evaluation via the EXTRACTACRACKING
Process of the in situ Occidental shale oil. The cognizant Air
Force Project Scientist was Dr. Ronald D. Butler. We must give
recognition to Carolyn Honaker for her assistance in preparing
this report.

The second second second



## TABLE OF CONTENTS

## VOLUME II

		PAGE
SECTION		
I.	INTRODUCTION	1
II	CRUDE SHALE ANALYSIS	4
	SUMMARY	. 4
	OBJECTIVE	4
	MECHANICAL DESCRIPTION	4
	EXPERIMENTAL	5
	EXPERIMENTAL RESULTS	6
	DISCUSSION	15
	CONCLUSIONS	15
III	CRUDE SHALE OIL HYDROTREATER	16
·	1. PARAMETER VARIATION STUDIES	17
	SUMMARY	17
	OBJECTIVE	19
	MECHANICAL DESCRIPTION	19
	CHEMICAL/CATALYST DESCRIPTION	21
	FEEDSTOCK	21
	EXPERIMENTAL	22
	EXPERIMENTAL RUN PREFIX 111	23
	EXPERIMENTAL RUN PREFIX 112	26

CTION	PAGE
EXPERIMENTAL RUN PREFIX 113	26
EXPERIMENTAL RUN PREFIX 121	29
EXPERIMENTAL RUN PREFIX 122	32
EXPERIMENTAL RUN PREFIX 123	37
EXPERIMENTAL RESULTS	37
DISCUSSION	58
2. M-SERIES STUDIES	59
SUMMARY	59
OBJECTIVES	68
UNIT DESCRIPTION	68
EXPERIMENTAL	68
FEEDSTOCK	70
RESULTS	70
DISCUSSION	70
3. CRUDE SHALE OIL HYDROTREATER AGING	80
SUMMARY	80
OBJECTIVES	80
UNIT DESCRIPTION	80
FEEDSTOCK	82
EXPERIMENTAL	82
RESULTS	85
DISCUSSION	99

SECTION	,	PAGE
IV	FLUID CATALYTIC CRACKING OF IN SITU SHALE OIL	. 100
	1. MAT TESTS - FRESH FEED	. 101
	SUMMARY	. 101
	OBJECTIVES	. 102
	UNIT DESCRIPTION	. 102
	EXPERIMENTAL	. 103
	FEEDSTOCK	. 104
	RESULTS	. 105
	DISCUSSION	. 105
	RECOMMENDATIONS AND CONCLUSIONS.	. 125
	2. FCR PROCESSING - M-SERIES (FRESH FEED)	. 126
	SUMMARY	
	OBJECTIVES	
	UNIT DESCRIPTION	. 126
	FEEDSTOCK	. 127
	EXPERIMENTAL	. 128
	RESULTS	. 128
	DISCUSSION	. 131
	RECOMMENDATIONS AND CONCLUSIONS.	. 131
	3. FCR PROCESSING - OXY SAMPLE (FRESH FEED)	. 132
	SUMMARY	. 132
	OBJECTIVES	. 132

SECTION		•	PAGE
		EXPERIMENTAL	133
		UNIT DESCRIPTION	133
		FEEDSTOCK	133
		DISCUSSION	138
		RECOMMENDATIONS AND CONCLUSIONS	139
	3.	MAT TESTS - RECYCLE	139
		SUMMARY	139
	•	OBJECTIVES	140
		UNIT DESCRIPTION	140
	•	EXPERIMENTAL	140
		FEEDSTOCK	140
		RESULTS	140
		DISCUSSION	140
		RECOMMENDATIONS AND CONCLUSIONS	144
v	EXTR	ACTION	145
		SUMMARY	145
		OBJECTIVE	146
		MECHANICAL DESCRIPTION	146
		CHEMICAL DESCRIPTION	147
	EXPE	RIMENTAL PROCEDURE	147
	1.	SCREENING STUDY	147
	2.	CONTINUOUS EXTRACTION OF M-SERIES SAMPLES	148

SECTION		PAGE
	3. CONTINUOUS EXTRACTION OF IN SITU RETORTED SHALE OIL	149
	EXPERIMENTAL RESULTS	150
	1. SCREENING STUDY	150
	2. M-SERIES	153
	3. IN SITU RETORTED SHALE OIL	156
	DISCUSSION OF RESULTS	156
	1. PRELIMINARY SCREENING STUDY	156
	2. M-SERIES	164
•	3. IN SITU RETORTED SHALE OIL	164
	CONCLUSIONS	166
vı	RECYCLE OIL HYDROTREATING	169
	SUMMARY	169
	OBJECTIVES	169
	FEEDSTOCKS	169
	EQUIPMENT	170
	EXPERIMENTAL PROCEDURE	174
	RESULTS	176
	DISCUSSION OF RESULTS	178
	CONCLUSIONS/RECOMMENDATIONS	182
VII	GUARDCASE HYDROTREATING	183
	1. M-SERIES GUARDCASE HYDROTREATING	183
	SUMMARY	183

SECTION		PAGE
	OBJECTIVES	184
	MECHANICAL DESCRIPTION	184
	CHEMICAL AND CATALYST DESCRIPTIONS	186
	EXPERIMENTAL PROCEDURE	187
	EXPERIMENTAL RESULTS	189
	INITIAL HYDROTREATING	189
	REHYDROTREATING	193
	DISCUSSION OF RESULTS	193
	INITIAL HYDROTREATING	193
	REHYDROTREATING	195
	CONCLUSIONS	195
•	2. GUARDCASE HYDROTREATING	196
	SUMMARY	196
	OBJECTIVES	197
	MECHANICAL DESCRIPTION	197
	CHEMICAL AND CATALYST DESCRIPTION	197
	DESCRIPTION OF EXPERIMENTAL PROCEDURE	198
	EXPERIMENTAL RESULTS	200
	DISCUSSION OF RESULTS	206
	CONCLUSIONS	206
VIII	FREEZE POINT MODIFICATION	207

SECTION			PAGE
	1. '	SIMULATION OF FREEZE POINT MODIFICATION BY BLENDS OF JET FUEL AND NORMAL PARAFFINS	207
		SUMMARY	207
		OBJECTIVES	208
		EQUIPMENT AND EXPERIMENTAL	208
		FEEDSTOCKS	210
		RESULTS	210
		DISCUSSION	233
	2.	M-S2RIES	236
		SUMMARY	236
		OBJECTIVES	236
		EQUIPMENT AND EXPERIMENTAL	237
		FEEDSTOCK	237
		RESULTS	237
		DISCUSSION	237
		RECOMMENDATIONS AND CONCLUSIONS	240
	3.	ACCELERATED AGING	241
		SUMMARY	241
		OBJECTIVES	241
		EQUIPMENT AND PROCEDURE	241
		FEEDSTOCK	242
		RESULTS	242

SECTION		PAGE
	DISCUSSION	244
	RECOMMENDATIONS AND CONCLUSIONS	244
ıx	AROMATIC SATURATION	246
	SUMMARY	246
	OBJECTIVES	246
	EQUIPMENT	246
	FEEDSTOCKS	247
	PROCEDURE	247
	RESULTS	248
	DISCUSSION	250
	CONCLUSIONS	255

#### LIST OF FIGURES

FIGURES		PAGE
II-1	PODBIELNIAK (I) AND HEMPEL (II) FRACTIONATION OF OCCIDENTAL CRUDE SHALE OIL	9
II-2	API GRAVITY AS A FUNCTION OF BOILING RANGE FOR CRUDE OCCIDENTAL SHALE OIL	10
II-3	SULFUR CONTENT AS A FUNCTION OF BOILING RANGE FOR CRUDE OCCIDENTAL SHALE OIL	11
II <b>-4</b>	NITROGEN CONTENT AS A FUNCTION OF BOILING RANGE FOR CRUDE OCCIDENTAL SHALE OIL	12
II-5	BASIC NITROGEN CONTENT AS A FUNCTION OF BOILING RANGE FOR CRUDE OCCIDENTAL SHALE OIL	13
II-6	ARSENIC CONTENT AS A FUNCTION OF BOILING RANGE FOR CRUDE OCCIDENTAL SHALE OIL	14
III-1	CRUDE SHALE HYDROTREATER CONFIGURATION	20
III-2	TEMPERATURE DEPENDENCE OF DESULFURIZATION AT 2 LHSV COBALT MOLYBDATE CATALYST	60
III-3	IN SITU SHALE OIL HYDROTREATING OVER NICKEL MOLYBDATE CATALYST: TEMPERATURE DEPENDENCE OF DESULFURIZATION AT 2 LHSV	61
III-4	TEMPERATURE DEPENDENCE OF DENITROGENATION AT 2 LHSV COBALT MOLYBDATE CATALYST	62
III-5	IN SITU SHALE OIL HYDROTREATING OVER NICKEL MOLYBDATE CATALYST: TEMPERATURE DEPENDENCE OF DENITROGENATION AT 2 LHSV	63
III-6	IN SITU SHALE OIL HYDROTREATING OVER COBALT MOLYBDATE CATALYST	64
III-7	IN SITU SHALE OIL HYDROTREATING OVER NICKEL MOLYBDATE CATALYST	65
III~8	SECOND ORDER SPACE VELOCITY RESPONSE OF DENITROGENATION AT 1000 PSIG PRESSURE OVER COBALT MOLYBDATE CATALYST	66
III-9	IN SITU SHALE OIL HYDROTREATING OVER NICKEL MOLYBDATE CATALYST: HYDROGEN CONSUMPTION TRENDS.	67

FIGURES		PAGE
111-10	HETEROATOM REMOVAL STABILITY: CRUDE SHALE OIL HYDROTREATING - SERIES M-11	77
111-11	HETEROATOM REMOVAL STABILITY: CRUDE SHALE OIL HYDROTREATING - SERIES M-12	78
III-12	MODIFIED CRUDE SHALE HYDROTREATER FOR 30-DAY RUNS	81
111-13	CSHT AGING RESULTS FOR IN SITU SHALE OIL - TEMPERATURE, PRESSURE, LHSV, HYDROGEN CONSUMPTION	93
III-14	CSHT AGING RESULTS FOR IN SITU SHALE OIL - *API, SULFUR, NITROGEN, BASIC NITROGEN	94
IV-1	MICROACTIVITY TEST CRACKING SAMPLE 11103 WT. % CONVERSION	112
IV-2	MICROACTIVITY TEST CRACKING SAMPLE 11103 VOL. % RECOVERY	113
IV-3	MICROACTIVITY TEST CRACKING SAMPLE 11205 WT. % CONVERSION	114
IV-4	MICROACTIVITY TEST CRACKING SAMPLE 11205 VOL. % RECOVERY	115
IV-5	MICROACTIVITY TEST CRACKING SAMPLE 11312 WT. % CONVERSION	116
IV-6	MICROACTIVITY TEST CRACKING SAMPLE 11312 VOL. % RECOVERY	117
IV-7	MICROACTIVITY TEST CRACKING SAMPLE M-112 WT. % CONVERSION	118
IV-8	MICROACTIVITY TEST CRACKING SAMPLE M-112 VOL. % RECOVERY	119
IV-9	MICROACTIVITY TEST CRACKING SAMPLE 12205 WT. & CONVERSION	120
IV-10	MICROACTIVITY TEST CRACKING SAMPLE 12205 VOL. % RECOVERY	121

FIGURES		PAGE
IV-11	FCC YIELD MAXIMA DEPENDENCE ON FEEDSTOCK BASIC NITROGEN CONTENT FOR IN SITU SHALE OIL	122
IV-12	COMPARISON OF NITROGEN DOPED AND THE ANALOGOUS FRESH FEED YIELDS	123
IV-13	MICROACTIVITY TEST CRACKING SAMPLE M-112 LCO - WT. % CONVERSION	142
IV-14	MICROACTIVITY TEST CRACKING SAMPLE M-112 LCO - VOL. % RECOVERY	143
V-1	BATCH EXTRACTION OF BASIC NITROGEN	160
V-2	RATIO OF HYDROCARBON TO NITROGEN REMOVAL IN BATCH EXTRACTION	161
V-3	CONTINUOUS COUNTERCURRENT ACID EXTRACTION OF BASIC NITROGEN	162
V-4	RATIO OF HYDROCARBON TO NITROGEN REMOVAL IN CONTINUOUS COUNTERCURRENT EXTRACTION	163
VI-1	CYCLE OIL HYDROTREATER CONFIGURATION	173
VI-2	CYCLE OIL HYDROTREATING - IN SITU SHALE OIL - TEMPERATURE DEPENDENCE OF DESULFURIZATION AT 1 LHSV	179
VI-3	CYCLE OIL HYDROTREATING - IN SITU SHALE OIL - TEMPERATURE DEPENDENCE OF DENITROGENATION AT 1 LHSV	180
VII-1	GUARDCASE HYDROTREATER CONFIGURATION	185

FIGURES		PAGE
VIII-1	EFFECT OF TEMPERATURE ON JP-4 YIELD DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	217
VIII-2	EFFECT OF PRESSURE ON JP-4 YIELD DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	218
VIII-3	DEPENDENCE OF JP-4 YIELD- AS BOTH WT. % OF LIQUID PRODUCT AND AS WT. % OF FEED - ON PRODUCT AROMATIC CONTENT DURING REFORMING OF PARAFFIN-DOPED JP-7 FUEL	219
VIII-4	VARIATION OF JP-8 YIELD WITH TEMPERATURE DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	220
VIII-5	VARIATION OF JP-8 YIELD WITH PRESSURE DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUBL	221
VIII-6	DEPENDENCE OF JP-8 YIELD - AS WT. % OF LIQUID PRODUCT - ON PRODUCT AROMATIC CONTENT DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	222
VIII-7	DEPENDENCE OF JP-8 YIELD AS WT. % OF FEED - ON PRODUCT AROMATIC CONTENT DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	223
VIII-8	EFFECT OF TEMPERATURE ON PRODUCT FREEZE POINT DURING REFORMING OF n-PARAFFIN- DOPED JP-7 JET FUEL	224
VIII-9	EFFECT OF PRESSURE ON PRODUCT FREEZE POINT DURING REFORMING OF n-PARAFFIN- DOPED JP-7 JET FUEL	225
VIII-10	DEPENDENCE OF PRODUCT FREEZE POINT ON PRODUCT AROMATIC CONTENT DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	226
VIII-11	EFFECT OF TEMPERATURE ON PRODUCT AROMATIC CONTENT - AS WT. & OF FEED - DURING REFORM-ING OF n-PARAFFIN-DOPED JP-7 JET FUEL	227

FIGURES		PAGE
VIII-12	EFFECT OF PRESSURE ON PRODUCT AROMATIC CONTENT ~ AS WT. % OF FEED - DURING REFORM-ING OF n-PARAFFIN-DOPED JP-7 JET FUEL	228
VIII-13	VARIATION OF JP-4 AND JP-8 YIELD - AS WT. % OF FEED - WITH PRODUCT FREEZE POINT DURING REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	229
VIII-14	ACCELERATED AGING RESPONSE OF SAMPLE GC-1 (NORMALIZED DATA)	245

## LIST OF TABLES

TABLE		PAGE
11-1	CRUDE SHALE OIL EVALUATION	7
II-2	CRUDE SHALE OIL EVALUATION	8
III-I	FEEDSTOCK PROPERTIES - PARAMETER VARIATION STUDIES - OCCIDENTAL IN SITU SHALE OIL	18
III-2	SERIES 111 RUN CHRONOLOGY	24
III-3	SERIES 112 RUN CHRONOLOGY	27
III-4	SERIES 113 RUN CHRONOLOGY	30
111-5	SERIES 121 RUN CHRONOLOGY	33
III-6	SERIES 122 RUN CHRONOLOGY	35
III-7	SERIES 123 RUN CHRONOLOGY	38
III-8	PARAMETER VARIATION HYDROTREATING SERIES 111 - OPERATING CONDITIONS/ MATERIAL BALANCE	40
111-9	PARAMETER VARIATION HYDROTREATING SERIES 111 - PRODUCT PROPERTIES/ HETEROATOM REMOVAL	41
III-10	PARAMETER VARIATION HYDROTREATING SERIES 112 - OPERATING CONDITIONS/ MATERIAL BALANCE	42
111-11	PARAMETER VARIATION HYDROTREATING SERIES 112 - PRODUCT PROPERTIES/ HETEROATOM REMOVAL	43
III-12	PARAMETER VARIATION HYDROTREATING SERIES 113 - OPERATING CONDITIONS/ MATERIAL BALANCE	44
III-13	PARAMETER VARIATION HYDROTREATING SERIES 113 - PRODUCT PROPERTIES/ HETEROATOM REMOVAL	45
III-14	PARAMETER VARIATION HYDROTREATING SERIES 121 - OPERATING CONDITIONS/ MATERIAL BALANCE	46

TABLE		PAGE
III <b>-</b> 15	PARAMETER VARIATION HYDROTREATING SERIES 121 - PRODUCT PROPERTIES/ HETEROATOM REMOVAL	47
III-16	PARAMETER VARIATION HYDROTREATING SERIES 122 - OPERATING CONDITIONS/ MATERIAL BALANCE	48
III-17	PARAMETER VARIATION HYDROTREATING SERIES 122 - PRODUCT PROPERTIES/ HETEROATOM REMOVAL	49
111-18	PARAMETER VARIATION HYDROTREATING SERIES 123 - OPERATING CONDITIONS/ MATERIAL BALANCE	50
111-19	PARAMETER VARIATION HYDROTREATING SERIES 123 - PRODUCT PROPERTIES/ HETEROATOM REMOVAL	51
III-20 ·	PARAMETER VARIATION HYDROTREATING SERIES 111 - HYDROGEN DISTRIBUTION	52
III-21	PARAMETER VARIATION HYDROTREATING SERIES 112 - HYDROGEN DISTRIBUTION	53
III-22	PARAMETER VARIATION HYDROTREATING SERIES 113 - HYDROGEN DISTRIBUTION	54
III-23	PARAMETER VARIATION HYDROTREATING SERIES 121 - HYDROGEN DISTRIBUTION	55
111-24	PARAMETER VARIATION HYDROTREATING SERIES 122 - HYDROGEN DISTRIBUTION	56
111-25	PARAMETER VARIATION HYDROTREATING SERIES 123 - HYDROGEN DISTRIBUTION	57
111-26	M-SERIES HYDROTREATING FEED PROPERTIES OF OCCIDENTAL IN SITU SHALE OIL	71
III-27	CRUDE SHALE OIL HYDROTREATING SERIES M-11 OPERATING CONDITIONS, MATERIAL BALANCE, PRODUCT PROPERTIES	72
111-28	CRUDE SHALE OIL HYDROTREATING SERIES M-11 PRODUCT FRACTIONATION - M-111	73

TABLE		PAGE
III-29	CRUDE SHALE OIL HYDROTREATING SERIES M-11 PRODUCT FRACTIONATION - M-112	74
III-30	CRUDE SHALE OIL HYDROTREATING SERIES M-12 OPERATING CONDITIONS, MATERIAL BALANCE, PRODUCT PROPERTIES	75
111-31	CRUDE SHALE OIL HYDROTREATING SERIES M-12 PRODUCT FRACTIONATION - M-121	76
111-32	CRUDE SHALE HYDROTREATER AGING OCCIDENTAL CRUDE OIL (06-SH-94-01) FEED PROPERTIES	83
111-33	CHRONOLOGICAL RUN DESCRIPTION	86
III-34	OPERATING CONDITIONS, MATERIAL BALANCE	90
111-35	COMPOSITE PRODUCT INSPECTION, COMPOSITE 3, PERIOPS 9-11	95
III <b>-</b> 36	COMPOSITE PRODUCT INSPECTION, COMPOSITE 5, PERIODS 15-20	96
III-37	COMPOSITE PRODUCT INSPECTION, COMPOSITE 7, PERIODS 25-28	97
111-38	CATALYST ANALYSIS-IN SITU SHALE OIL HYDROTREATER CATALYST AGING 10/5/79 - 11/2/79	98
IV-1	MICROACTIVITY TESTS - FRESH FEEDSTOCK	106
IV-2	MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY - FEEDSTOCK 11103	107
IV-3	MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY - FEEDSTOCK 11205	108
IV-4	MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY - FEEDSTOCK 11312	109
IV-5	MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY - FEEDSTOCK M-112	110
IV-6	MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY - FEEDSTOCK 12205	111

TABLE		PAGE
IV-7	FLUID CATALYTIC CRACKING FCR PROCESSING - SAMPLE M-112	129
IV-8	FLUID CATALYTIC CRACKING FCR PROCESSING - SAMPLE M-121	130
IV-9	FLUID CATALYTIC CRACKING FCR PROCESSING - SAMPLE OXY-1	134
IV-10	FLUID CATALYTIC CRACKING FCR PROCESSING - SAMPLE OXY-2	135
IV-11	FLUID CATALYTIC CRACKING FCR PROCESSING - SAMPLE OXY-3	136
IV-12	FLUID CATALYTIC CRACKING FCR PROCESSING - SAMPLE OXY-4	137
IV-13	MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY - M-112-LCO	141
V-1	EXPERIMENTAL DESIGN FOR THE HYDROTREATED IN SITU SHALE OIL DISTILLATE EXTRACTION PARAMETER STUDY	151
V-2	DATA SUMMARY ~ PARAMETER STUDY FOR BATCH EXTRACTION OF IN SITU HYDROTREATED SHALE OIL DISTILLATES	152
V-3	EXTRACTION OF BLENDED PRODUCT - DATA SUMMARY	154
V-4	CONTINUOUS COUNTERCURRENT EXTRACTION OF SHALE OIL	155
v-5	M-SERIES CONTINUOUS EXTRACTION RESULTS FOR IN SITU SHALE DISTILLATES	157
V-6	CONTINUOUS EXTRACTION RESULTS FOR GC-1 STUDY - IN SITU SHALE DISTILLATES	158
V-7	OPERATING DATA TABLE FOR CONTINUOUS EXTRACTION OF IN SITU SHALE OIL DISTILLATES	165
V-8	OPERATING PARAMETERS FOR EXTRACTION	

TABLE		PAGE
VI-1	FEED CHARACTERIZATION: CYCLE OIL HYDROTREATING - SAMPLE M-112	171
VI-2	FEED CHARACTERIZATION: CYCLE OIL HYDROTREATING - SAMPLE M-121	172
VI-3	M-SERIES OPERATING CONDITIONS FOR RECYCLE HYDROTREATING OF IN SITU OIL	175
VI-4	CYCLE OIL HYDROTREATING: IN SITU SHALE OIL	177
VI-5	DISTILLATION OF FULL RANGE CYCLE OIL HYDROTREATED IN SITU SHALE OIL	181
VII-1	INITIAL HYDROTREATING	186
VII-2	REHYDROTREATING	187
VII-3	INITIAL GUARDCASE HYDROTREATING - SAMPLE M-112	190
VII-4	REHYDROTREATING OF SAMPLE M-112 THROUGH GUARDCASE HYDROTREATER	191
VII-5	INITIAL GUARDCASE HYDROTREATING - SAMPLE M-121	192
VII-6	REHYDROTREATING OF SAMPLE M-121 THROUGH GUARDCASE HYDROTREATER	194
VII-7	GUARDCASE HYDROTREATING OF SAMPLE GC-1	201
VII-8	TODD PSEUDOCOMPONENTS - GC-1 FEED	202
VII-9	TODD PSEUDOCOMPONENTS - GC-1 LINEOUT 1	203
VII-10	TODD PSEUDOCOMPONENTS - GC-1 LINEOUT 2, 3, 4, 5 AND TEST 1	204
VII-11	TODD PSEUDOCOMPONENTS - GC-1 ON SPEC	205

TABLE	•	PAGE
VIII-1	FREEZE POINT MODIFICATION SIMULATION - FEEDSTOCK PROPERTIES	211
VIII-2	REFORMING OF CETANE	213
VIII-3	REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	214
VIII-4	REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	215
VIII-5	REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL	216
VIII-6	REFORMING OF JET FUEL CONTAINING n-PARAFFIN	231
VIII-7	REFORMING OF JET FUEL CONTAINING n-PARAFFIN	232
VIII-8	MILITARY JET FUEL FROM SHALE OIL SAMPLE PREPARATION SEQUENCE	238
VIII-9	M-SERIES FREZE POINT MODIFICATION	239
VIII-10	GC-1 ACCELERATED AGING	· 243
IX-1	ANALYTICAL RESULTS FOR AROMATIC SATURATED PRODUCTS	249
IX-2	MATERIAL BALANCE RESULTS FOR AROMATIC SATURATION OF SAMPLE M-112	251
IX-3	PERTINENT JET FUEL SPECIFICATIONS FOR AROMATIC SATURATE SAMPLES	252
IX-4	MATERIAL BALANCE RESULTS FOR AROMATIC SATURATION OF SAMPLE M-121	253
IX-5	MATERIAL BALANCE RESULTS FOR AROMATIC SATURATION OF SAMPLE GC-1	254

#### **ABBREVIATIONS**

Aromatics A

Acid Str Acid Strength

APL Aero Propulsion Laboratory

Air Force Wright Aeronautical Laboratories **AFWAL** 

American Petroleum Institute API

AO Aqueous

Aromatic Saturation Unit AR SAT

Arsenic As

American Society for Testing Materials ASTM

Barrel bbl

Basic Nitrogen BN

Ratio of Basic Nitrogen over Total Nitrogen (BN)

(TN)p (TN)f in Product to Basic Nitrogen over Total

Nitrogen in Feed

**BPD** Barrels per Day

Bromine Number BR NO.

Bottoms, Sediment and Water BS&W

British Thermal Unit BTU

°C Degrees Celsius

C Carbon or Heteroatom Concentration

Methane  $c_1$ 

 $C_2$ Ethane

C<sub>3</sub> Propane

Propylene C3=

n-C<sub>4</sub> Normal Butane

i-C<sub>4</sub> Isobutane

C4 Butylene

C<sub>5</sub> Pentane

C5's Pentane and Pentenes

C5<sup>+</sup> Compounds Heavier than Pentane in Gas

Streams

C<sub>6</sub> Hexane

C/H Carbon to Hydrogen Ratio

cc Cubic Centimeters

cm Centimeter

CO Carbon Monoxide or Cycle Oil

C/O Catalyst-to-Oil Ratio

COHT Cycle Oil Hydrotreater

Co/Mo or Cobalt Molybdate Catalyst

CoMo

Cont. Time Contact Time

Conv. Conversion

CR Cracked

CSHT Crude Shale Hydrotreater

cSt Centistoke

CW Cooling Water

E Activation Energy

EP End Point in a Distillation

EQ. SUPER DX Equilibrium Super DX - a cracking catalyst

EXTD Extracted

°F Degrees Fahrenheit

FCC Fluid Catalytic Cracker

FCR Fluid Catalytic Reactor

Fe Iron

FI Flow Indicator

FIA Fluorescent Indicator Adsorption - Test

Method to Determine Hydrocarbon Types

g Gram

GC Guardcase Hydrotreater

GC-1 Guardcase Study on In Situ Shale Oil

Gms Grams

H<sub>2</sub> Hydrogen

HC Hydrocarbon

HC/N Ratio of Hydrocarbons to Nitrogen

Hg Mercury

Hr or hr Hours

Hr<sup>-1</sup> or Inverse Hours

hr-1

H<sub>2</sub>S Hydrogen Sulfide

Hydrog. Hydrogenated

IBP Initial Boiling Point

I.D. Inside Diameter

in Inches

THE RESERVE THE PROPERTY OF TH

JP-4 Military Specification MIL-T-5624K Turbine

Fuel

JP-5 Military Specification MIL-T-5624H Turbine

**Fuel** 

JP-7 Military Specification MIL-T-38219 Turbine

Fuel

JP-8 Military Specification MIL-T-83133 Turbine

Fuel

K<sub>O</sub> Rate Constant

lb Pound

LCO Light Cycle Oil

LHSV Liquid Hourly Space Velocity

LPG Liquefied Petroleum Gas

LV% Liquid Volume Percent

MAT Microactivity Test

MAV Maleic Anhydride Value

MAX Maximum

mg Milligram

MIN or min Minimum

min Minute

ml Milliliter

mm Millimeter

MT'L Material

N Nitrogen

No or Nf Nitrogen in Feed

Np Nitrogen in Product

Na<sup>+</sup> Sodium Ion

Na<sub>2</sub>CO<sub>3</sub> Sodium Carbonate

NaOH Sodium Hydroxide

NH<sub>3</sub> Ammonia

Ni Nickel

Ni/Mo or Nickel Molybdate Catalyst

NiMo

NO. Number

NO<sub>x</sub> Oxides of Nitrogen

n-PARAFFIN Normal Paraffin

O Olefins

ON SPEC Meeting Specifications

OXY Test Series on In Situ Shale Oil

P Pressure

(P + N) Paraffins and Naphthenes

PRO Test Series on Above Ground Shale Oil

PI Pressure Indicator

PID Proportional-Integral-Derivative Controller

Controller

ppb Parts per Billion

ppm Parts per Million

psia Pounds of Force per Square Inch Absolute

PSIG or psig Pounds of Force per Square Inch Gauge

Pt/Al Platinum/Aluminum Catalyst

Pt/Re Platinum/Rhenium Catalyst

PV Parameter Variation

R Gas Constant

°R Degrees Rankine

RAFF Raffinate

RAMS C Ramsbottom Carbon

RCVY Recovery

R&D Research & Development

REF Reformer

REGEN Regenerator Section of FCC

RVP Reid Vapor Pressure

S Sulfur

SAT Saturates

SCFB Standard Cubic Feet per Barrel

SCFH Standard Cubic Feet per Hour

SIM-D Simulated Distillation

SR Straight Run

SS Stainless Steel

SU or SUS Saybolt Universal Seconds

TBP True Boiling Point

TEMP or t Temperature

TIC Temperature Indicator and Controller

V Vanadium

VIS Viscosity

VOL Volume

VO1 % Volume Percent

WBS Work Breakdown Structure

WHSV Weight Hourly Space Velocity

WT Weight

WT %, % W, Weight Percent

or % WT

# SYMBOLS

=	Equals
+	Positive, plus, or greater than as in 600°F+
-	Negative, minus, or less than as in -600°F
8	Percent
>	Greater Than
<	Less Than
•	Inches
*	Number or Pound
Δ	Difference

#### SUMMARY

A complete EXTRACTACRACKING evaluation sequence has been performed for Occidental in situ shale oil. Extensive tests for hydrotreating, catalytic cracking, extraction and product upgrading were successfully performed. Data from these tests are contained in this volume.

These tests demonstrate that this oil is an acceptable feedstock for EXTRACTACRACKING. Further, specification JP-4 and JP-8 turbine fuels were produced from these materials and provided to the Air Force.

#### SECTION I

#### INTRODUCTION

On February 15, 1979, Ashland Petroleum Company Research and Development Department began work under contract F33615-78-C-2080, An Exploratory Research and Development Program Leading to Specifications for Aviation Turbine Fuel from Whole Crude Shale Oil. This program, funded by the Department of Defense, is designed to evaluate the EXTRACTACRACKING process as a potential improvement in shale oil refining technology.

ed to reduce costs by minimizing hydrogen consumption and operating severity requirements. A more detailed discussion of the process is contained in Volume I of this report.

Phase I of this study was performed as a preliminary economic analysis of the process and was completed on June 15, 1979. Phase II was performed to evaluate process parameters in order to define optimum operating regions based on cost and to prepare five 500-ml samples of aviation turbine fuel from the complete process. Data obtained during Phase II is reported in three volumes:

Volume I: Preparation of Laboratory-Scale Fuel Samples

Volume II: In Situ Shale Oil Process Data

Volume III: Above-Ground Shale Oil Process Data

This volume collects all data obtained during processing of in situ shale oil during Phase II, for the period June 15, 1979 through October 1, 1980.

Work in Phase II was broken into major functional areas for completion. Since EXTRACTACRACKING is an integrated process, both screening and sample production studies for further processing were required. Where possible, the program was defined such that small sample quantities could be utilized.

Initial rapid parameter screening tests were performed for the crude shale hydrotreater module. These tests comprised a wide range of operating severities over two catalyst types. Products from these screening tests were fractionated and selected bottoms samples evaluated for FCC performance on the microactivity unit (MAT). Selected distillate samples were processed by batch extraction.

Results from this preliminary screening were utilized to define anticipated required operating severities for these units. Those conditions were then utilized for production of the M-Series samples which were processed through the entire flow sheet.

The M-Series data, plus the earlier screening data, were utilized to define preliminary optimum operating regions by the process computer model developed during

Phase I. Conditions indicated by that preliminary optimization were then utilized for preparation of the listed Oxy sample, which also was processed through the entire processing sequence.

All data obtained for in situ shale oil is contained in this volume. Very little effort has been made herein to define module interactive or other effects; these will be discussed fully in Volume I. The purpose of this volume is to delineate data bases and procedures for further evaluation.

#### SECTION II

#### CRUDE SHALE ANALYSIS

#### SUMMARY

Two types of shale oils, in situ and above ground retorted, have been characterized by fractionation and subsequent analysis of the fractional products. These efforts were required as outlined for fulfillment of Phase II objectives in the Military Jet Fuels from Shale Oil Program.

The shale oils, Occidental Petroleum's in situ oil produced during Retort \$6 and Paraho above-ground retorted oil obtained from the government's Anvil Points facility, were handled identically and simultaneously.

#### **OBJECTIVE**

The raw shale oils under investigation were fractionated as described in order to provide a detailed definition of physicochemical properties. This data set becomes the basis for all subsequent observations regarding the effectiveness of EXTRACTACRACKING process modules.

#### MECHANICAL DESCRIPTION

The fractionation of each oil sample was accomplished in two increments of work. Each sample was first distilled in increments to an endpoint of approximately 600°F, after which the >600°F boiling range liquids were transferred to an alternate apparatus and the distillation was continued.

A Podbielniak Semi-Cal model 3650 fractional distillation apparatus was utilized for the low temperature portion of the fractionation. This column is a completely vacuum - jacketed model packed with .09" x .175" x .175" stain-less steel Heli-Pak column internals. The column was operated at both atmospheric and subatmospheric pressures at a 5:1 reflux ratio.

In order to perform the high temperature segment of the characterization, the >600°F liquid was transferred to a modified Hempel apparatus. This consisted of a round bottom distillation flask with thermowell, a liquid dividing still head with warm and cold condensing water capability, product receiver, tilting McLeod manometer and vapor trap. The system was evacuated with a Welch brand mechanically backed diffusion pump.

#### EXPERIMENTAL

A five gallon sample was removed from Occidental in situ oil drum #06-SH-96-01. The 55 gallon drum was heated for two hours on the high setting of a band-type drum heater wrapped around the lower third of the drum. The drum was then agitated vigorously for approximately one minute and a sample withdrawn through a spigot on the drum cap. No filtration was performed.

About one gallon of the oil sample was then dewatered by centrifugation at 140°F. Samples of the raw dewatered

crude were submitted for analytical evaluation and the remainder of the oil was charged to a  $1.18^{4} \times 48^{8}$  Podbielniak fractionating column.

The Podbielniak still was operated at a 5:1 reflux ratio to collect fractions in 100°F increments from the IBP up to 600°F. Both volume and weight measurements of the charge and product were made.

The >600°F residue was transmitted to the Synthetic Oils lab where it was subjected to further fractionation. Equipment limitations precluded fractionation beyond the 600°F limit on the Podbielniak unit without fear of thermal decomposition. An alternate method was devised for this high temperature fractionation by distilling at low pressures (<lmm Hg) and total product take-off on the modified Hempel apparatus described previously. The material was fractionated again in 100°F increments above the 600°F initial point to the temperature at which thermal decomposition was observed. Once again, both weight and volume measurements of the charge and product fractions were made.

Samples of each discrete fraction were subjected to analysis as shown in Table II-2.

# EXPERIMENTAL RESULTS

Raw analytical results are shown in Tables II-1 and II-2. Fractionation documentation can be found in Figure II-1. Properties of each discrete fraction are displayed graphically in Figures II-2 through II-6.

TABLE II-1

# CRUDE SHALE OIL EVALUATION

SAMPLE TYPE:

OCCIDENTAL IN SITU CRUDE SHALE OIL

SAMPLE NUMBER: 06-SH-96-01

ITEM		ITEM	
CABBON WT %	85.9	GRAVITY. OAP!	23.6
% IM Naccedan	11.0	SPECIFIC GRAVITY	0.9123
NITROGEN WT %	1.42	VIS at 100°F. SU	161.8
SIII 5110 WT %	0.53	VIS at 140°F.SU	75.1
OXYGEN WT %	1.21	K FACTOR	11.6
BASIC NITROGEN WT %	0.859	REID WAPOR PRESSURE	0.6
PHENOIS WT %	0.578	DISSOLVED HAS, DOM	21
NO NO O	32	RAMSBOTTOM CARBON	1.22
NICKEL DOM	11	SALT. #/1000 bbl	2.47
MANANIM SOM	1>1	% lov W & 88	0.1
ABSENIC DOM	24	ACID NUMBER	1.36
ORGANIC CHIORIDE pom	4	POUR POINT OF	65
COLOR	8+	MALEIC ANHYDRIDE VALUE	67.4
RECOVERY ON DEWATERING, Vol %	7.66		

TABLE IF2

# CRUDE SHALE OIL EVALUATION

OCCIDENTAL IN SITU

SAMPLE NUMBER: 06-SH-96-01

ITEM .	(278)-	300-	400-	-909	1007	-009	700-	-008	1 000
FRACTION	300	400	200	009	900T	700	800	906	¥00+
Yield. LV %	0.5	3.2	12.5	17.0	66.2	5.8	9.1	23.1	28.4
Yield, Wt %	•	2.9	11 7	16.3	68.0	5.8	9.2	23.3	29.8
K Factor	•	11.6	11.7	11.6	11.6	11.4	11.42	11.7	-
API Gravity		40.7	34.0	29.9	19.7	24.6	23.2	22.9	17.1
Specific Gravity	1	.8217	.8550	.8767	. 9358	. 9065	.9147	. 9165	. 9522
Carbon, Wt %	1	-	•	_	85.0	86.1	85.3	84.6	85.03
Hydrogen. Wt %	•	•	12.06	11.1	11.5	12.2	11.9	11.6	11.09
Nitrogen. Wt %	0.87	1.04	1.01	1.29	1.49	1.58	1.76	1.58	1.54
Basic Nitrogen, Wt %	•	0.83	16.0	0.87	0.84	0.74	0.81	0.71	0,75
Sulfur, Wt %	1	08.0	0.57	09.0	0.55	0.47	0.43	0.55	0.49
Oxvgen. Wt %	1	•	1	-	1	-	•	•	1
Phenols, Wt %	,	3.12	,	•	•	•	•	ı	•
Iron, ppm		1	•	•	•	7	<1	<١	111
Nickel, ppm	•	•		•	•	<1	<1	<1	23
Vanadium, ppm	1	-	1	1	-	<1	<1	<1	2
Arsenic, ppm	•	1	4	٠5	•	16	19	25	40
Organic CI, ppm	•	1	,	1	•	•	•	ı	
Ramsbottom Carbon	ŧ	ŧ	-	1	1.94	0.31	0.35	ე. 40	•
Saturates, Vol %	•	-	31.8	•		*	•	•	_
Olefins, Vol %	-	-	32.9	•		•	1	•	8
Aromatics, Vol %	•	•	35.3	•	1	•	1	1	•
Pour Point, OF	1	•		•	•	35	55	95	•
Viscosity, 60°F, SUS	_	-		•	•	1	ı	1	
Viscosity, 100°F, SUS	-	1.24	4.95	4.9	80.1	10.52	21.02	72.20	8
Viscosity, 210°F, SUS	-	0.86	1.06	1.74	15.45	2.35	3.41	6.74	1

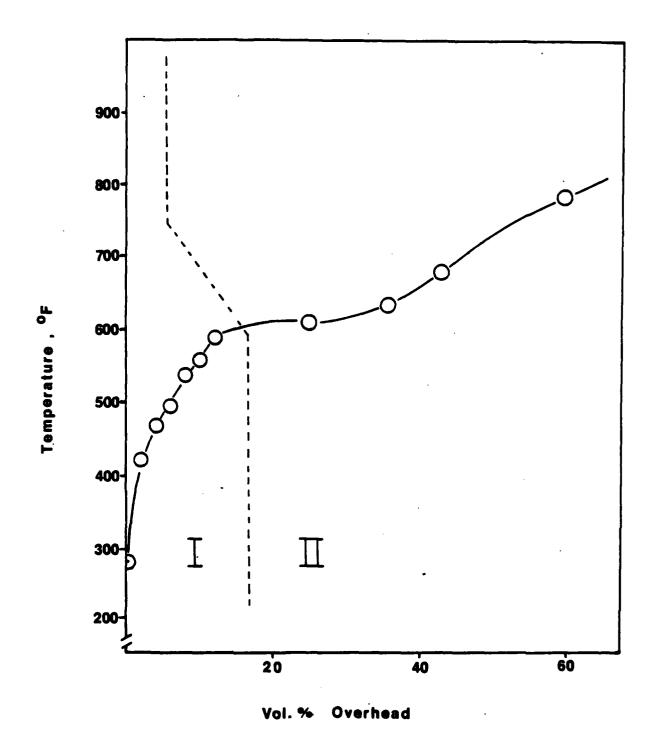


Figure II-1. Podbielniak (I) And Hempel (II) Fractionation Of Occidental Crude Shale Oil

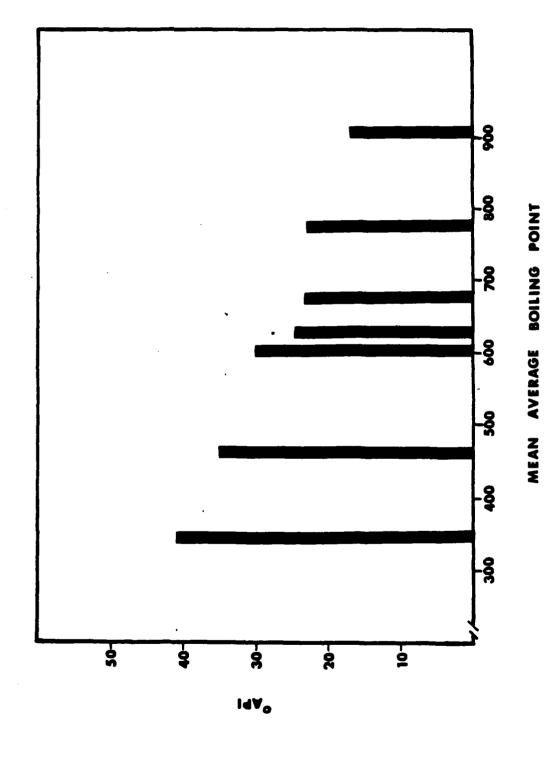


Figure 11-2. API Gravity As A Function Of Boiling Range For Crude Occidental Shale Oil

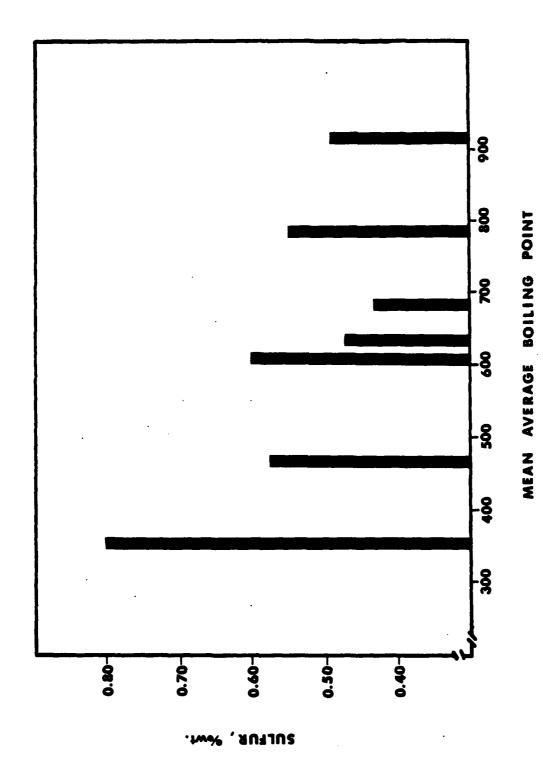


Figure II-3. Sulfur Content As A Function Of Boiling Range For Crude Occidental Shale Oil

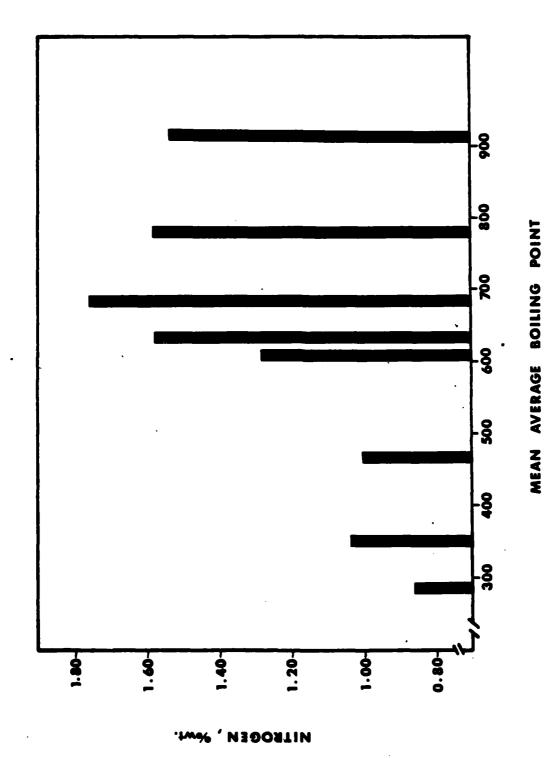


Figure II-4. Nitrogen Content As A Function Of Boiling Range For Crude Occidental Shale Oil

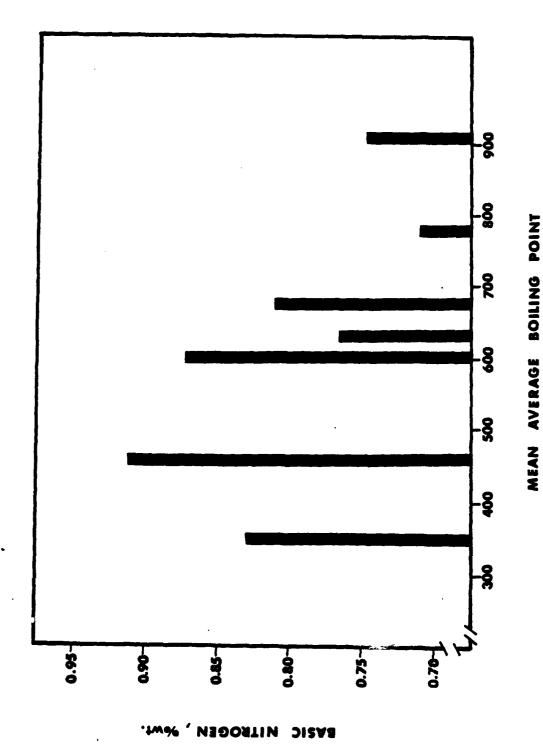


Figure II-5. Basic Nitrogen Content As A Function Of Boiling Range For Crude Occidental Shale Oil

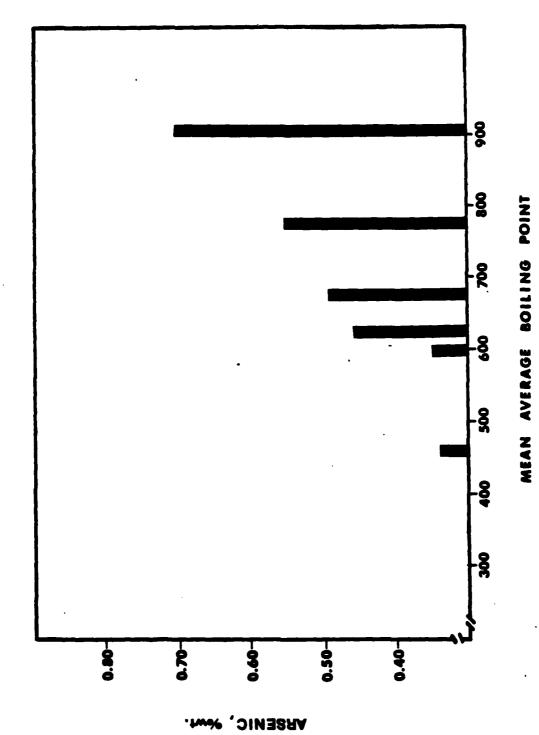


Figure II-6. Arsenic Content As A Function Of Boiling Range For Crude Occidental Shale Oil

#### **DISCUSSION**

Boiling ranges quoted for the fractions above 600°F are nominal ranges. This is due mainly to the fractionating efficiency sacrificed with the previously described apparatus in order to maintain low system pressures and temperatures, as well as short column residence time to avoid thermal cracking.

While the majority of the analytical functions were performed within the Research Department, some analyses were performed by independent analysts due to internal equipment limitations. Quantities of certain distillate fractions produced limited the number of analyses which could be performed on those fractions (e.g., IBP-300°F, 300-400°F).

The Watson characterization factor, K, for each of the fractions was derived from correlations using viscosity and API gravity.

#### CONCLUSIONS

The Occidental crude oil has been successfully separated into the nominal boiling ranges shown in Table II-2. Considerable effort was expended to avoid thermal cracking, a phenomenon more likely observed at lower temperatures for shale oils than for most petroleum crudes.

#### SECTION III

#### CRUDE SHALE OIL HYDROTREATER

The crude shale hydrotreater provided for the EXTRACT-ACRACKING process supplies the following functions:

- 1. Desulfurization
- 2. Demetallization, and in particular, dearsenation
- 3. Olefin saturation
- 4. Denitrogenation
- 5. Deoxygenation

In contrast to most work in this field, this crude shale hydrotreater is intended to be a low severity, low hydrogen consumption unit. It provides feedstock cleanup with respect to items 1-3, with as little denitrogenation as possible. The combination of low hydrogen addition, while removing metallic contaminants and olefins for handling purposes, is projected to provide an ideal combination of feedstock preparation for both extraction and fluid catalytic cracking.

Work in this area was of necessity extensive, particularly for preparation of samples for further processing.

The studies performed were:

- Parameter screening studies, including operating severity and catalyst evaluations,
- 2. Sample preparation and catalyst stability studies,
- Extended sample preparation and catalyst aging studies.

These studies ranged from 16 hour tests for the parameter screening studies to over 700 hours for the aging tests.

## 1. PARAMETER VARIATION STUDIES

# SUMMARY

The overall Military Jet Fuels from Shale Oil program provided a broad range of crude shale hydrotreater data with a minimum of experimental time. This was achieved through a series of parameter variation experiments which are detailed herein.

Each discrete test period from this series of experiments was assigned a three digit prefix followed by a two digit test number. This five digit code was used for documentation and sample retention.

Feedstock for experimental series 111, 112, 113, 121, 122 and 123 was Occidental oil produced in Retort #6 provided by the Air Force in a 96 drum shipment. The drum used for run prefix 111 and 112 experiments was assigned Synthetic Oils #06-SH-96-01, while the feed for runs 113, 121, 122 and 123 was taken from drum #06-SH-95-01. Feedstock samples were drawn directly from the drums and were fed to the reactors without filtration. Feed properties are shown in Table III-1.

TABLE III-1

# FEEDSTOCK PROPERTIES - PARAMETER VARIATION STUDIES - OCCIDENTAL IN SITU SHALE OIL

	Run	Ser	ies
	111	113	122
	112	121	123
Drum ID	06-SH-96-01	06-SH-	95-01
°API	23.7	23.	7
Sulfur, Wt. %	0.53	0.	53
Nitrogen, Wt. %	1.42	1.	40
Oxygen, Wt. %	1.21	1.	20

#### OBJECTIVE

This series of parameter variation studies was designed to reveal the short term hydrotreating response of full range in situ shale oil under multiple sets of conditions. A by-product of this arrangement was the production of sufficient refined shale oil products for further studies on subsequent process modules.

The primary objectives of the crude shale hydrotreater are:

- 1. Sulfur removal
- 2. Arsenic removal
- 3. Olefin saturation
- 4. Removal of iron and other catalyst poisoning metals
- 5. A small degree of nitrogen and oxygen removal
- 6. An increase in basic/total nitrogen ratio
- 7. Some aromatic saturation with minimal hydrogen consumption

#### MECHANICAL DESCRIPTION

Hydrotreatment was performed through a 1° I.D. universal trickle-flow reactor (see Figure III-1 for the specific reactor configuration). After mixing with hydrogen, the oil passes in a downflow manner across consecutive layers of tabular alumina, the hydrotreating catalyst bed and again over alumina before it is discharged into the low pressure liquid collection portion of the system. Product

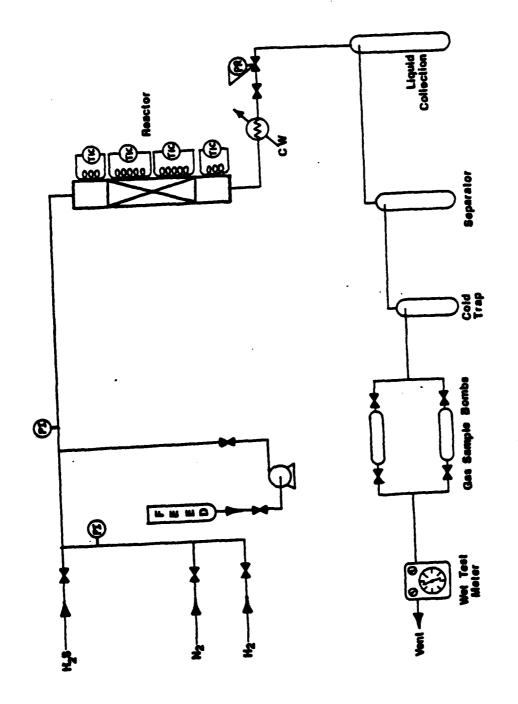


Figure III-1. Crude Shale Hydrotreater Configuration

liquids are collected and weighed while gases are chilled, sampled for analysis and measured through a wet test meter before being vented externally.

#### CHEMICAL/CATALYST DESCRIPTION

With the exception of the type of commercial hydrotreating catalyst used, reactors were assembled and packed identically for each of the six series of parameter variation studies. Specific descriptions of packing sequences will be given individually for the various experiments. In general, a 100-ml bed of extruded hydrotreating catalyst was mixed in alternating 10-ml increments with an equal volume of sized Ottawa sand and this bed was placed approximately in the center of the universal reactor. As stated previously, the catalyst/sand mixture was "sandwiched" between columns of tabular alumina chips.

Reactors used for treatment of 111, 112 and 113 series experiments were packed with 100-m1 of cobalt molybdate commercial hydrotreating catalyst. A nickel molybdate type catalyst was used in series 121, 122 and 123 hydrotreating.

Each fresh charge quantity of catalyst (i.e., each 100- $^{\rm ml}$ ) was presulfided in situ with H<sub>2</sub>S at maximum run temperature and bottle pressure for two hours.

#### FEEDSTOCK

Two feedstock drums were utilized for these studies.

Properties of these drums are shown in Table III-1, and, as

can readily be seen, these properties are essentially identical.

#### EXPERIMENTAL

Each run series consisted of 12 discrete tests (e.g., 11101 through 11112). Eight-hour material balance periods were used in order to gather as much data as possible in the shortest period of time. In each series the catalyst was dumped and the reactor recharged between the sixth and seventh tests. Specific operations for each series can be found under each run-prefixed experimental description.

Each discrete test period from this series of experiments was assigned a three-digit prefix followed by a two-digit test number. This five digit code was used for documentation and sample retention.

Prefixes were generated in the form:

#### XYZ

Where X = Shale Type	l = In Situ
	2 = Above Ground
Where Y = Catalyst Type	1 = Co/Mo
	2 = Ni/Mo
Where Z = Pressure	1 = 500 psig
	2 = 1000 psig
	3 = 2000 psia

Total sample number, for example, for a 500 psig run on in situ oil, over Co/Mo catalyst, at 600°F and 1 LHSV would be: 11101. The in situ oil feed was hydrotreated over the range of conditions shown below:

Temperature:

600-750°F

LHSV:

 $1-4 hr^{-1}$ 

Pressure:

500-2000 psig

Catalyst:

Ni/Mo, Co/Mo

## **EXPERIMENTAL RUN PREFIX 111**

Occidental oil from drum 06-SH-96-01 was hydrotreated over Ni/Mo catalyst at 500 psig total pressure and over the ranges 600-750°F and 1-4 LHSV.

The reactor was packed with 100-ml of 1/8" catalyst and 100-ml of Ottawa sand in alternating 10-ml increments, followed by tapping to pack the bed. The postheat alumina rose to a depth 34" from the reactor inlet, a 13" catalyst bed rested on top of the postheat zone, and the remainder of the reactor space was filled with tabular alumina to comprise the preheat zone.

After pressure testing, the catalyst was presulfided with excess  $H_2S$  at  $750\,^{\circ}F$  and bottle pressure for two hours. The unit was cooled with flowing hydrogen and feed was initiated. Temperatures were raised in  $50\,^{\circ}F$  increments every 1/2 hour until the desired operating temperature was achieved.

A chronological run description is shown in Table III-2.

Product from each material balance period was weighed and fractionated into IBP-400, 400-600 and 600°F+ fractions for analysis.

# TABLE III-2

# SERIES 111 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
6/22		Reactor packed, pressure tested
6/25		Catalyst presulfided
6/26	0350 1150 1950	Begin 11101 lineout Begin 11101 test End 11101 test, begin 11102 lineout
6/27	0350 1150 1950 2055 2155	Begin 11102 test End 11102 test, begin 11103 lineout Begin 11103 test H2 rate reduced to 10 CFH Feed cans changed, new feed gravity measured Oil in off-gas line; line removed and cleaned
6/28	0350 0615 0650 0920 1150 1950 2150 2335	End 11103 test, begin 11104 lineout Pump failure Replacement pump failed, temperatures lowered 50°F Pump rebuilt and working H2 rate corrected, begin 11104 test End 11104 test, begin 11105 lineout H2 rate adjusted, begin 11105 test H2 rate twice desired rate, adjusted
6/29	0550 0750 1550	End 11105 test, begin 11106 lineout End 11106 lineout, begin test 11106 End 11106 test, unit shutdown
7/3		Reactor repacked
7/5		Pressure check, presulfiding
7/9	1520 2320	Begin 11107 lineout Begin 11107 test

# TABLE III-2 (CONT'D)

# SERIES 111 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/10	0720 1520 1820	End 11107 test, begin 11108 lineout Begin 11108 test Collection vessel stopper off, wet test meter readings off
	2320	End 11108 test, begin 11109 lineout
7/11	0720 1520 1610 2320	Begin 11109 test End 11109 test, begin 11110 lineout H <sub>2</sub> rate adjusted Begin 11110 test
7/12	0320 0720 1520 1920 2320	Wet test meter replaced End 11110 test, begin 11111 lineout Begin 11111 test H <sub>2</sub> rate reduced End 11111 test, begin 11112 lineout
7/13	0620 0720 1050 . 1520	Wet test meter replaced Begin 11112 test Pump replaced End 11112 test; unit shutdown

#### EXPERIMENTAL RUN PREFIX 112

The Occidental oil feed from drum 06-SH-96-01 was hydrotreated over a 100-ml void-filled bed of 1/8" Co/Mo extrudate. System pressure was maintained at 1000 psig while temperatures and space rates were varied from 600-750°F and 1-4 hr<sup>-1</sup>, respectively.

A bed of tabular alumina rising to within 34.5" of the reactor inlet constituted the postheat zone; on top of this bed rested a 13" catalyst bed void-filled with Ottawa sand, and the remainder of the reactor was filled with tabular alumina as preheat.

After pressure testing, the catalyst was presulfided with  $H_2S$  at 775°F and bottle pressure for two hours. The unit was cooled with flowing hydrogen and feed initiated. Temperatures were increased in 50°F increments each 1/2 hour to the desired level.

A chronological run description is shown in Table III-3.

Each material balance product was weighed and fractionated into IBP-400, 400-600 and 600°F+ fractions for analysis.

# EXPERIMENTAL RUN PREFIX 113

The Occidental oil feed from drum \$06-SH-95-01 was hydrotreated over Co/Mo catalyst at a total system pressure of 2000 psig. Temperatures and space rates were varied over the ranges 600-750°F and 1-4 hr<sup>-1</sup>, respectively.

TABLE III-3

# SERIES 112 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
6/25	1230 1635	Reactor packed, pressure tested, pre- sulfided Feed initiated Problems controlling reactor pressure and feed rate encountered
6/26	0130 0720 0930 1730	Begin 11201 lineout Unit pressure high, adjusted Begin 11201 test End 11201 test, begin 11202 lineout
6/27	0025 0130 0625 0812 1550	Reactor plugged, blown out with pressure Begin 11202 test H <sub>2</sub> flowmeter malfunction Reactor shutdown Reactor restarted @ 500°F, 50 ml/hr feed, 10 CFH H <sub>2</sub>
6/28	0230 1030 1530 2330	End 11203 lineout, begin 11203 test End 11203 test, begin 11204 lineout Begin 11204 test End 11204 test, begin 11205 lineout
6/29	0130 0930 1130 1930	End lineout, begin 11205 test End 11205 test, begin 11206 lineout End lineout, begin 11206 test End 11206 test, unit shutdown
7/3		Reactor packed
7/5		Pressure check
7/6		Presulfiding
7/9	0440 1640 2200	Feed initiated Begin 11207 lineout Pump failure

# TABLE III-3 (CONT'D)

# SERIES 112 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/10	0040 0840 1640	End lineout, begin 11207 test End 11207 test, begin 11208 lineout End lineout, begin 11208 test
7/11	0040 0840 1640 1740	End 11208 test, begin 11209 lineout End lineout, begin 11209 test End 11209 test, begin 11210 lineout H <sub>2</sub> rate running low
7/12	0040 0840 1640 1650	End lineout, begin 11210 test End 11210 test, begin 11211 lineout End lineout, begin 11211 test Wet test meter changed
7/13	0040 0840 0940 2050	End ll2ll test, begin ll2l2 lineout End lineout, begin ll2l2 test Problems with preheat controller, test to be restarted Begin test ll2l2
7/14	0450	End 11212 test, reactor shutdown

The reactor was packed to within 34" of the inlet with tabular alumina as postheat, a 13" bed of 100-ml of void-filled 1/8" Co/Mo catalyst was added, and the remainder of the reactor was filled with tabular alumina to make up the preheat zone.

After pressure testing, the catalyst was presulfided to 800°F at full bottle pressure for one hour. The unit was cooled with flowing hydrogen and feed initiated. Temperatures were increased in 50°F increments each 1/2 hour to the desired level.

A chronological run description is shown in Table III-4.

Each material balance product was weighed and fractionated into IBP-400, 400-600 and 600°F+ fractions for analysis.

#### EXPERIMENTAL RUN PREFIX 121

This feedstock was hydrotreated over nickel molybdate catalyst at 500 psig total pressure, over a temperature and LHSV range of  $600-750\,^{\circ}\text{F}$  and  $1-4\,\text{hr}^{-1}$ , respectively.

The reactor was packed with 100-ml (85.3g) of 1/8" catalyst and 100-ml of Ottawa sand. A postheat zone of tabular alumina was added to within 34" of the reactor top, followed by catalyst and sand in 10-ml increments to a total length of 12.75", and the remainder packed with tabular alumina as preheat. Following pressure testing, the

TABLE III-4

# SERIES 113 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/31		Reactor packed, pressure tested
8/1		Presulfiding
0,1	1045	Wet test meter replaced
8/2	0040	Begin 11301 lineout
•	0840	Pump failure
	1040	Pump failure
	1220	Pump failure
	1840	End lineout, begin 11301 test
	2115	Pump failure
	2150	Pump failure
8/3	0340	Test 11301 restarted after pump situation rectified
	1140	End 11301 test, begin 11302 lineout
	1940	End lineout, begin 11302 test
8/4	0340	End 11302 test, begin 11303 lineout
. *	0930	Erratic H2 rate
	1340	End lineout, begin 11303 test
	2140	End 11303 test, begin 11304 lineout
8/5	0540	End lineout, begin 11304 test
·	1340	End 11304 test, begin 11305 lineout
	2140	End lineout, begin 11305 test
8/6	0540	End 11305 test, begin 11306 lineout
•	1340	End lineout, begin 11306 test
	2140	End 11306 test, unit shutdown
7/31		Reactor packed, pressure tested
8/1		Presulfiding
	2320	Begin 11307 lineout
8/2	0920	End lineout, begin 11307 test
•	1720	End 11307 test, begin 11308 lineout

# TABLE III-4 (CONT'D)

# SERIES 113 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
8/3	0120 0520 0615	End lineout, begin 11308 test Pump failure Pump failure
	1100 2020	Feed buret broken, test to be restarted Begin 11308 test
8/4	0420 1220 2020	End 11308 test, begin 11309 lineout End lineout, begin 11309 test End 11309 test, begin 11310 lineout
8/5	6420 1220 2020	End lineout, begin 11310 test End 11310 test, begin 11311 lineout End lineout, begin 11311 test
8/6	0420 1220 2020	End 11311 test, begin 11312 lineout End lineout, begin 11312 test End 11312 test, unit shutdown

catalyst was presulfided with H<sub>2</sub>S at 650°F and bottle pressure for one hour. The unit was cooled to 200°F with flowing hydrogen and feed introduced. Temperatures were raised in 50°F increments every 1/2 hour to 600°F.

A chronological run description is shown in Table III-5.

Each material balance product was weighed and fractionated into IBP-400, 400-600 and 600°F fractions for analysis.

# EXPERIMENTAL RUN PREFIX 122

This feedstock was hydrotreated over nickel molybdate catalyst at 1000 psig total pressure, over a temperature and LHSV range of 600-750°F and 1-4 hr<sup>-1</sup>, respectively.

The reactor was packed with 100-ml (85.3g) of 1/8" catalyst and 100-ml of Ottawa sand. Tabular alumina was charged as postheat to a level 34" from the reactor top, followed by catalyst and sand in 10-ml increments to a total length of 13", and the remainder packed with tabular alumina as preheat. Following pressure testing, the catalyst was presulfided with H<sub>2</sub>S at 650°F and bottle pressure for one hour. The unit was cooled to 200°F with flowing hydrogen and feed introduced. Temperatures were raised in 50°F increments every 1/2 hour to 600°F.

A chronological run description is attached as Table III-6.

TABLE III-5

# SERIES 121 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/17	0340 1240 2040	Start lineout Start material balance, 12101 End material balance, 12101; lineout 12102
7/18	0440 1240 2040	Start material balance, 12102 End material balance, 12102; lineout 12103 Start material balance, 12103
7/19	0440 0750 0822 0900	End material balance, 12103; start lineout for 12104 Temperature, gas rates low, hydrogen supply low Hydrogen flow lost Hydrogen flow re-established; lineout for 12106 instead of 12104 Start material balance, 12106
7/20	2240 0440 0530 0710 0845 1030 1205 1640 2335 2440	Pump failed; diaphragm ruptured, some product contamination could have occurred Pump failed Start material balance, 12105 Pump failed End material balance 12105; lineout 12104
7/21	0840 1640	Start material balance, 12104 End material balance, 12104; cool reactor and shut down
7/22 <b>-</b> 7/25		Reactor repacked with fresh catalyst analogous to initial charge. Presulfiding and start-up procedures same, except presulfided to 750°F

# TABLE III-5 (CONT'D)

# SERIES 121 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/26	0940	Start lineout, 12107
•	1740	Start material balance, 12107
7/27	0140	End material balance, 12107; lineout 12108
	0940	Start material balance 12108
	1540	Frozen cold trap causes gas loss
•	1740	End material balance, 12108; lineout 12109
7/28	0140	Start material balance, 12109
	0220	Gas loss
	0940	End material balance, 12109; lineout 12110
	1740	Start material balance 12110
7/29	0140 -	End material balance, 12110; lineout 12111
•	0940	Start material balance, 12111
	1740	End material balance, 12111; lineout 12112
7/30	0140	Start material balance, 12112
-	0940	End material balance, 12112; unit shut down, cooled and disassembled

TABLE III-6

# SERIES 122 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/15		Reactor packed, pressure tested
7/16	1230	Presulfiding Feed initiated
7/17	0120 0420 0530 0630 0730 1120 1640 1720	Begin lineout 12201 Pump failure Pump failure Pump failure Pump failure Start material balance 12201 Pump failure cancel material balance Start material balance 12201
7/18	0120 0920 1720 2320	End material balance 12201; lineout 12202 Start material balance 12202 End material balance 12202; lineout 12203 Product effluent blockage caused product losses
7/19	0020 0730 0822 0900 0940	Start material balance 12203 Product effluent blockage caused product losses Loss of hydrogen flow Hydrogen flow reinitiated End material balance 12203; lineout 12206
7/20	0120 0920 1720	End material balance 12206; lineout 12205 Start material balance 12205 End material balance 12205; lineout 12204
7/21	0120 0920	Start material balance 12204 End material balance 12204; shutdown and dump reactor
7/22- 7/23		Reactor repacked and pressure tested
7/24- 7/25		Presulfide to 750°F
7/25	1520	Feed started

# TABLE III-6 (CONT'D)

# SERIES 122 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
7/26	0420 1220 2020	Start lineout for 12207 Start material balance 12207 End material balance 12207; lineout 12208
7/27	0420 1220 2020	Start material balance 12208 End material balance 12208; lineout 12209 Start material balance 12209
7/28	0420 1220 2020	End material balance 12209; lineout 12210 Start material balance 12210 End material balance 12210; lineout 12211
7/29	0420 1220 2020	Start material balance 12211 End material balance 12211; lineout 12212 Start material balance 12212
7/30	0420	End material balance 12212, shutdown and dump reactor

Each material balance product was weighed and fractionated into IBP-400, 400-600 and 600°F fractions for analysis.

#### EXPERIMENTAL RUN PREFIX 123

This feedstock was hydrotreated over nickel molybdate catalyst at 2000 psig total pressure, over a temperature and LHSV range of  $600-750\,^{\circ}\text{F}$  and  $1-4\,^{\circ}\text{hr}^{-1}$ , respectively.

The reactor was packed with 100-ml (85.3g) of 1/8" catalyst and 100-ml of Ottawa sand. First tabular alumina was charged as postheat until the top of this zone measured 34" from the reactor inlet. Then catalyst and sand were added in 10-ml increments to a total length of 13.25", and the remainder of the reactor packed with tabular alumina as preheat. Following pressure testing, the catalyst was presulfided with H<sub>2</sub>S at 650°F and bottle pressure for one hour. The unit was cooled to 200°F with flowing hydrogen and feed introduced. Temperatures were raised in 50°F increments every 1/2 hour to 600°F.

A chronological run description is attached as Table III-7.

Each material balance product was stabilized, but no fractionations were performed on these materials.

#### EXPERIMENTAL RESULTS

The full data results are presented in Tables III-8 through III-25.

# TABLE III-7

# SERIES 123 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
8/23		Pack reactor, pressure test
8/24		Presulfide to 650°F
8/27	1200 1250 1920	Feed initiated Pump failure Start lineout 12301
8/28	0320 1120 1920	Start material balance 12301 End material balance 12301; lineout 12302 Start material balance 12302
8/29	0320. 0920 1120 1920	End material balance 12302; lineout 12303 Change wet test meter - low water level Start material balance 12303 End material balance 12303; lineout 12304
8/30	0320 1120 1920	Start material balance 12304 End material balance 12304; lineout 12305 Start material balance 12305
8/31	0320 1120 1920	End material balance 12305; lineout 12306 Start material balance 12306 End material balance 12306; shutdown and cool reactor
9/6		Pack reactor, pressure test, presulfide to 850°F
9/10	0100 1700	Feed introduced Start lineout 12307
9/11	0100 0900 1700	Start material balance 12307 End material balance 12307; lineout 12308 Start material balance 12308

### TABLE III-7 (CONT'D)

### SERIES 123 RUN CHRONOLOGY

DATE	TIME, HRS	ITEM
9/12	0100	End material balance 12308; lineout 12309
	0615	Sudden 400 psig system pressure loss; no apparent reason, repressurized
	0715	System pressure increased to 2300 psig
	0847	Back pressure control valve replaced
	0950	Unit back on line
	1200	Unit pressure drops to 1850 psig; repressurized
	1500	Start material balance 12309
	1900	System pressure loss 150 psig; repressurized
	2315	End material balance 12309; lineout 12310
9/13	0700	Start material balance 12310
	0800	Increase in reactor pressure; corrected
	1000	Gas line plugged
	1200	Gas line plugged
	1300	Gas line plugged
	1500	End material balance 12310; lineout 12311
	1830	Gas line plugged
	2100	Start material balance 12311
	0500 -	End material balance 12311; lineout 12312
	0800	Pump failure
	0900	Pump failure
	1000	Pump failure
	1200	Start material balance 12312
	2000	End material balance 12312, shutdown unit
		dump

PARAMETER VARIAT		N	YDR	ION HYDROTREATING	EATI	U				SERI	SERIES 111	
Period	-	2	3	4	2	9	7	8	6	9	11	12
		0	OPERATING	TING	CON	CONDITIONS	2					
Temperature, <sup>O</sup> F	<b>9</b> 09	965	597	647	849	648	700	669	703	131	752	754
Pressure, PSIG	521	521	ŝ	522	513	523	115	\$15	517	519	800	789
	1.01	2.5	2.8	3.94	1.74	0.92	98.0	1.97	3.81	2.89	1.60	0.80 80
Hydrogen Rate, SCFB	3948	4140	2486	4026	5734	4214	0095	4113	4283	5703	5315	5020
			MATE	MATERIAL BALANCE	BALAI	CE						
		!	Wei	Weight Percent of Feed	ent of F	<b>pee</b> .						
Food	100	100	100	100	100	100	81	81	100	100	100	100
Hydrogen in	6.54	97.9	60.6	6.67	9.50	6.98	7.60	9.90	7.08	9.43	79	6.30
Subtotal In	106.54	106.86	109.09	106.67	109.50	106.98	107.60	106.80	107.08	109.43	106.79	108.30
H20	(0.61)	(0.24)	(0.46)	(0.31)	(0.59)	(0.87)	(1.09)	(0.82)	(0.42)	(0.72)	(1.10)	(1.33)
22	(0.37)	(0.32)	(0.30)	(0.32)	(0.37)	(0.43)	(67.0)	(0.48)	(0.35)	(0.49)	(9.54)	(0.55)
	(0.18)	•	•	(0.11)	(0.26)	(0.23)	(0.35)	(0.37)	(0.27)	(0.13)	(0.45)	(0.70)
Hydrogen Out	6.22	6.39	8.21	5.71	.9.10	6.16	7.75	6.71	7.16		9.02	8.53
<u>5</u>			•		•	•	90.0	•	-		0.21	0.42
Z)	•		•	0.05	•	0.22	0.77	90.0	10.0	•	90.0	1.57
ទ	•		•	0.02	•	0.25	0.95	0.05	0.02	•	90.0	2.39
<b>3</b> (	•	•	•	•	•	0.24	1.37	90.0	-	-	0.02	2.75
5	•	-	•	•	•	0.25	1.91		-	-	0.07	2.43
- 400 F	1.10	0.20	1.14	1.87	0.23	3.48	3.96	4.22	2.55	9.66	5.49	4.31
1,000 -007	21.83	21.26	21.66	22.29	25.14	26.21	26.82	25.11	16.09	30.01	28.29	32.15
+ 4.000	73.69	81.29	67.43	75.25	71.90	<b>68</b> .60	74.31	62.80	79.94	69.30	60.09	43.71
Sublotal Out	104.00	109.70	98.9	105.93	107.59	106.94	119.61	100.64	19.901	•	105.41	100.84
Total Liquids	96.62	102.75	89.73	99.41	97.27	98.29	10.201	92.13	98.58	٠	93.87	80.17
	7 20	. 601	. 8	6								
	2	206.7	Ř	3.5	95.20	100.0	111.4	7.5	99.7	-	96.9	23.1
Hydrogen Consumption, SCFB	194	280	527	277	240	492	•	•	•	•	ı	ı
					brack	brack  brack						

The state of the s

HYDROTREATING
VARIATION
PARAMETER

SERIES 111

Period	-	2	3	4	2	9	7	8	6	10		12
			PRO	PRODUCT	PROF	ERTII	PROPERTIES (Wt. %)	(%:				
TOTAL LIQUID:	26.0	25.5	24.6	25.4	25.6	26.2	28.6	27.1	26.2	17.12	28.9	30.7
# Selfer	0.19	0.21	0.31	0.23	0.19	0.13	90.0	0.0	0.19	90.0	0.03	0.02
** Nitrogen	(11.31)	(1.42)	(1.42)	(1.33)	(1.23)	(1.24)	(1.07)	(1.20)	(1.21)	(1.24)	(11.11)	(F. R.
** Basic Nitrogen	0.93	96.0	10.1	0.95	0.91	1.07	0.82	0.93	0.93	8.0	<b>98.</b> 0	0.70
_			•	•	•	•	•			•	•	
18P-400°F	,	1	•	43.3	•	-	•	43.7	45.7	42.9	64.3	,
	0.31		0.33	0.33	0.25	0.14	0.0049	0.0	0.21	0.12	0.04	0.02
Nitrogen	(1.13)	4	(1.41)	(1.40)	(1.25)	(1.30)	(0.93)	(1.26)	(11.11)	(1.58)	(1.11)	(0.82)
Basic Nitrogen	1.21	#	1 53	1.34	1.27	1.20	0.92	1.28	1.32	1.31	0.98	0 65
400 - 600 °F	35.4	34.5	34.1	34.2	34.3	33.1	34.8	33.8	34.3	33.1	31.7	34.8
Sulter	0.14	0.14	0.31	0.26	0.16	0.13	0.03	0.04	0.14	90.0	0.02	10.0
Nitrogen	(1.16)	(1.39)	(1.25)	(1.12)	(1.16)	(1.15)	(1.01)	(1.19)	(1.41)	(1.22)	(1.11)	(1.21)
Basic Mitrogen	0.97	1.06	1.15	0.98	1.10	1.24	0.93	1.08	1.08	0.97	0.97	0.82
+ 4 0009	22.6	22.2	21.3	23.3	22.5	22.5	25.0	24.5	23.5	24.3	25.3	25.9
Sultur	0.21	0.22	0.31	0.22	0.20	0.24	0.09	0.11	07.0	0.05	9.0	0.05
Nitrogen	(1.36)	(1.61)	(1.54)	(1.41)	(1.26)	(1.27)	(1.10)	(17.1)	(1.13)	(1.23)	(1.07)	(96.0)
Basic Nitrogen	0.92	0.94	0.95	0.93	0.85	1.00	0.78	0.84	0.89	0.84	0.77	0.62
· · · · · · · · · · · · · · · · · · ·			HEIE	HEIEROAIOM	¥	KEMOVAL	(W1.%)	<b>%</b>				
Desulfurization	64.2	60.4	41.5	9.95	64.2	75.5	88.7	83.0	64.2	88.7	94.3	96.23
Denitrogenation	7.7	-0-	-0-	6.3	13.4	12.7	24.6	15.5	14.8	12.7	21.8	26.8

Decaygenation

\* - Insufficient Sample for Analysis
() - Calculated

. Calculated from Fractionation Data

-41-

TABLE III-10

UF 603 600 1G 1092 1081 1G 1092 1.41 1.12 1.41 1.12 1.41 1.12 1.41 1.12 1.41 1.12 1.41 1.12 1.41 1.12 1.68 1.13 1.68 1.14 1.68 1.12 1.51 1.12 1.51 1.13 1.15 1.14 1.68 1.15 1.68 1.15 1.68 1.16 1.16	OPERATING CONDITION 1000 1000 1000 1000 1000 1000 1000 10	CONDITIONS  653 653 652 653 652 656 656 656 656 656 656 656 657 656 656	6 1017 10N 1000 1000 1000 1000 1000 1000	7	æ	6	9	1	64
PSIG PSIG	# 651 000 1000 1000 1000 1000 1000 1000 1	COND 553 596 4564 4564	TION SEE SEE SEE SEE SEE SEE SEE SEE SEE SEE					-	7
ure, UF  PSIG  PSI	10 651 1000 1000 11 4539 1 4539 1 TERIAL	653 998 1,80 4504 BALAN	527 1900 C E	ဟ					
Felia 1092 1081  Fig. 19 141  Fig. 19 141  Fig. 100 100  F	1000 1000 11 4539 17 ERIAL	998 1.80 4564 BALAN	1000 CE	969	697	669	753	753	756
Rate, SCFB	15 4539 TERIAL	4504 BALAN	1232 ICE	1000	1003	1004	1000	166	1000
In the series 100 100 100 100 100 100 100 100 100 10	TERIAL	BALAN	S C E	0.83	1.76	3.52	3.63	1.65	1.01
100 100 100 100 100 100 100 100 100 100	TERIAL	BALAN	E C	4920	8897	4519	4389	4473	4011
100 100 100 100 100 100 100 100 100 100			P						
100 100 100 100 100 100 100 100 100 100	Weight Percent of								
107,15 9,37 1 (0,45) (0,45) (0,45) (0,45) (0,45) (0,45) (0,28) (0,09) (0,28) (0,09) (0,28) (0,09) (0,28) (0,09) (0,28) (0,09) (0,28) (0,09) (0,28) (0			100	100	100	100	100	100	100
107,15 109.37 (0.72) (0.45) (0.45) (0.46) (0.23) (0.46) (0.23) (0.28) (0.06) (0.28) (0.06) (0.28) (0	.60 7.52	95.7	6.84	80.8	7.70	7.42	7.21	7.35	6.63
0.40) (0.45) (0.45) (0.28) (0.69) (0.23) (0.28) (0.06) (0.23) (0.28) (0.06) (0.28) (0.	92		106.84	102.08	107.70	107.42	107.21	107.35	106.63
Out (0.40) (0.23) (0.28) (0.09) (0.28) (0.09	.30) (0.45)	(69.0)	(10.1)	-	(0.87)	(0.80)	(0.85)	(11:12)	3)
0.04 (0.28) (0.08) 6.42 8.84 7 1.68 7 1.20 15.92 041 104.55 111.91	4	(0.45)	(0.44)	•	(0.30)	(0.48)	(0.54)	(0.55)	(0.51)
DF 1.56 11.68 11.00 15.92 1	4	(0.28)	(0.45)	•	(0, 55)	(0.44)	(0.68)	(0.98)	(1.10)
PF	. 6.42	6.81	6.33	4.89	5.86	9 . 10	5.49	5.40	3.30
PF	4	•	•	0.17	60°0		91.0	0.42	0.55
PF 1.56 1.68 1.20 15.92 1 100.11 100.11 100.11 100.11 100.11 111.91 111.91 111.91	4	0.44	0.24	0.70	0.80	0.02	0.24	0.69	1.40
PF	0.0%	0.49	0.24	0.72	0.67	0.03	97.0	9.76	1.52
PF 1.54 1.68 11.20 15.92 83.99 64.71 Out 104.55 111.91	+	0.46	0.23	0.67	0.37	0.02	0.24	0.71	1.33
0ul 104.55 111.91	+	0.52	92	•	0.11	•	0.20	0.58	1.05
0ut 104.55 111.91	+	0.88	2.61	0.84	4.46	4.42	4.5	8.13	7.8
Out 83.99 84.71	7	22.10	25.19	24.32	24.10	24.64	31.23	27.17	35.28
Out 104.55 111.91	7	7	65.64	41.19	67.47	67.24	54.13	53.41	21.67
	7	7	102.64	78.14	105.85	104.19	98 . 26	99.10	106.01
101al Liquids 96.73 102 31 97.02	.02 97.25	95.90	93.44	66.35	<b>%</b> .03	<b>3</b> 6	89.90	88.71	93.95
Closure 97 6 102 3 97 6	4 1001	7 40	3	,	90	9			
	T				20.2	2.5	21.2	77.7	•
Hydrogen Consumption, SCFB 439 319 391	662	391	306		1121	803	1048	1167	2012

( ) - Calculated

_
TREAT
HYDRO
MOIT
VARIA
PARAMETER

SERIES 112

												1
Period	1	2	က	4	5	9	7	8	6	10	11	12
			PROI	PRODUCT	PRO	ERTII	PROPERTIES (WL. %)	*				
TOTAL LIQUID:	27.1	26.1	25.1	26.2	26.6	27.7	31.0	28.7	27.3	30.0	31.4	34.2
# Sulfur	0.15	0.29	0.28	0.21	0.11	0.13	0.02	90.0	90.0	0.03	0.02	0.05
**Nitrogen	(1.23)	(1.32)	(1.37)	(1.30)	(1.24)	(1.12)	(0.82)	(1.01)	(1.10)	(9.0)	(69.0)	(0.55)
**Basic Nitrogen	96.0	0.96	1.06	0.93	0.90	0.91	0.57	0.72	0.0	69.0	0.53	0.43
Oxygen	,		•		•	•	0.24	0.468	0.532	,	0.269	
18P- 400° F 0API	,		,	· ·	•	•	•	45.1	43.2	43.6	44.4	•
Sulfur	0.03	*	0.48	*	0.05	90.0	*	0.03	8.0	0.02	0.05	0.07
Nitrogen	(1.12)	4	(1.62)	*	(1.34)	(1.05)	*	(0.79)	(1.37)	(1.05)	(0.71)	(0.33)
Basic Nitrogen	1.11	*	1.56	#	1.61	1.02	#	9.76	1.14	0.81	0.49	0.17
400-600°F	37.2	36.2	36.5	32.7	35.2	34.8	₹.9€	35.7	34.1	33.4	34.9	36.2
Sultur	0.05	0.26	0.54	0.14	0.05	0.02	0.07	0.02	0.01	0.02	9.0	0.05
Nitrogen	(1.05)	(1.1)	(1.19)	(1.20)	(1.10)	(1.06)	(0.68)	(0.88)	(1.07)	(0.00)	(0.79)	(0, 54)
Basic Nitrogen	1.02	0.984	1.15	1.20	1.00	0.89	0.53	19.0	0.92	0 729	0.581	0.325
+ 4 000 P + 4 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	25.0	23.3	23.4	21.8	22.8	23.8	24.6	24.8	23.5	24.5	26.3	27.8
Sultur	0.16	0.30	0.26	0.22	0.13	0.18	0.07	0.08	0.11	0.03	0.01	0.05
Nitrogen	(1.26)	(1.36)	(1.41)	(1.33)	(1.29)	(1.14)	(0.91)	(1.07)	(1.09)	(0.99)	(0.64)	(0.59)
Basic Nifrogen	0.93	0.95	2	0.88	0.86	0.92	0.59	0.69	0.88	.0.86	0.51	0.55
			HETE	HETEROATOM REMOVAL	M RE	MOVA	.L (Wi. %)	(%)				
Desulturization	11.7	45.3	47.2	4.09	79.2	75.5	86.8	88.7	84.9	94.3	96.2	90.6
Denitrogenation	13.4	7.0	3.9	8.5	12.7	21,1	42.3	78.9	22.5	32.4	57.4	61.3
Deoxygenation	,	'	•	ı	•		80.2	61.3	0.95	,	77.8	

- Deckygenation
+ INSUFFICIENT SAMPLE FOR ANALYSIS
++ CALCULATED FROM FRACTIONATION DATA
( ) CALCULATED

PARAMETER VARIAT		N	YDR	TION HYDROTREATING	EATI	D N				SERIES 113	8 113	
Period	1	2	3	4	2	8	7	8	6	10	11	12
		0	OPERATING	TING	COND	CONDITIONS	2					
Temperature, of	603	109	602	959	651	650	90,	70%	101	762	756	7%
Pressure. PSIG	2002	2007	2023	1997	1997	2014	2013	2005	2032	2006	1994	2017
LHSV. Hr.1	1.00	1.60	3.48	3.75	1.69	0.95	0.92	2.13	3.95	4.14	1.68	1.15
Hydrogen Rate, SCFB	3813	5395	4964	4176	4968	4365	4451	3907	9944	3727	4263	3463
			MATE	MATERIAL BALANCE	BALAF	CE			•			
			Weig	Weight Percent of Feed	ant of F	P.						
Feed	100	901	901	200	100	100	200	100	100	901	100	100
Hedrones In	6:30	8.92	8.21	6.70	8.21	7.22	7.36	97.9	7.38	6.16	7.05	5.73
Subtotal In	106.30	106.92	108.21	106.70	108.21	107.22	107.36	106.46		106.16	107.05	105.73
H20	(0.08)	(0.10)	(0.33)	(0.43)	(0.84)	(1.10)	(1.34)	(1.05)	(0.69)	(0.89)	(1, 28)	(1.40
S. I	(7, 37)	(0.34)	(0.23)	(0.33)	(0.44)	(0.37)	(0.53)	(0.45)		(0.41)	, zs 0)	(0.52)
E E	(0.40)		(0.26)	(0.37)	(0.52)	(0, 59)	(0.91)	(0.11)	(0.70)	(1.05)	9	17.17
Hydrogen Out	4.82	6.68	5.69	5.15	6.07	5.19	5.81	69.4	4.57	4.33	4.36 1	3.25
	•	•	•	•	•	•		0.07	-	0.18	0.31	0.24
CS	•	•		0.29	0.11	0.08	0.27	0.95	-	0.71	0.86	0.96
<b>6</b> 3	•	•		0.25	0.11	90.0	0.58	16.0	•	0.77	2.66	1.03
3	•	•	•	0.48	0.09	0.05	1.11	0.97	-	0.82	2.04	1.03
- Cs	•	•		•	•	0.03	•	0.72		0.40	0 · 79	0.41
1000-400 L	3.8	3.69	3.76	•	4.35	•	6.27	4 .25	3.69	•	7.47	9.62
400 - 600°F	29.47	29.21	28.07	•	27.52	-	31.67	30.71	31.17		35.58	37.40
+ 4,000	<b>76.99</b>	63,67	15 . 99	•	63.73		57.25	58.02	60.39		49.15	46.73
Subjotal Out	105.62	104.09	104.85	105.85	103.79	107.25	105.74	103.82	101.65	106.79	106.78	103.86
Total Liquids	99.95	97.57	98.34	98.55	95.60	99.77	95.19	95.98	95.25	97.23	92.20	93.75
•												•
Closure	99.36	95.57	96.89	99.20	95.92	100.03	98.49	97.52	94.66	100.59	99.75	98.23
Mydrogen Consumption, SCFB	898	1355	1524	1064	1296	1224	935	952	1700	1109	1261	1498

( ) - Calculated

CONTRACT CON

PARAMETER VARIATION HYDROTREATING

PARAMETER	VARIAT	IATION	¥	DRO	HYDROTREATING	Ž	G			is.	SERIES 113	<b>9</b> .
Period	-	2	က	4	3	9	7	8	6	10	11	12
			PRO	PRODUCT	PROF	ERTI	PROPERTIES (WL. %)	(%:				
TOTAL LIQUID:	27.3	26.5	25.7	26.9	27.8	38.7	31.6	29.9	28.2	31.4	33.7	34.7
** Suffer	0.18	0.23	0.32	0.20	0.12	0.16	0.04	0.12	0.13	0.14	0.02	0.0
** Nitrogen	(1.00)	(1.04)	(1.16)	(1.05)	(0.92)	(0.81)	(0.51)	(0.68)	(0.74)	0.57	(0.281)	(0.14)
** Basic Nitrogen	0.78	0.93	0.92	,	0.82		0.32	0.57	0.72		0.22	0.08
Oxygen	•	•		616		•	-	0.402				
idY <sub>0</sub> 1 <sub>0</sub> 001 - d8i		44.5	43.3		6.44		1	45.9	46.4		46.2	46.7
Sulfur	90.0	0.11	0.24		9.05		0.07	90.0	0.07		0.03	0.03
Nitrogen	(1.29)	(1.480)	(1.739)	•	(1.057)	42	(0.104)	(0.535)	(0.725)	*	(0.253)	(0.03)
Basic Nitrogen	1.408	1.613	1.666		1.062		0.102	0.428	0.787		0.111	0.025
400-500°F	34.3	33.7	33.3		34.8		36.9	36.0	35.0		36.8	37.3
Sulfur	80.0	0.15	0.22		0.07		0.01	90.0	0.07		0.03	0.03
Nitrogen	(0.91)	(0.929)	(1.159)	*	(0.964)	*	(0.29)	(0.693)	(0.779)	*	(0.27)	(0.10)
Basic Nilrogen	0.628	0.991	0.965		0.890		0.238	0.564	0.721		0.199	0.066
600°F+	22.8	21.5	21.5		23.0		26.4	24.4	23.3		27.7	28.9
Suffer	0.23		0.36		0.14		90.0	0.15	0.17		00,01	0.05
Nitrogen	(1.030)	(1.0%)	(1.121)	*	(0.889)	*	(0.57)			*	(0.43)	(0.213)
Basic Nitrogen	0.810	0.858	798.0		0.777		0.383	0.590	0.720		0.260	0.102
			HETE	HETEROATOM		REMOVAL	/L (W1.%)	<b>%</b>		-		
Desulturization	0.99	\$6.6	39.6	62.3	77.4	8.69	92.5	77.4	75.5	73.6	96.2	92.5
Denitrogenation	28.6	25.6	17.1	25.0	34.3	42.1	63.6	51.4	47.1	61.0	80.0	90.0
Deoxygenation	•	,	•	23.4	•		•	66.5				•

\* NO PRACTIONATION PERPONHED ON THESE PRODUCTS \*\* VALUES CALCULATED FROM PRACTIONATION DATA ( ) CALCULATED

TABLE III-14

PARAMETER VARIAT		NOL	HYDROTREATING	OTRI	EATI	<b>5</b>				SERI	SERIES 121	
Period	-	2	3	4	2	9	7	8	6	9	11	12
			OPERATING	TING	COND	CONDITIONS	<u>s</u>					
Temoerature. OF	399	99	599	159	650	159	700	902	707	751	751	751
Pressure. PSIG	492	489	499	523	508	523	518	510	512	510	200	519
LHSV. Fr.	0.99	1.89	3.76	3.58	1.73	0.89	0.89	1.82	3.58	3.33	1.00	0.78
Hydrogen Rate, SCFB	4904	4144	4355	4403	1994	4512	4575	4413	1977	4553	9177	5333
			MATE	RIAL	MATERIAL BALANCE	CE						
			Weig	Weight Percent of	ant of F	Fe ed			:			
Food	100	100	100	901	901	100	100	100	100	100	100	100
Hydrogen in	6.71	6.85	7.19	7.27	7.70	7.45	6.89	7.29	7.38	7.52	7.30	9.84
-	106.71	106.85	107.19	107.27	107.70	107.45	106.89	107.29	107.38	107.52	107.30	109.84
H3O	(0.38)	(0.40)	(0.25)	(0.61)	(0.69)	(1.05)	(1.18)	(0.82)	(0.50)	(0.72)	(20.1)	CL-3D
228	(0.14)	(0,06)	(0.0)	(0,29)	(0.34)	(0.46)	(0, 53)	(0.49)	(0.43)	(0.44)	(0.53)	(0, 20)
	(0.40)	(0.23)	Ц	(0.38)	(0.37)	(99.0)	(0.81)	(0.34)	(0.30)	(0.63)	(0.63)	(0.82)
Hydrogen Out	5.62	5.72	5.91	5.85	6.33	5.81	4.67	5.54	5.19	5.37	5.42	8.
	•	•	•	•	•	•	0.10	•	•	0.12	0.31	0.27
C <sub>2</sub>	•	•		•	•	•	0.30	0.24	0,34	0.55	0. 70	1.14
S	•	•	•		•	•	0.78	0.26	0.12	0.55	0.63	1.17
3	•			1	ı	•	2.05	0.91	0.35	0.38	0.43	8
<u></u>	•	•	•	•	•	•	1.67	0.23	0.28	90.0	0.41	0.60
IBP - 400°F	2.70	6.64	3.52	1.13	2.29	2.19	4.41	4.55	4.17	3.27	4.06	3.62
400- 600°F	19.15	8.93	24.66	26.51	28.48	26.39	25.75	24.18	29.55	30.72	29.01	39.03
+ 4 <sub>0</sub> 000	78.32	83.99	70.10	64.92	64.16	58.69	56.75	68.72	64.48	62.05	64.16	51.02
Subtotal Out	106.91	105.97	104.74	99.69	102.66	95.25	99.00	106.28	105.Zi	104.88	107.31	107.36
Total Liquids	100.17	99.56	98.28	92.56	94.93	87.27	86.91	97.45	99.20	8.5	97.23	93.67
									ļ			
Closure	100.2	99.5	27.7	92.9	95.3	98.6	92.6	99.1	98.4	97.5	100.0	27.76
Hydrogen Consumption, SCFB	659	96	778	964	829	992	1473	1058	1325	1303	1137	1720

( ) Calculated from inlet and outlet liquid properties

PARAMETER	VARIAT	IATION		HYDROTREATING	rRE/	NIL	G			Ø	SERIES 121	æ
Period	-	2	3	4	2	9	7	8	6	9	1	12
			PRO	PRODUCT	PRO	PROPERTIES	ES (w	(Wt. %)				
TOTAL LIQUID: OAPI	25.3	27.0	•	25.3	26.0	26.3	28.4	27.5	25.8	27.5	28.6	30.8
Sulfur	0.38	0.47	97.0	0,30	0.23	0,13	0.03	0.07	0.13	0.12	0.03	0.07
•	(1.07)	(17.22)	(1.25)	(1.12)	11.15)	(0.99)	(0,87)	(1.08)	(1.16)	(0.92)	(16.0)	(11.0)
Basic Nitrogen	0.85	1.03	8.0	1.01	1.02	0.95	0.67	0.82	0.89	0.81	0.78	0.62
Oxygen	•	•	-	10.767	-	·	-	(0.501)	•	,	•	•
18P - 400° F 0API		39.9	41.9	•	ı	1	45.0	43.4	41.9	43.6	45.7	1
Sulter	67.0	0.51	99.0	0.45	0.34	0.25	0.01	9.0	0.20	0.18	0.04	0.07
Nitrogen	(1,50)	(1.53)	(0.73)	(1.72)	(1.61)	(1.78)	(0.83)	(1.32)	(1.62)	(1.39)	(1.13)	(0.70)
Basic Nitrogen	1.265	1.262	1.442	1.430	1.54	1.48	0, 708	1.506	1.313)	1.083	0, 930	0.631
400-600°F	33.4	33.7	32.2	33.6	33.0	33.2	34.6	33.9	32.8	33.4	34.0	34.3
Sultar	8.0	85	0 48	S	0 22	13	0 0	8	91 0	13	0 03	2
Nitrogen	(1.08)	1.10	(1.12)	(1.02)	(1.46)	(1.47)	(0.85)	(0.98)	(1.02)	(0.92)	(1.05)	(1.03)
Basic Nitrogen	0.983	0.995	0.907	0.875	1.04	1.05	0.701	0.924	0.905	0.991	0.883	0.727
++ 0009 H	23.0	23.0	21.8	21.5	21.4	22.2	24.2	23.7	22.0	24.0	25.0	25.8
Sullur	0.37	97.0	0.45	9.30	0.23	0.13	0.03	0.07	0.14	0.11	0.03	0.08
Nitrogen	(1.05)	(1.17)	(1.28)	(1.27)	(1.25)	(1.83)	(0.95)	(1.10)	(11.21)	(1.13)	(1.02)	(1.121)
Basic Nitrogen	0.780	988.0	0.873	0.978	0.959	0.965	0.645	0.781	0.826	0.785	0.720	0.501
			HETE	HETEROATOM REMOVAL	M R	MOV		(W1. %)				
Desulturization	38.3	11.3	13.2	43.4	9.95	2.21	94.3	86.8	25.2	17.4	6.49	86.8
Denitrogenation	23.6	12.9	10.7	15.4	17.9	28.3	37.9	22.0	17.1	34.3	35.0	45.0
Decaygenation	.			9.00	1			28.6	  -  -			

PARAMETER VARIAT		H NOI	HYDROTREATING	OTRI	EATI	ت ع				SERIE	SERIES 122	
Period	1	2	က	4	9	9	7	8	6	10	11	12
			OPERATING	TING	COND	CONDITIONS	<u>s</u>					
Temperature, OF	209	209	503	652	459	655	703	70%	70%	7.8	256	752
Pressure. PSIG	1021	1003	1000	1000	1024	1018	966	0001	38	266	966	86
LHSV. Hr.	0.81	1.84	3.53	3.52	1.89	0.826	0.898	1.89	3.33	3.68	1.62	99.0
Hydrogen Rate, SCFB	4937	4375	3948	4536	4264	4850	4476	4258	5012	4378	4971	1509
			MATE	RIAL	MATERIAL BALANCE	CE						
			Weig	ht Perc	Weight Percent of Feed	<b>P</b> • •						
Feed	180	81	901	901	100	201	100	991	901	901	200	100
Hydrogen in	8.15	7.23	6.52	7.49	7.04	8.01	7.39	7.0	8.28	7.24	8.22	10.00
Subtotal in	108.15	ᆫ	106.52	107.49	107.04	10.901	107.39	107.04	106.28	107.24	108.22	110.00
Hoo	(0.68)		(0.33)	(0.44)	(0.83)	(0.0%)	(0.95)	(0.99)	(0.83)	(0.06)	(1.13)	(1.34)
SC	(0.39)	(0.29)		(0.39)	(0.49)				}	(0.51)	(0.54)	(0.53
EIZ.	(0.44)	(0.36)	(0.27)	(0.28)	(09:0)	(0.23)	(86.0)	)	(07.0)	(06.0)	(0.98)	(1.41)
Hydrogen Out	7.22	5.43	5.97.	6.21	5.88	90.9	5.33	5.60	9.76	5.70	6.14	7.7
<u>.</u>	•	•	•	•	•	•	0.12	0.21	0.30	0.11	0.22	
. Z	10.01	•	-	•	•	0.11	1.45	0.26	0.29	98.0	1.04	2.09
<u>.</u>	10.01	•	-	•	•	90.08	2.08	0.27	0.28	0.57	1.01	2.63
3	•	•		•	•	•	2.78	0.30	0.28	0.67	0.93	3.66
CS	•	•	-	1	•	•	2.13	0.33	0.28	16.0	0.72	0.69
18P - 400°F	4.67	3.52	3.30	1.20	3.88	60.4	69' 5	£0.8	3.26	4.24	2.98	9.97
400 - 600°F	27.42	29.11	17.16	25.14	25.88	42.50	33.08	31.96	28.53	33.72	36.71	40.43
600°F +	63.71	67.84	74.22	72.51	66.59	91.36	24 27	29.40	65.52	52.09	54.17	59.28
Subtotal Out	104.75	10.701	101.46	106.17	104.15	19.441	108.40	105.57	107.21	100.24	106.57	129.59
	95.80	100.47	89.46	98.85	96.35	137.89	92.04	96.39	97.31	90.05	93.86	109.68
Closure	8 9.	8.8	95.2	98.8	97.3	136.7	100.9	98.6	99.0	93.5	99.17	117.8
Hydrogen Consumption, SCFB	264	1068	346	778	705	1178	1251	898	920	927	1257	1608

	ĺ
VARIATION HYDROTREATING	
$\stackrel{\smile}{=}$	
<b> -</b>	
4	
ш	
Œ	
<b> -</b>	
0	
Œ	
<b>&gt;</b>	
I	
Z	
ਨ	
=	
<b>H</b>	
4	
7	
7	
>	
Œ	
竝	
F	
Ш	
PARAMETER	
3	
~	
7	
2	

SERIES 122

Derive	-	2	3	4	2	9	7	8	6	9	=	12
			PRO	PRODUCT	PROF	ERTI	PROPERTIES (Wt. %)	3.				
TOTAL LIQUID:	3, 6	36.0	25.3	26.2	27.1	28.9	31.1	29.2	26.8	29.6	31.5	33.7
IdVo	5.73	36.0	33	0.17	0.07	0.09	0.03	0.0	0.08	90.0	0.02	0.01
** Sulfur	(2)	3	3	(81.5)	(%)	(0.88)	(0.64)	(0.86)	(1.10)	(0.73)	(0.63)	(0.22)
** Nifrogen		0 PK	939	0.93	0.855	0.66	0.37	0.59	0.73	0.65	0.41	0.135
THE BASIC NITOGEN					0.497	0.587	0.401	0.359	0.489	•	0.227	
IBP-400F		,	42.0		1	•	•	•	44.4	46.3	•	46.1
IdVo	5	,	22	41.0	0.032		0.03	10.0	0.03	0.07	0.005	0.01
Sulfer	13.50	() 686)	(31,4,1)	(60% 1)	(1, 190)		(0.367)	(0.685)	(1.193)	(0.823)	(0.484)	6.12
Nifogen Boot Nitogen	1.1/3	1 693	1 406	1.13	1.05		0.184	0.462	0.944	0.555	0.321	0.117
400 - 600 F	33.9	33.8	33.6	38.2	34.6	35.5	36.4	35.4	34.5	34.5	36.0	36.2
I N	9	81.0	0.76	0.11	0.035	9.0	0.04	10.0	10.0	0.03	0.04	0.01
Seller	3 5	(808)	(1.093)	(1.258)		(0.775)	(0.545)	(0.810)	(1.158)	(0.957)	(0.624)	(0.17)
Nirogen Book Nirogen	0.812	0 873	756 0	8		0.651	0.337	0.598	0.722	0.721	0.452	0.152
600 F +	22.6	21.6	21.9	24.5	23.5	25.6	25.2	,	23.6	23.8	26.3	27.8
146			25 0	0 10	0.082	0.11	0.02	0.05	0.11	90.0	0.01	0.02
	(1,042)	(1,166)	(1, 252)	(1.144)	(1.158)	(0.962)	(0.722)	(0.897)	(1.070)	(0.919)	(0.659)	0.25)
Mirogen Basia Misoson	0 779	0.831	0.882	0.888	0.854	0.689	0.413	0.596	0.717	0.604	0.380	0.126
			HETE	:ROAT	HETEROATOM REMOVAL	EMOV.		(W1. %)				
			1	9:3	9 70	A 2 0	١ 76	94.3	84.9	88.7	96.2	96.2
Desulturization	67.9	37.70	77.7	45.7	32.9	37.1	X 3	38.6	24.4	47.9	55.0	84.3
Denitrogenation	2.77	-	-	-	58.93	51.49	98.99	70.33	59.59		81.24	
Deoxygenation	-		1									

( ) - Calculated ## - Calculated from Practionation Data

TABLE III-18

The second secon

PARAMETER VAI	VARIATIC	H	ION HYDROTREATING	OTRI	EATI	SN				SER	BERIES 123	
Period	•	7	ဂ	4	9	8	7	8	6	10	11	12
		0	OPERATING	TING	CON	CONDITIONS	<u>S</u>					
Temperature, OF	909	603	ğ	655	889	654	704	707	718	765	753	751
Pressure, PSIG	2012	1963	2009	2003	2017	2009	2009	2050	3661	2		1997
•	0.83	1.01	3.67	3.61	1.60	0.00	1.16	1.68	3. 28	3.85	Γ	25.0
Hydrogen Rate, SCFB	4369	4971	4 506	4613	5661	4502	3791	4833	4994			4901
			MATERIAL		BALANCE	CE						
			Weig	Weight Percent of Feed	ent of F	<b>D</b> •••						
F904	100	100	100	100	100	001	100	100	200	100	100	100
Hydrogen in	•	.8.21	7.45	7.62	9.35	7.43	6.25	7.97	7.69	7.63	8.8	80.8
Subtotal In	107.21	108.21	107.45	107.62	109.35	107.43	106.25	107.97	107.69	107.63	108.00	108.08
120	(0.98)	(0.61)	(0.02)	(0.57)	(0.95)	(0.19)	(1.30)	(1.29)	(1, 09)	(0.97)	(1.27)	(1.43)
H2S	(0.42)	(0.36)	(0.15)	(0.36)	(0.43)	(0.44)	(0.51)	(0.50)	(0.52)	(o. 50)	(0,49)	(0.52)
	69.34)	(9.16)	(0.00)	(0.25)	(0.67)	(0.81)	(1.35)	(1.30)	(1.16)	(1.28)	(1.46)	(1.65)
Hydrogen Out	5.27	6.23	6.67	5.23	7.79	5.76	3.40	9 05		4.74	4.40	4.75
5		•	•	•	•	•	•	90.0	•	0.12	0.01	0.36
20	0.08	97.0	90.0	0.20	0.55	0.16	1.24	1.83	1.71	0.92	2.16	0.43
50	0.12	0.14	•	0.13	0.44	0.12	8	1.31	1.28	0.71	0.45	0.21
36	•	0.15	•	1.19	•	•	0.19	0.69	5.52	0.60	0.27	0.14
-640		9.13		į	0.37	•	0.17	0.25	0.15	0.19	0.20	0.05
167 - 400 V	3.07	2.68	4.17	1.94	2.15	10.9	1.23	8.45	3.86	8.52	17.67	8.57
1 200 - 201	32.11	32.95	37.14	34.41	34.88	36.61	36.04	36.02	25.90	35.88	40.87	39.38
\$00 F +	63.2	63.35	72.80	60,58	61.28	59.67	57.80	55.84	43.54	52.13	42.04	50.57
Sublotal Out	105.62	107.04	120.98	104.86	109.51	110.77	104.25	113.59	90.45	106.36	111.29	108.06
Total Liquids	98.40	96.98	116.11	96.93	98.31	102.29	95.07	100.31	73.30	96.53	100.58	98.52
	8	9	]:									
	C.R	?	117:10	97.4	100.1	103.1	28.1	105.2	84.0	98.8	वस्व	8
Hydrogen Consumption, SCFB	1053	1189	\$14	144	848	1006	1725	1162	1189	1750	2180	2023

( ) - Calculated

TABLE III-19

PARAMETER VAR	_	NOI	HYC	JROT	ATION HYDROTREATING	TING	<b>(15</b>			Š	SERIES 123	
Period	-	2	8	4	3	9	7	8	6	10	-	12
			PRO	PRODUCT	PROF	PROPERTIES (WI. %)	S (w	<b>%</b>				
TOTAL LIQUID:	27.3	1.97	25.7	26.5	28.2	28.6	31.2	31.6	29.6	31.8	33.0	34.7
Suffur	0.14	9.19	0.30	0.20	0.13	0.11	90.0	90.0	0.08	90.0	0.07	0.04
Nitrogen	(1.14)	(1.28)	(1.33)	(1.23)	(0.86)	(0.72)	(0.30)	(0.33)	(0.61)	(96.96)	(0.20)	(0.044)
Basic Nifrogen		1	:			-	-	-	-			1
Oxygen	-	-	;		-	-	1	0.14	-	-		
18P - 400° F	*	*	*	*	*	*	*	*	*	*	*	*
Sulfur												
Nitrogen												
Basic Nitrogen												
400-600°F												
Suffur												
Nitrogen												
Basic Nitrogen												
+40009 F+												
Sullur												
Nitrogen												
Basic Nitrogen												
			HETE	ROATC	HETEROATOM REMOVAL (W1.%)	MOVA	L W.	<b>?</b>				
Desulturization	73.6	64.2	43.4	62.3	75.5	79.2	88 7	7 88 7	84.9	88 7	86.8	97.5
Denitrogenation	18.6	8.6	5.0	12.1	38.6	9.87	78.6	76.4	7.95	74.3	85.7	96.9
Deoxygenation								89.2				

\* SIMILATED DISTILLATION PERFORMED ONLY

SERIES 111 PARAMETER VARIATION HYDROTREATING

<b>I</b>	HYDROGEN	OGE	3	SEN DISTRI	DISTRIBUTION	ET.	N O					
Period	-	7	က	4	2	9	7	8	6	10	=	12
H2S	7	4	2	2	S.	8	,	•	•	,	•	-
NH3	9	0	0	2	12	2		,	•	•	1	•
H <sub>2</sub> O	21	9	9	4	16	12	,	,	1	1	•	•
C1 - C5	0	0	0	1	0	21	,		•	•	1	1
Liquid	69	96	92	91	29	59	•	1	•	•	•	1
Total, SCFB	194	280	527	577	240	492		1				1

TABLE III-21

PARAMETER VARIATION HYDROTREATING

		Č				ŀ						
	HYDROGEN Perce	5 %	Percent	ָה בּי פֿי	Total	DISTRIBUTION of Total	<u>z</u>					
Period	-	2	က	4	3	9	7	8	6	10	11	12
H <sub>2</sub> S	3	3	3	2	4	5	•	2	2	2	2	
NH3	7	3	3	3	8	16	ı	5	9	7	6	9
H <sub>2</sub> 0	11	6	5	5	12	22		5	7	5	9	4
C1 - C5	0	0	2	54	53	34	1	21	1	12	31	33
Liquid	79	85	87	36	23	23		67	84	74	52	56
Total, SCFB	439	319	391	662	391	306	-	1121	803	1048	1187	2015

# PARAMETER VARIATION HYDROTREATING

	HYDROGEN	OGE		DIS	TRI	DISTRIBUTION	NO O					
		Pe	Percent	ō	Total							
Períod	<del></del>	8	က	4	5	9	7	8	0	10	1:1	12
H <sub>2</sub> S	H		1	1	1	1	2	2	1	p=4	2	1
NH3	2	3	2	4	4	5	10	6	7	10	. 10	6
Н20	1	1	1	3	4	9	10	7	3	S	7	9
c1 - C5	0	0	0	12	3	2	23	43	0	29	58	28
Liquid	93	95	96	. 80	88	86	55	39	65	55	23	28
Total, SCFB	868	1355	1524	1064	1296	1224	935	952	1700	1109	1261	1498

TABLE III-23

PARAMETER VARIATION HYDROTREATING

•	HYDROGEN	OGE	Z	DIS	TRI	BUT	DISTRIBUTION					
		ď	ercen	Percent of Total	Total					}		
Period	-	2	3	4	2	6	7	8	6	10	11	12
H <sub>2</sub> S	1	0	0	1	1	2	ι	7	ι	1	2	-
NH <sub>3</sub>	9	7	3	5	5	7	9	3	2	5	9	5
Н20	9	7	2	5	9	7	9	5	3	7	9	2
C <sub>1</sub> ·C <sub>5</sub>	0	0	0	0	0	0	39	11	6	15	25	. 27
Liquid	87	92	56	89	88	84	84	73	85	75	61	62
Total, SCFB	659	686	877	864	829	992	1473	1058 1325	1325	1303	1137	1720

TABLE III-24

PARAMETER VARIATION HYDROTREATING

		(		1	1	1						
·	HYDROGEN			EN DISTRI	DISTRIBUTION	BUT	<u> </u>					
		5		5	loigi							
Period	1	2	3	4	5	9	7	8	6	10	11	12
H <sub>2</sub> S	2	1	2	2	2	1	2	7	2	2	7	1
NH3	8	4	<b>80</b>	4	6	2	80	8	2	. 01	8	6
Н20	11	3	9	4	€	0	2	8	9	9	9	9
c1 - C5	0	0	0	0	0	2	74	18	18	27	35	63
Liquid	79	92	84	06	81	95	11	99	69	55	67	21
Total, SCFB	564	1088	346	778	705	1178	1251	898	920	927	1257	1608

# PARAMETER VARIATION HYDROTREATING

	HYDROGEN	100E	Z	DISTRIBUTION	- E	BUT	<u>N</u>					
		P.	rcen	Percent of Total	[otal							
Period	<del></del>	7	က	4	5	9	2	8	6	10	11	12
H <sub>2</sub> S	1	1	1	1	2	7	1	2	2	-		1
NH <sub>3</sub>	3	1	. 0	2	80	6	8	12	01	∞	7	6
Н20	9	3	0	3	7	8	5	7	9	4	4	2
C1 - C5	2	9		11	16	3	17	15	62	15	17	2
Liquid	88	68	86	83	29	78	69	38	က	72	11	78
Total, SCFB	1053	1189	514	1445	848	1008	1725	1162	1725 1162 1189 1750		2180	2023

### **DISCUSSION**

Note that some product values have been estimated or calculated, as indicated by ( ).

The reader should note that the quoted full-range data are for unstabilized product and may not accurately portray product properties. The fractionation data are designed to provide better quantification of this information.

All synthetic materials are extremely sensitive to product work-up procedures. A particular case in point is the nitrogen and sulfur contents of the fractionated naphtha (IBP-400) products. These data are very high, and result from sulfur and nitrogen washing of the fractionator overhead glassware. During the early stages of the fractionations, sulfur and/or ammonia deposition is visible on the product collection system. As system temperature increases, condensing liquid effectively washes this material into the first collected liquid fractions. This could be corrected by separate stripping operations or other means, but has not been accomplished to date due to extra manpower requirements.

Results from runs 111 and 121 are somewhat erratic, as might be expected for shale oil processing at this low temperature/hydrogen availability. This is due to the fact that at these operating severities, significant quantities of ammonia and sulfur compounds were produced. These materials caused significant pressure regulator problems due to deposition on the stem and seat. Water injection prior to

the high-pressure flash should have been practiced, and will be in the future.

Figures III-2 through III-5 present the desulfurization and denitrogenation response obtained for these materials. Correlations between denitrogenation and desulfurization are presented in Figures III-6 and III-7. An interesting aspect of these figures is the apparent pressure functionality of this relationship for the cobalt molybdate catalyst, whereas the nickel molybdate does not show this apparent response. The second order denitrogenation response to space velocity is shown in Figure III-8.

Finally, hydrogen consumption versus product nitrogen is shown in Figure III-9.

### 2. M-SERIES STUDIES

### SUMMARY

Crude in situ shale oil was processed for a nominal 100 hours at constant (target) operating conditions in order to evaluate initial catalyst stability, as well as determine the length of time required to reach full catalyst equilibration. Studies were made over both cobalt molybdate and nickel molybdate hydrotreating catalysts.

Both catalysts produced denitrogenation ranging 20-30 percent and desulfurization of 60-70 percent, with the nickel molybdate catalyst showing slightly higher heteroatom

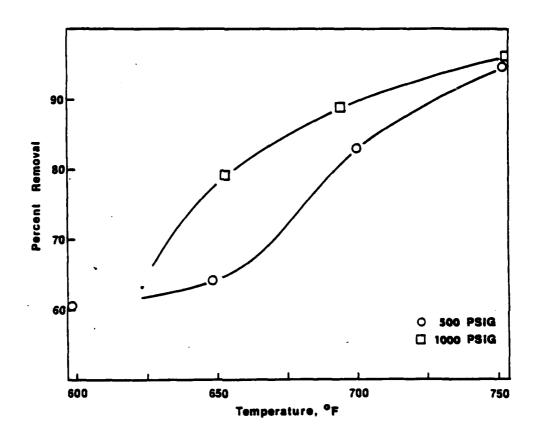


Figure III-2. Temperature Dependence Of Desulfurization At 2 LHSV Cobalt Molybdate Catalyst

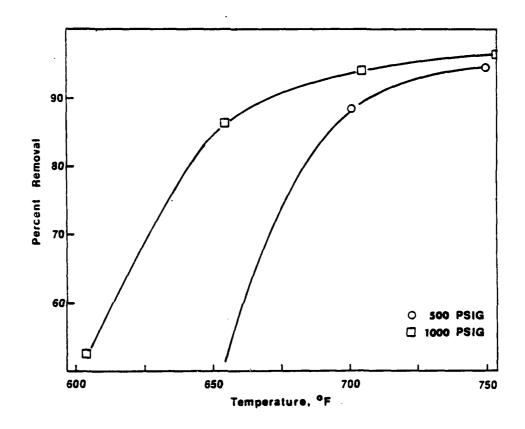


Figure III-3. In Situ Shale Oil Hydrotreating
Over Nickel Molybdate Catalyst: Temperature
Dependence Of Desulfurization At 2 LHSV

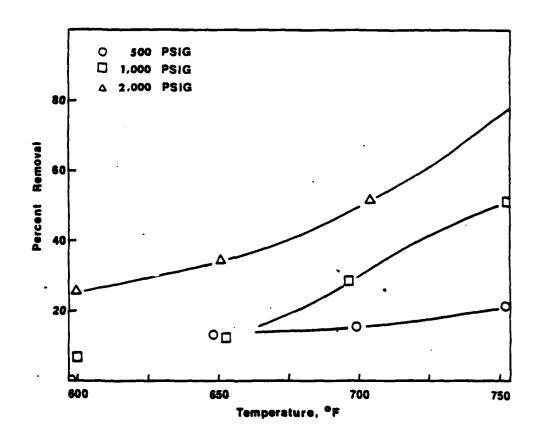
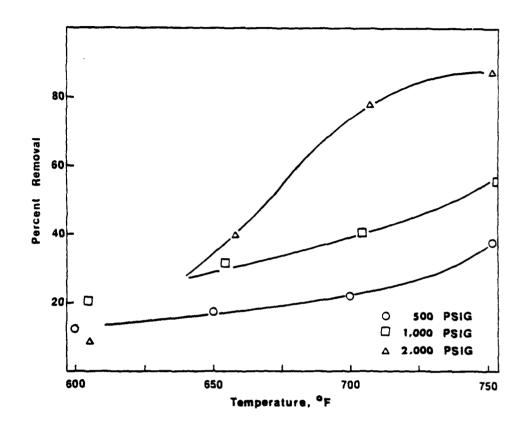
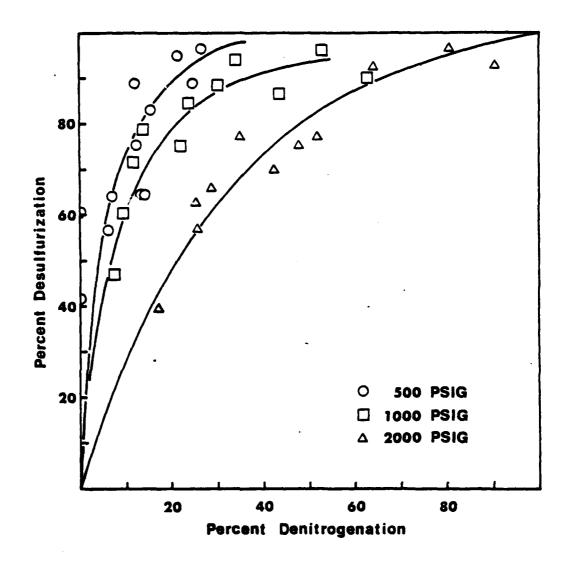


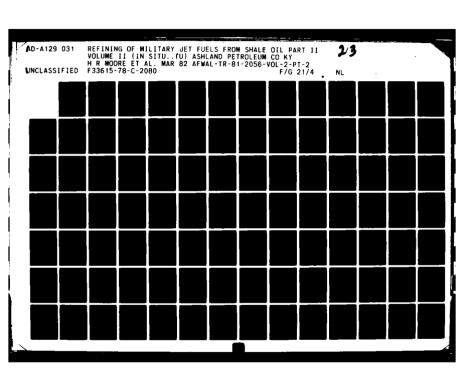
Figure III-4. Temperature Dependence Of Denitrogenation At 2 LHSV Cobalt Molybdate Catalyst

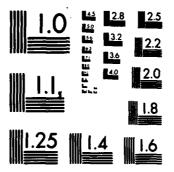


Eigure III-5. In Situ Shale Oil Hydrotreating Over Nickel Molybdate Catalyst: Temperature Dependence Of Denitrogenation At 2 LHSV



\_Figure III-6. In Situ Shale Oil Hydrotreating.
\_Over Cobalt Molybdate Catalyst





MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

14

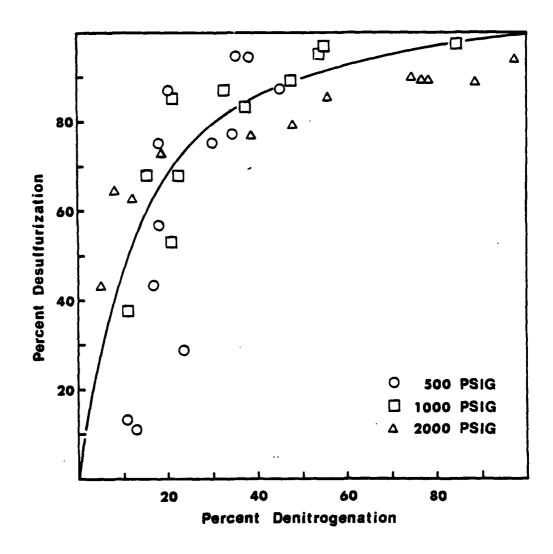


Figure III-7. In Situ Shale Oil Hydrotreating Over Nickel Molybdate Catalyst

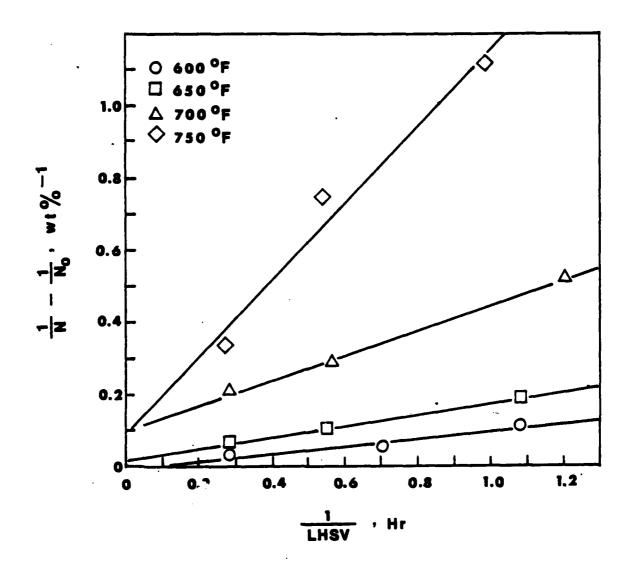


Figure III-8. Second Order Space Velocity
Response Of Denitrogenation At 1000 PSIG
Pressure Over Cobalt Molybdate Catalyst

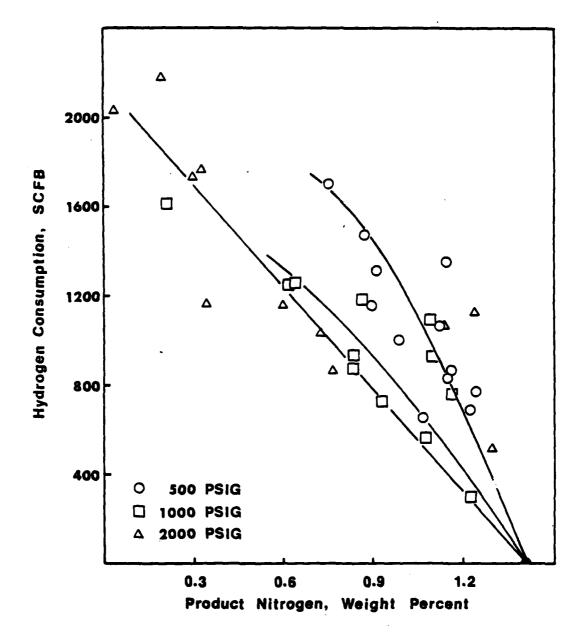


Figure III-9. In Situ Shale Oil Hydrotreating
Over Nickel Molybdate Catalyst: Hydrogen
Consumption Trends

removals at lower apparent hydrogen consumption. For both catalysts, approximately 70 hours of operation were required for (apparent) equilibration of catalyst activity.

### OBJECTIVES

Objectives for these runs were: (1) examine initial catalyst stability parameters with respect to heteroatom removal, (2) determine the apparent time required for catalyst equilibration, (3) produce comparative data for cobalt molybdate and nickel molybdate catalysts and (4) produce adequate quantities of products for further processing and detailed evaluation.

### UNIT DESCRIPTION

Nominal 1" I.D. trickle-flow reactors, described in Section 1, were used for the M-ll study (cobalt molybdate catalyst). No external guardbed was used, as an internal tabular alumina preheat section provided this purpose.

Run M-12 was also performed on nominal 1" hydrotreating systems, but with a slightly different configuration.

These systems are described in detail in Section VII.

### EXPERIMENTAL

Raw in situ shale oil was hydrotreated at constant targeted conditions of 650°F, 2 LHSV, 3000 SCFB hydrogen circulation and 1000 psig total system pressure for both units. The charge was not filtered or pretreated in any way.

For run M-11 the reactor was packed to within 33.75" of reactor inlet with tabular alumina as postheat, onto which an 18.75" bed of 150-ml of 1/8" Co/Mo catalyst void-filled with 150-ml of Ottawa sand was added. The remaining space in the reactor was then filled with tabular alumina as preheat.

Run M-12 was accomplished with a slightly smaller reactor which was charged with 90-ml of nickel molybdate catalyst. Packing and void-filling procedures were followed in a manner analogous to that used for run M-11.

Pressure testing was accomplished to 1500 psig with nitrogen and hydrogen, after which the catalyst was sulfided to  $700\,^{\circ}\text{F}$  with  $\text{H}_2\text{S}$  at bottle pressure for four hours. The unit was cooled and purged of  $\text{H}_2\text{S}$  with flowing hydrogen, the hydrogen flow rate set and the unit pressured to 1000 psig.

Samples from each material balance period were stabilized by heating under vacuum to 300°F for at least two hours. This method proved most effective at removing entrained water from the sample. Stabilized samples were submitted for total nitrogen and sulfur analysis. Material balances were performed every twelve hours early in each run and at 24-hour intervals thereafter.

Hydrotreated product from run M-11 was composited in bulk, mixed thoroughly and split into halves. Each half was fractionated on a batch still: one half to a 500°F nominal cut point and the other to a 600°F cut point, producing samples M-111 and M-112, respectively.

Product from run M-12 was also composited and fractionated at 600°F atmospheric equivalent on the batch still to produce sample M-121.

### FEEDSTOCK

Sufficient raw in situ shale oil was mixed and with-drawn from drum #06-SH-95-01 and supplied to lab personnel as feed to hydrotreating runs M-11 and M-12. The raw oil was obtained from the original 96-drum Air Force shipment and its properties are shown in Table III-26.

### RESULTS

Hydrotreating material balances for both runs are shown in Tables III-27 and III-30. Fractionation of the products and the detailed analysis of each is delineated in Tables III-28, III-29 and III-31. The reader should note that oxygen balances are calculated, based on data obtained during Part 1 and Part 3 of this section, and are not measured values.

### DISCUSSION

Both runs were stable over approximately 100 hours of operation. Figures III-10 and III-11 show material balance sample data versus time on stream, not including a nominal eight hour break-in and lineout period. Some scatter is noted in the M-12 denitrogenation data in particular, but overall results are reasonably smooth. Two intriguing trends show in both data sets: initial operation (hours

## M-SERIES HYDROTREATING FEED PROPERTIES OF OCCIDENTAL IN SITU SHALE OIL

Gravity, *API	23.7
Sulfur, Wt. %	0.53
Nitrogen, Wt. %	1.42
Oxvgen. Wt. %	1.21

TABLE III-27

## CRUDE SHALE OIL HYDROTREATING SERIES M-11

Period	1	2	3	4	5	Total
OPE	RATIN	NG C	ONDIT	IONS		
Temperature, <sup>O</sup> F	652	651	651	651	651	651
Pressure, PSIG	1027	1031	2031	1024	1025	1028
LHSV, Hr -1	1.90	1.76	1 67	1.78	1.79	1.78
H <sub>2</sub> Rate, SCFB	3149	3384	3614	3403	3301	3370
M	ATERI	AL B	ALAN	CE		
	Weight	Percent	of Feed			·
Feed	100	100	100	100	100	100
H <sub>2</sub> In	5.21	5. 59	5.97	5.62	5.46	5.61
Subtotal in	105.21	105.59	105.97	105.62	105.46	105.61
H <sub>2</sub> O	(0.84)	(0.85)	(0.84)	(0.83)	(0.83)	(0.84)
NH <sub>3</sub>	(0.52)	(0.45)	(0.53)	(0.56)	(0.49)	(0.46)
H <sub>2</sub> S	(0.33)	(0.32)	(0.39)	(0.40)	(0.37)	(0.39)
H <sub>2</sub> Out	4.28	4.07	4.25	4.25	4.28	4.24
C <sub>1</sub>		-				
C <sub>2</sub>	0.05	0.08	0.26	0.01	0.01	0.09
C <sub>3</sub>	0.04	0.06	0.19	0.01	0.08	0.08
C4	0.03	0.10	0.04	-	1.00	0.28
C <sub>5</sub>	0.03	-	-	•	-	0.01
Stabilized Liquid	96.51	94.70	95.70	98.77	98.91	97.25
Subtotal	102.63	100.63	102.2	104.83	105.97	103.64
Closure	97.5	95.3	96.4	99.3	100.5	98.1
Hydrogen Consumption, SCFB	559	923	1042	829	709	828
PR	ODUC	T PR	OPER	TIES		
OAPI	27.1	27.1	27.0	26.7	26.7	26.9
Sulfur, Wt %	0.23	. 0.24	0.17	0.16	0.18	0.17
Nitrogen, Wt %	(1.011)	(1.088)	(1.008)	(0.950)	(1.008)	(1.056)
Oxygen, Wt %		-	-	-		(0.47)

TABLE III-28

# CRUDE SHALE OIL HYDROTREATING SERIES M-11 PRODUCT FRACTIONATION M-111

Fraction, °F	IBP-500	500 +
Yield, Wt % of Feed	17.7	79.2
Yield, LV % of Feed	20.0	80.5
OAPI	43.5	26.2
Carbon, Wt %	84.98	86.90
Hydrogen, Wt %	12.98	11.43
Nitrogen, Wt %	(1.07)	(1.01)
Basic Nitrogen, Wt %	1.077	0.887
Sulfur, Wt %	G 11	0.17
Oxygen, Wt %	•	•
Phenois, Wt %	•	-
Ramsbottom Carbon	-	1.06
Paraffins (P+N), LV %	(76.2)	<u>-</u> ·
Olefins, LV %	12.0	-
Naphthenes, LV %	•	-
Aromatics, LV %	11.8	-
Gross Heating Value, Btu/lb	19,720	19,068

# CRUDE SHALE OIL HYDROTREATING SERIES M-11 PRODUCT FRACTIONATION M-112

Fraction, °F	IBP-600	600 +
Yield, Wt % of Feed	36.53	60.72
Yield, LV % of Feed	39.5	60.0
<sup>o</sup> API	36.8	22.4
Carbon, Wt %	85.84	86.65
Hydrogen, Wt %	12.51	11.70
Nitrogen, Wt %	(1.15)	(1.09)
Basic Nitrogen, Wt %	0.942-	0.816
Sulfur, Wt %	0.10	0.17
Oxygen, Wt %	•	•
Phenois, Wt %	٠	-
Ramsbottom Carbon		1.34
Paraffins (P+N), LV %	(71.1)	•
Olefins, LV %	9.4	•
Naphthenes, LV %	•	_
Aromatics, LV %	19.5	•
Gross Heating Value, Btu/lb	19,321	18,977

# CRUDE SHALE OIL HYDROTREATING SERIES M-12

Period	1	2	3	4	5	Total
<u></u>	RATIN	NG C	ONDIT	IONS		
Temperature, <sup>O</sup> F	647	650	651	650	649	649
Pressure, PSIG	1000	1000	1010	1010	1022	1008
LHSV, Hr <sup>-1</sup>	2.04	2.01	2.04	1.98	1.96	2.01
H <sub>2</sub> Rate, SCFB	2950	2992	2944	3033	3064	2997
M	ATERI	AL B	ALAN	CE		
	Weight	Percent	of Feed	· · ·		
Feed	100	100	100	100	100	100
H <sub>2</sub> In	4.90	4.97	4.90	5.12	5.09	5.00
Subtotal In	104.90	104.97	104.90	105.12	105.09	105.00
H <sub>2</sub> O	(C.81)	(0.82)	(0.82)	(0.81)	(0.81)	(0.81)
NH <sub>3</sub>	(0.68)	(0.34)	(0.46)	(0.57)	(0.45)	(0.50)
H <sub>2</sub> S	(0.40)	(0.42)	(0.41)	(0.43)	(0.43)	(0.42)
H <sub>2</sub> Out	3.99	4.02	4.00	4.19		4.05
C <sub>1</sub>		-	-		-	-
C <sub>2</sub>	0.07	0.16	0.06	0.08		0.09
C <sub>3</sub>	0.05	0.14	0 05	0.08	-	0.08
C4	0.06	0.13	0.03	0.07	•	0.07
C <sub>5</sub>	0.02	0.21	0.04	0.08		0.09
Stabilized Liquid	97.97	96.94	95.81	97.34	97.25	97.06
Subtotal	104.05	103.18	101.68	103.65		103.17
Closure	99.2	98.3	96.9	98.6		98.3
Hydrogen Consumption, SCFB	545	573	542	511	•	573
	ODUC.	T PR	OPER	TIES		
OAPI	27.3	27.3	27.2	27.3	27.3	27.3
Sulfur, Wt %	0.16	0.14	0.15	0.13	0.13	0.14
Nitrogen, Wt %	(0.856)	(1.153)	(1.068)	(0.953)	(1.061)	(1.02)
Oxygen, Wt %	•	-	-	•	-	(0.4)

# CRUDE SHALE OIL HYDROTREATING SERIES M-12 PRODUCT FRACTIONATION M-121

Fraction, °F	IBP-600	600 +
Yield, Wt % of Feed	32.1	65.0
Yield, LV % of Feed	34.4	65.0
OAPI	35.0	23.9
Carbon, Wi %	85.68	87.82·
Hydrogen, Wt %	13.23	11.31
Nitrogen, Wt %	(1.03)	(1.02)
Basic Nitrogen, Wt %	0.887	0.737
Sulfur, Wt %	0.09	0.14
Oxygen, Wt %		-
Phenois, Wt %	•	-
Ramsbottom Carbon	•	1.53
Paraffins (P+N), LV %	(71.1)	
Olefins, LV %	6 .8	_
Naphthenes, LV %	-	_
Aromatics, LV %	22.1	•
Gross Heating Value, Btu/lb	19,286	18,938
}		

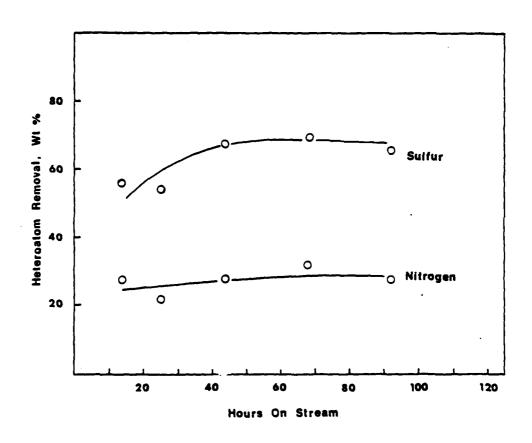
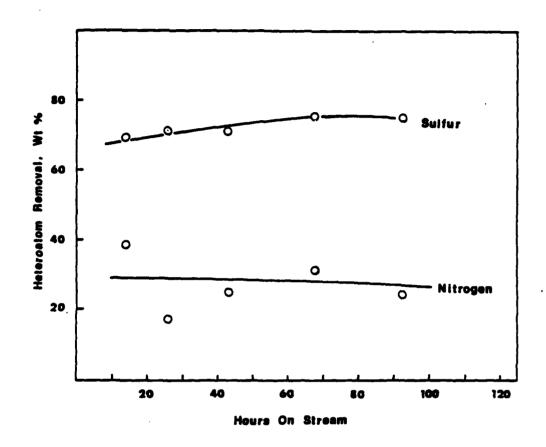


Figure III-10. Heteroatom Removal Stability: Crude Shale Oil Hydrotreating - Series M-11



Eigure III-11. Heteroatom Removal Stability: Crude Shale Oil Hydrotreating - Series M-12

10-20) were followed by a data point of lower apparent activity, then followed by a period of gradually increasing activity. This increasing activity continued for both samples until apparent heteroatom removal peaked in the region of seventy hours on-stream. The data is of particular interest because these runs were performed independently by two different groups, on two different sets of equipment, without knowledge of the results from the other lab. The relative agreement between the first data point and the 70-hour data tends to confirm the parameter variation studies as valid for screening purposes.

The nickel molybdate catalyst appears to provide higher heteroatom removal at lower hydrogen consumption. One factor involved may have been the lower actual space velocity used in the cobalt molybdate runs. Detailed examination of the data indicates, however, that the probable cause was a malfunctioning gas meter. Weighted average hydrogen content of the cobalt molybdate product is 12.00 versus 11.94 for the nickel molybdate product (by hydrogen analyses), which, when given the similar gas production, indicates an erroneously high apparent hydrogen consumption for the M-11 data.

A further interesting observation is that the nitrogen content for the distillate  $(I-600^{\circ}F)$  is higher than for the bottoms product in both runs.

# 3. CRUDE SHALE HYDROTREATER AGING

### SUMMARY

A nominal 1-inch I.D. trickle flow reactor, packed with cobalt molybdate hydrotreating catalyst and provided with parallel guardbeds for arsenic removal, was operated for a period of 28 days (677 total hours) with crude in situ shale oil feedstock. Results from this run showed good stability, particularly in light of several operating upsets which occurred. This run successfully demonstrated the ability of the crude shale hydrotreater to operate for extended periods at low to moderate total pressure.

### **OBJECTIVES**

Objectives for this run were to: (1) demonstrate catalyst stability, in particular, for sulfur removal and hydrogen consumption, over an extended period of time, (2) confirm earlier results obtained at shorter lineout and run lengths and (3) produce adequate quantities of products for both detailed analyses and further processing.

### UNIT DESCRIPTION

The 1" nominal laboratory reactor utilized for this study has been adequately described in the earlier sections of this chapter. Due to the planned duration of the run, the reactor system was modified as shown in Figure III-12 to include parallel guardbeds for catalyst protection. The guardbeds were installed to allow independent operation, with the capability of replacing either unit while the

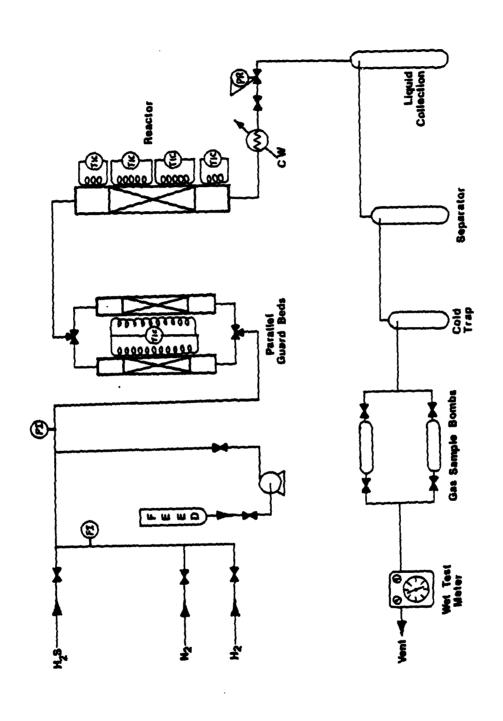


Figure III-12. Modified Crude Shale Hydrotreater For 30-Day Runs

other was on stream. The guardbeds were operated upflow, with both hydrogen and feedstock flowing through the unit. Pressure control was provided through the reactor outlet pressure control valve, allowing the capability to take any pressure drop across the guardbed(s) necessary up to the maximum discharge pressure capabilities of the pump. Temperature control was provided by an external heating tape and PID controller, driven by external skin temperature thermocouples.

# FEEDSTOCK

All feedstock for this run was from a single, reserved drum of in situ shale oil supplied by the Air Force. The drum, coded \$06-SH-94-01, was heated to approximately 100°F by a drum heater and five gallon samples were removed when necessary to replenish the operating supply. No dewatering or other pre-processing was practiced. Properties of the feed drum, taken for a well mixed grab sample from the full drum, are shown in Table III-32.

### EXPERIMENTAL

Sample #06-SH-94-01 Occidental crude shale oil was hydrotreated at nominal conditions of 675°F, 1000 psig, 2.00 LHSV and 3000 SCFB hydrogen circulation by the Synthetic Oils Research Group.

The reactor was packed first with inert tabular alumina to a distance 33-7/8° from the reactor inlet. The catalyst bed consisted of 100-ml (72.3g) of catalyst and

# CRUDE SHALE HYDROTREATER AGING OCCIDENTAL CRUDE OIL (06-SH-94-01) FEED PROPERTIES

Gravity, °API	23.5
Simulated Distillation, °F	
IBP 10% 50% 90% EP	344 470 672 874 950
Elemental, Wt. %	
Hydrogen Sulfur Total nitrogen Oxygen Fe, ppm Ni, ppm Va, ppm As, ppm C1, ppm	11.7 0.64 1.41 1.13 71 13 2 39 23.9
Ramsbottom Carbon, Wt. %	1.40
Basic Nitrogen	0.875
Color	8+
Paraffins, Wt. %	22.1
Naphthenes, Wt. %	31.6
Aromatics, Wt. %	46.3

100-ml of Ottawa sand. The sand and catalyst were added alternately in 10-ml increments to a level of 19" from the reactor inlet. The remaining preheat section, as measured from the reactor inlet, was tabular alumina to 11.75", activated alumina to 8.5" and the remaining space in the reactor was packed with tabular alumina. Thermocouples were located as follows: preheat at 47", upper bed at 42", lower bed at 34.5", post heat at 29", all with respect to the reactor outlet.

Following pressure testing to 1500 psig with nitrogen, then hydrogen, the catalyst was presulfided with H<sub>2</sub>S at bottle pressure and 725°F for two hours. The reactor was allowed to cool nearly to room temperature with flowing hydrogen, hydrogen rate was set and the reactor was pressurized to 1000 psig. Feed was initiated and the entire system was slowly brought up to the desired operating conditions. The unit was lined out for 5.5 hours before the start of the first material balance period.

Two parallel guardbeds were installed in-line directly before the hydrotreater. The parallel arrangement provided the convenience of performing guardbed maintenance without interrupting unit operations. All oil and hydrogen which passed through the guardbed were controlled to maintain operating conditions of 500°F and reactor pressure, with a bed of 170-ml of 1/8" activated alumina. The beds were intended to remove arsenic, iron and other metals from the shale oil feed before entering the reactor preheat zone.

Conditions were monitored continuously and recorded hourly to maintain a 675°F bed temperature, 1000 psig reactor pressure, 2 LHSV and 3000 SCFB (3.77 SCF/Hr) hydrogen circulation.

From each (nominal 24 hour) material balance, a small sample was stripped at 300-350°F for at least two hours. The remaining product was combined into composite cans. Composites 1, 3, 5 and 7 were stabilized, and the stabilized product provided to the pilot plant to be cracked on the FCR unit.

### RESULTS

A chronological description of the run is shown in Table III-33. The major problem encountered during the run was pump failure.

Operating results are shown in Table III-34 and Figures III-13 and III-14. Note that the reactor temperature is an hourly average temperature of the upper and lower portions of the bed, for each 24 hour period. Also, reactor pressure is an average pressure of the reactor inlet and outlet for the same duration of time.

Product fractionations were performed on small grab samples of several composite cans. These results are presented in Tables III-35, III-36 and III-37. Analysis of reactor catalyst bed and preheat alumina can be seen in Table III-38.

# CHRONOLOGICAL RUN DESCRIPTION

DATE	TIME, HRS	ITEM
10/5/79	0525 0725-1030 1037	Start material balance - period 1 Difficulty maintaining constant H <sub>2</sub> flow By-passed H <sub>2</sub> flow sight glass to stabi- lize H <sub>2</sub> flow rate
	1050 2330	Guardbed reactor 490°F at top Repair pump
10/6/79	0525 0525	End material balance - period 1 Begin material balance - period 2
10/7/79	0525 0525	End material balance - period 2 Begin material balance - period 3
10/8/79	0525 0525	End material balance - period 3 Begin material bálance - period 4
10/9/79	0525 0525 1225	End material balance - period 4 Begin material balance - period 5 Plug in cold trap causing wet test meter to malfunction
	1545-2100	Cold trap center post plugged causing high separator pressure
10/10/79	0525 0525 0740	End material balance - period 5 Begin material balance - period 6 Plug in cold trap causing gas readings to be low
	1820	Priming adjustment valve fell off and pump rate took off
10/11/79	0525 0525 0525 0525 0825 0905 1200 1800 1820	End material balance - period 6 Begin material balance - period 7 Pump failed to maintain feed rate Feed buret broken Pump stopped Pump stopped Feed buret leaking Hydrogen rate fluctuating Switched guard beds
	1920-2225	Pump lost prime

# TABLE III-33 (CONT'D)

# CHRONOLOGICAL RUN DESCRIPTION

DATE	TIME, HRS	ITEM
10/12/79	0430	Vent valve fitting vibrated loose causing loss of pump
	0525	End material balance - period 7
	0525	Begin material balance - period 8
	0735	Off-gas rate low
	1105	Pump back on
	1240-1255	Pump down
10/13/79	0045	Switched guardbeds
	0525	End material balance - period 8
	0525	Begin material balance - period 9
	0935	Preheat temperature and bed temperature decreased
	1230	Pump stopped and replaced
	1500	Pump reprimed
10/14/79	0430	Pump replaced
	0525	End material balance - period 9
•	0525	Begin material balance - period 10
10/15/79	0525	End material balance - period 10
	0525	Begin material balance - period 11
	1325	High level in sight glass
10/16/79	0525	End material balance - period 11
	0525	Begin material balance - period 12
	0725	Adjusted H2 rate
	1335	Added water to wet test meter to bring up to proper level
	1825	· Water in product when drained
10/17/79	0325	Pump lost prime
	0400	Replaced pump
	0445	Replaced second pump
	0525	End material balance - period 12
	0525	Begin material balance - period 13
10/18/79	0525	End material balance - period 13
=-, =-, .	0525	Begin material balance - period 14
		Factor Comments

# TABLE III-33 (CONT'D)

# CHRONOLOGICAL RUN DESCRIPTION

DATE	TIME, HRS	ITEM
10/10/70		End make with helpings a posited 14
10/19/79	0525	End material balance - period 14
	0525 0525-0625	Begin material balance - period 15 Feed pump down
	1325	Feed pump lost prime
	1400	Pump replaced. Pump pressure got too
	1400	high before opening feed valve. Pres- sure gauge reads extremely high
	1825	Pump stopped - replaced pump
	2325	Replaced second pump
10/00/70	0.505	• •
10/20/79	0525	End material balance - period 15
	0525	Begin material balance - period 16
10/21/79	0125-0450	Extreme pressures were noted on pump
		pressure gauge. Hydrogen rate con-
		tinues at an increased rate. Pump
•		failed. No oil flowing; lines plug-
	• 0500	ged. Guardbed temperatures shut off Replaced leaking pump
	0500	Switched guardbeds
	0825	Pump lost prime
•	0925	End material balance - period 16
	0925	Begin material balance - period 17
	1245	Pump lost prime
10/00/70		-
10/22/79	0200	Switch guardbeds
	0525	End material balance - period 17
	0525	Begin material balance - period 18
10/23/79	0525	End material balance - period 18
	0525	Begin material balance - period 19
	0755	Plug in inlet size of reactor
	1000	Pump lost prime
	1845	Pump lost prime
10/24/79	0525	End material balance - period 19
	0525	Begin material balance - period 20
•	0850	Pump pumping pump oil through discharge.  Replaced pump. Leak developed around feed buret bottom. New pump would not
		pump
	1000-0025	All temperatures decreased. Rebuilding pump
	1545	Feed line to guardbed replaced
	2025	End lineout of material balance - period 20
	2025	Begin material balance - period 20

# TABLE III-33 (CONT'D)

# CHRONOLOGICAL RUN DESCRIPTION

DATE	TIME, HRS	ITEM
10/25/79	0525 0525 0825 1010 1225 1325	End material balance - period 20 Begin material balance - period 21 Plug in cold trap Pump lost prime Pump replaced Exothermic reaction when feed was introduced Pump replaced
10/26/79	0525 0525 1040 1040 1355 1925-2015 2015 2015	End material balance - period 21 Begin material balance - period 22 Pump leaking - replaced with two pumps Rebuilt pump Exothermic reaction - heater temperature decreased Pump failure - replaced vent screw Screw head broke off Replaced pump
10/27/79	0250 0525 0530 0530	Replaced pump twice. Plugged suction piping Pump ran dry End material balance - period 22 Begin material balance - period 23
10/28/79	0525 0525	End material balance - period 23 Begin material balance - period 24
10/29/79	0525 0525	End material balance - period 24 Begin material balance - period 25
10/30/79	0525 0525 2325 2340	End material balance - period 25 Begin material balance - period 26 Replace pump Pump leaks around checks - removed and cleaned pump
10/31/79	0525 0525	End material balance - period 26 Begin material balance - period 27
11/1/78	0525 0525 1910 1925	End material balance - period 27 Begin material balance - period 28 Pump down Replaced pump
11/2/79	0525	End material balance - period 28 shutdown

TABLE III-34

Doubel	-	c	6		4	q	•	C				
20191		4	2	•	2	2		0	2	2		12
		0	OPERATING	TING	CON	CONDITIONS	IS					
Temperature, "F	099	677	119	113	119	9/9	9/9	299	678	678	678	679
Pressure, PSIG	1000	999	1005	1013	1020	1015	1006	1012	1016	1010	1601	1024
LHSV, H-1	2.00	1.76	1.85	1.67	2.06	1.15	1.59	2.6	1.78	2.7	200	7.98
Hydrogen Rate, SCFB	2987	3403	3247	3268	2916	3412	3886	3654	3375	2568	2902	3025
			MATE	RIAL	MATERIAL BALANCE	NCE						
			Weig	the Perci	Weight Percent of Feed	D						
Feed	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Hydrogen In	4.93	5.62	5.36	5.36	4.81	5.63	6.41	6.03	5.57	4.24	6.79	4.99
Subjoin in	104.93	105.62	105.36	105.36	104.81	105.63	106.41	106.03	105.57	104.24	104.79	104.99
H20	(0.61)	(0.78)	(0.82)	(0.79)	(0.82)	(0.82)	(0.64)	•	(0.84)	(0.82)	(0.83)	(0.94)
<b>128</b>	(0.53)	(0.52)	(0.53)	(0.52)	(0.55)	(0.46)	(0. 52)		(0.58)	(0.52)	(0.48)	(05.00)
EX	(0.29)	(0.20)	(0.31)	(0.23)	(0.20)	(0.17)	(0.34)		(0.34)	(0.21)	(0.28)	(0.28)
Hydrogen Out	3.26	3.75	3.65	4.06	3.86	4.13	5.25		4.26	3.35	3.66	4.07
5	0.64	0.58	0.14	0.03		'	٠	•	0.02	•	•	0.11
C2	0.24	9.30	0.34	0.02	0.07	0.20	0.70	•	1.42	07.0	0,40	0.28
Ĉ.	0.24	0.25	0.27	0.03	9.0	0.15	0.78	•	1.03	0.36	0.23	0.26
3	91.0	0.05	0.07	•	0.05	0.77	1.20		0.14	75 T	0.10	0.11
2	0.02				•	0.03			•	•	0.03	0.06
Stabilized Liquid	101.52	108.58	100.67	105.92	99.35	28.24	94.86	98.63	94.89	100.41	97.15	95.71
-1	107.74	115.01	106.80	111.60	105.06	106.27	104.49		103.52	107.91	103.16	102.22
Closure, %wt	102.7	106.9	101.4	105.9	100.2	100.6	98.2	•	98.1	103.5	98.4	\$7.4
Ideo	27.4	11.1	27.8	28.3	27.7	27.4	27.4	27.1	27.4	27.6	27.2	27.5
Selfer, % wi	•	•	•	91.0	0.12	0.18	0.16	-	0.10	0.15	0,19	0.18
Mitrogen, % wt	•			(113)	(1.25)	(1.28)	(1.19)	٠	(1.19)	(1.23)	(1.21)	(1.23)
Basic Nitrogen, %wt	•			0.77	·	98.0	0.81		•	0.84		
Hydragen Consumption, SCFB	1013	1129	1036	786	519	912	706	•	161	13	687	262

( ) - Calculated

TABLE III-34 (Cont'd)

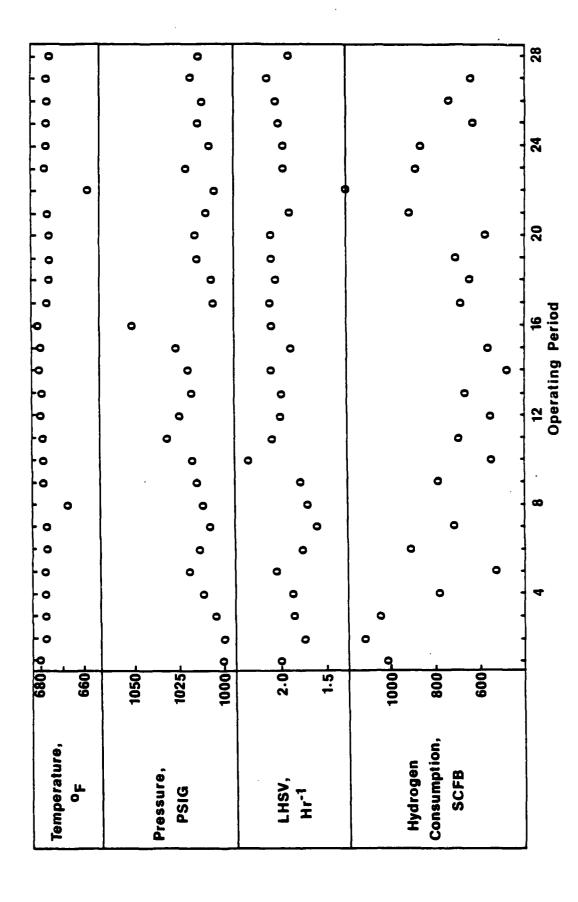
Period	13	14	15	16	17	18	19	20	21	22	23	24
		0	OPERATING	TING	CONE	CONDITIONS	<u>S1</u>					•
Temperature, OF	878	679	879	679	9/9	675	675	675	9/9	859	677	929
Pressure. PSIG	1016	1020	1027	1052	1006	1007	1014	1016	1011	1004	1020	1011
LHSV. Hr -1	1:96	2.10	1.84	2.07	2.10	2.02	5.06	2.06	1.83	1.26	1.92	1.92
Hydrogen Rate, SCFB	3050	2899	3307	2939	2858	3017	2954	2943	3265	4773	3175	3153
			MATERIAL		BALANCE	NCE						
			Weig	Weight Percent of Feed	ent of F	pee						,
Food	100.00	100.00	100.00	100.00	100 00	100	100 00	100 00	100.00	100.00	100.00	100.00
H. d. c.	5.04	4.79	5.46	4.85	4.72	86.7	88. 7	4.86	5.39	7.88	5.24	5.20
Subtotal in	105.04	104. 79	105.46	104.85	104.72	104.98	104,88	104.86	105.39	107.88	105.24	105.20
	(0.83)	(0,83)	(0.83)		(0.82)	(0,82)	(0,83)	(0.88)	(0.86)	•	(0.83)	(6.93)
Hos	(0.48)	(0.47	(0.48)	1	(0.47)	(0.51)	(0.54)	(0.53)	(0.53)	•	(0.51)	0.50
Z.Z.	(0.25)	(0, 28)	(32	•	(0,29)	(0.18)	(0.31)	(0.36)	(0.34)	•	(0.24)	(0.26)
Hydrogen Out	3.96	4.01	*. 54		3.60	3.94	3.72	3.90	3.88	-	3.79	3.81
	,		0.07	•	-	0.07	0.12	0.18	90.0		10.0	0.20
C2	0.81	0.15	0.07		0.12	0.07	91.0	0.07	0.24	•	0.01	0.05
C3	0.05	0.11	0.04		0,11	0.05	0.10	90.0	0.26	•	0.34	0.04
20	0.83	0.11	.03	•		0.03	0.07	0.05	0.19	•	0.27	0.08
Cs	•	0.03	•	•	0.03	•	•	•	1	•	•	٠
Stabilized Liquid	98.89	101.07	>8.6	98.75	93.10	105.00	96.39	87.22	90.57	106.96	97.52	97.14
Subtotal Out. %wt	106.10	107.05	104.92	•	98.57	110.67	102.24	93.25	96.93	•	103.52	102.91
% w	101.0	102.2	99.5	,	94.1	105.4	97.5	88.9	92.0	•	98.4	97.8
ſ	27.2	27.5	27.2	27.7	28.0	27.3	27.2	27.3	27.7	28.0	27.3	27.2
Sulfur. % wt	0.19		0.19		0.21	0.15	0.14	0.16	0.16	•	0.16	0.17
	(1.22)	(1.17)	(1.21)		(1,26)	(1.20)	(1.20)	(1.28)	(1.25)	•	(1.24)	(1.23)
Basic Nitrogen, "wt	•	•	0.83	•		0.83	0.87	0.85	0.85	•	98.0	0.86
Hydrogen Consumption, SCFB	652	695	260		675	632	869	619	216		878	944

( ) - Calculated

TABLE III-34 (Cont'd)

Period	25	76	27	28	29	ဗ္ဂ
OPERATING		CONDI	CONDITIONS	<b>'</b> 6		
Temperature, OF	929	9/9	9/9	9/9		
PS	1012	1010	1018	1013		
-	2.00	2,03	2.13	1.90		
_	3062	2992	2846	3206		
		BALANCE	ICE			•
Weigh	Weight Percent of Feed	nt of Fe	ed			
Feed	100.00	100.00	100.00	100.00		
Hydrogen In	5.05	4.94	4.70	5.29		
<i> </i>	105.05	104.94	104.70	105.29		
	(0.83)	(0.83)	(0.88)	•		
120	(0,49)	(0.51)	(0.55)	•		
E I	(0,37)	(0.36)	(0.42)	•		
Hydrogen Out	4.02	3.74	3.67	•		
2	1	1	•	•		
C2	0.50	0.31	0.50	•		
CJ	0.43	.0.26	0.45	-		
24	0.77	0.13	1.07	-		
CS	0.22	8	•	•		
Stabilized Liquid	97.85	96.06	86.09	98.51		
Subtotal Out, %wt	105.48	102.20	93.68	٠		
Closure, %wt	100.4	97.4	89.5			
	27.3	27.3	27.1	27.1		,
Sulfur, % wi	0.18	0.17	0.14	0.10		
Nitrogen, %wt	(1.13)	(1.16)	(1.19)	(1.29)		
=	0.87	0.86	0.88	0.88		
Hydrogen Consumption, SCFB	625	727	622	í		

( ) -. Calculated



では、これでは、10mmのでは、10mm

Figure III-13. CSHT Results For In Situ Shale Oil

						1-1-1-	F
JÁV <sub>o</sub>	28		6 0		0	0	0
Sulfur, Wr%	0.2	•	. 0	•	0		•
	0.1	•	0				•
Nitrogen,	1.3	•	•	•	•	9	•
% is	7	•	• .			•	0
Basic Nitrogen,	6.0	•	0	•	0	•	0
<b>*</b> 1*	0	4 8	12 16 Operating Period	16 Period	20	24	28

Figure III-14. CSHT Aging Results For In Situ Shale Oil

# COMPOSITE PRODUCT INSPECTION COMPOSITE 3 PERIODS 9-11

Fraction, °F	-600	+ 600
Yield, Wt % of Feed	33.0	64.5
Yield, LV % of Feed	35.6	64.9
OAPI	35.5	24.4
Carbon, Wt %	-	•
Hydrogen, Wt %	13.0	-
Nitrogen, Wt %	***	
Basic Nitrogen, Wt %	0.975	0.821
Sulfur, Wt %	0.06	0.15
Oxygen, Wt %	•	-
Phenois, Wt %		-
Ramsbottom Carbon	0.23	0.93
Paraffins (P+N), LV %	•	•
Olefins, LV %	•	•
Naphthenes, LV %	•	-
Aromatics, LV %	•	-
Gross Heating Value, Btu/lb	•	-
Arsenic, ppb	11	5000

# COMPOSITE PRODUCT INSPECTION COMPOSITE 5 PERIODS 15-20

Fraction, °F	-600	+ 600
Yield, Wt % of Feed	32.5	64.0
Yield, LV % of Feed	35.1	64.1
OAPI	35.9	23.7
Carbon, Wt %		
Hydrogen, Wt %	13.0	
Nitrogen, Wt %	(1.34)	(1.41)
Basic Nitrogen, Wt %	0.962	0.962
Sulfur, Wt %	0.06	0.18
Oxygen, Wi %	•	-
Phenois, Wt %	•	-
Ramsbottom Carbon	•	0.99
Paraffins (P+N), LV %	•	-
Olefins, LV %	•	-
Naphthenes, LV %	•	•
Aromatics, LV %	•	-
Gross Heating Value, Btu/lb	•	•
Arsenic , ppb	10	5000

<sup>() -</sup> Calculated

# COMPOSITE PRODUCT INSPECTION COMPOSITE 7 PERIODS 25-28

Fraction, °F	-600	+ 600
Yield, Wt % of Feed	33.0	61.6
Yield, LV % of Feed	35.8	61.9
OAPI	36.5	24.2
Carbon, Wt %	<b></b>	
Hydrogen, Wt %	12.9	
Nitrogen, Wt %	40 40 4 <sub>0</sub> 40	
Basic Nitrogen, Wt %	0.821	0.828
Sulfur, Wt %	0.07	0.16
Oxygen, Wi %	•	-
Phenois, Wt %	•	-
Ramsbottom Carbon	0.24	0.96
Paraffins (P+N), LV %		
Olefins, LV %		
Naphthenes, LV %		
Aromatics, LV %	****	-
Gross Heating Value, Btu/lb	3000	
Arsenic, ppb	10	2000

TABLE III-38

# CATALYST ANALYSIS - IN SITU SHALE OIL HYDROTREATER CATALYST AGING 10/5/79 - 11/2/79

Pre-Heat Section of Tab Alumina	8 C	% Fe	1 As
Upper Bed	3.26	0.16	0.06
Middle Bed	2.22	0.16	0.02
Lower Bed	1.31	0.05	0.04
Catalyst Bed			•
Top Section	21.7	0.64	0.25
Middle Section	21.2	0.33	0.11
Bottom Section	23.7	0.21	0.06

# DISCUSSION

This run successfully demonstrated the potential for a crude shale hydrotreater to be operated at relatively modest pressure while maintaining reasonable operating results. Of particular importance, maintenance of hydrogenation and stable denitrogenation activity was demonstrated. Actual desulfurization attained was somewhat disappointing and may be due to the product work-up procedure used.

### SECTION IV

### FLUID CATALYTIC CRACKING OF IN SITU SHALE OIL

A major key of the EXTRACTACRACKING process is fluid catalytic cracking (FCC) of high nitrogen feedstocks.

Objectives for this module are to convert an economic maximum of gas oil and heavier material into jet fuel and gasoline boiling range products. Secondary objectives for the FCC module are reduction of heteroatom (sulfur, nitrogen, oxygen) content without hydrotreating, and selectively redistributing the available hydrogen already in the heavy shale fractions so that the distillate product exhibits a hydrogen content greater than the FCC feed, again without external addition of hydrogen.

Due to the short experimental time available, which required a priori definition of experimental conditions with little opportunity for feedback from other process modules, a wide range of conditions and experimentation was provided. Fluid catalytic cracking studies were provided in the following manner:

- Microactivity tests provide yield-response curves as a function of feedstock, hydrotreater severity and FCC severity.
- FCC M-series tests provide single condition
   source specific data, including detailed product

properties and material for further processing, as a function of hydrotreater catalyst and fractionator endpoint.

- 3. FCR Oxy tests provide single condition source specific data, including detailed product properties and material for further processing as a function of hydrotreater catalyst age. Evaluate charging the entire crude shale hydrotreater to the FCC unit.
- 4. MAT recycle tests provide yield-response data for selected recycle streams.

Test sets 1-3 are on a fresh-feed basis, while test set 4 evaluates options available on an independent recycle basis. Test sets 2 and 3 provide independent comparison of continuous versus MAT (batch) data, and allow some normalization and/or confirmation of the smaller scale data.

### 1. MAT TESTS-FRESH FEED

### SUMMARY

A total of 50 catalytic cracking data points (yield structures) have been developed for hydrotreated 600°F+ in situ shale oil. Operating severities were varied, resulting in 33 to 95 weight percent conversions to <600°F material. Data were developed to demonstrate the impact of both basic nitrogen content and hydrotreater catalyst type. Cobalt and nickel molybdate catalysts provided equivalent

cracking yields, except for the higher severity region where nickel molybdate may be favored. Very surprising results were obtained when nitrogen extract was included in the FCC yields, in that the expected yield depression was not observed.

### **OBJECTIVES**

Target objectives for these studies were to develop, in a rapid manner, FCC yield response parameters for crude shale hydrotreater products as a function of hydrotreater severity, boiling fraction, nitrogen content and FCC unit severity.

### UNIT DESCRIPTION

These MAT small scale screening runs were being carried out in the 15mm I.D., 38cm length quartz reactor of a Micro Activity Test (MAT) Unit of the type used for some years by oil industry and catalyst manufacturer research laboratories. This "MAT" is not universally standardized, but one version has been recommended for acceptance as a standard by the ASTM D-32 committee on catalysts. The reactor is positioned vertically. A 2 to 8 gram charge of molecular sieve-containing (and/or amorphous) fluidized bed type cracking catalyst is held stationary in a constant temperature zone of an electrically heated furnace. Feedstock in the amount of 0.5 to 4 grams is delivered at constant rate downflow from a syringe. The feed oil is vaporized before contact with the catalyst. Vaporization is

accomplished by low voltage resistance heating of the 18 gauge stainless steel tubing of a deadman-preheater insert above the catalyst bed within the quartz reactor. Product synthetic crude liquid is condensed and where necessary weathered at the end of the on-stream period. Gas produced is collected and measured above brine solution in a calibrated Pyrex vessel.

# **EXPERIMENTAL**

For each sample subjected to MAT analysis, a range of conditions comprising varying temperature, catalyst-to-oil ratio and/or WHSV were selected based on a priori estimation of conditions required to obtain 30-90 percent conversion. Each data point was evaluated and, where necessary, repeated. Additional runs were performed as needed to define the "knee" or maximum distillate liquid yield region.

For each run, the standard catalyst (equilibrium commercial FCC sample) was regenerated in a muffle furnace for 2 hours at 1100°F. The catalyst charge, actual weight depending on the WHSV of operation, was placed in the unit and purged of air. After equilibration at temperature, feed was injected and the unit nitrogen purged to a specified total nitrogen flow.

Product liquids were collected in tared liquid traps, and analyzed with a gravitometer and by simulated distillation (ASTM D-2887). Product gas was collected via calibrated brine displacement and analyzed by gas chromatography. Coke yield was measured by carbon content of the recovered catalyst.

Yield calculations and data reported were based on the specified feed cut point, after adjusting for the amount of material lighter than the specified cut point in the feed. The material lighter than the specified cut point is assumed to be unreactive.

# FEEDSTOCK

A broad range of feedstocks was used for these tests. In particular, the evaluation sequence was to evaluate options available in the crude shale hydrotreater, including severity, catalyst and fractionation cut point. A further sample was prepared by doping with nitrogen extract to a very high basic nitrogen level, giving four increments of basic nitrogen feed.

The five digit figures refer to samples prepared during the previously reported (see Section III) parameter variation studies. Samples 11103, 11205 and 11312 were produced over cobalt molybdate catalyst, while 12205 was produced over nickel molybdate catalysts. The M-112 600°F+ hydrotreater product described in Section III, which had also been cracked on the FCR unit (Part 2 of this Section), was used to make sample DM 112 by blending this material

with extract from the nitrogen extraction section. This sample was processed to evaluate the feasibility of utilizing the by-product extract by blending it to the FCC unit. Since the M-112 sample was prepared using cobalt molybdate catalyst, this also provides an upper bound for the effect of nitrogen on FCC operations when charging cobalt molybdate hydrotreater products. Pertinent properties for each of these feedstocks are shown in Table IV-1.

### RESULTS

Detailed results and graphical presentation of these data are shown in Tables IV-2 through IV-6 and Figures IV-1 through IV-12. For definition of terms, the reader should consult the list of abbreviations and terms.

### DISCUSSION

A very large quantity of data was collected in these studies. Conversions ranging from 33.8 to 95.6 weight percent, based on a 600°F reference, were obtained for feed-stocks ranging from 0.1 to 1.02 weight percent basic nitrogen. For modeling purposes, yield structures were determined over a range of conversions which included definition of the maximum distillate yield region.

A datum of particular interest is the dependence of conversion and yields on basic nitrogen content of comparable feedstocks. Selected data are plotted for the three cobalt molybdate based stocks in Figure IV-11. Even though a shortage of data exists in the moderate nitrogen region

TABLE IV-1

MICROACTIVITY TESTS FRESH FEEDSTOCK

Sample	11103	11205	11312	DM112	12205
OAPI	21.3	22.8	28.9	23.3	23.5
Carbon	ı	87.3	•	ı	ı
Hydrogen	ı	11.73	1	ı	1
Nitrogen	(1.42)	(1.29)	(0.213)	(1.60)	(1.16)
Basic Nitrogen	0.95	0.86	0.10	1.02	0.85
Sulfur	0.31	0.13	0.05	•	0.08
Simulated Distillation	_				
189	529	. 715	969	377	559
10	599	582	642	603	809
20	751	738	756	745	745
96	915	906	902	806	897
EP	982	978	963	686	957

All units are weight percent unless specified

TABLE IV-2

# MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY

RUN NO.	20	22	25	28	35	\$\$	52
CONDITIONS TEMP., "P CAT./01L	1000	1000	1050	950	1000	3.1	3.0
TEST RESULTS RECOVERY, HTX	9.7	95.5	92.6	99.7	96.8	93.1	100.4
CONV. OF 600 F+ TOTAL CONV. WIX	9.68	78.5	86.0	8.69	57.5	72.4	46.6
to C <sub>5</sub> -400°F, WIX	27.8	31.3	13.1	27.0	20.7	26.9	17.7
VOLX SELECTIVITY	0.37	0.47	37.6 0.43	97.7 0.46	0.41	32.5 0.44	0.43
to 400-600°E, WTX	5.6	13.6	80 (	13.2	8.41	16.9	15.0
VOLX SELECTIVITY	0.01	14.9 0.19	9.6	0.20	16.3	0.25	16.5 0.33
to CA MINUS GAS, WIX	37.9	22.7	32.9	20.8	18.2	23.9	10.9
*(GAS, WTZ FEED) *(GAS, WTZ RECOVERY)	37.0	26.5 27.8	32.0 34.6	24.7 24.7	20.0 20.6	23.6 25.6	1.3
to COKE, WTZ	18.2	10.9	13.3	8.7	3.8	4.7	3.0
(WIX C ON RECOVERY)	16.3	. 6	11.9	7.8	3.4	4.2	2.7
UNCONVERTED 600°F+, WIX UNCONVERTED 600°F+, WOLX	10.4 10.8	21.5	14.0	30.2	42.5	27.6	53.4 55.7
EPRODUCTS, WIX RECOVERY	6.66	001	100.1	99.9	100	100	100
C <sub>5</sub> - 600°F RECOVERED, VOLX	39.9	52.8	47.2	47.2	41.3	51.0	37.9

Note: Conversions Determined on a Recovered Weight Bases  $^\star$  Total Gas, Includes  $C_5+$  Components

TABLE IV-3

The second of th

# MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY

MIN NO.	11208	•	•	2	21	*	16	8	8
CMDITIONS TEMP., *F CAT./vil.	900 3.0 15.8		1000 3.0 15.5	1050 2.9 15.9	950 6.1	1000	1050 6.2 7.7	-	_
TEST RESULTS RECOVERY, WIT	101.9	102.5	97.3	2.2	<b>%</b> .0	100.7	91.8		
TOTAL CONT. WIE	33.8	46.8	55.8	78.3	67.4	91.8	<b>84</b> .0		
to C400°F, WIX	15.2	17.0	24.6	30.2	28.5	35.0	31.9		
SELECTIVITY	16.3	20.4	29.5	36.2 0.45	7. 0.4 8	42.0 0.50	38.3		
to 400-600F, WTX	6.5	9.3	13.1	9.5	12.3	10.1	5.7		
VOLT SELECTIVITY	0.17	10.1 0.19	0.23	0.13	0.19	0.13	0.07		
to C <sub>4</sub> minus gas, wtz *(gas, wtz feed) *(gas, wtz recovery)	8.8 . 13.1 12.9	16.7 18.4 16.0	14.6 18.5 19.0	31.7 33.2 35.1	18.5 21.6 22.5	26.0 31.8 31.6	33.3 35.4 38.6		
to COKE, WTE (WTE C on PEED) (WTE C on RECOVERY)	2.9.3		2.6	5.0		10.7 9.3 9.2	13.1 10.3 11.3	14.1 11.8 12.1	16.0 13.3 13.8
UNCONVERTED 600°F+, WIR	66.2	53.2 55.0	44.2	21.7	32.6	18.2	16.0 16.5		
EPRODUCTS, WIR RECOVERY	100	99.9	.00	100.1	99.9	. 100	100		
C <sub>5</sub> - 600°F RECOVERED, VOLT	25.3	30.5	42.7	46.5	47.6	53.0	44.5		

Note: Conversions Determined on a Recovered Weight Bases  $^4$  Total Gas, Includes  $C_5^+$  Components

TABLE IV-4

では、「日本のでは、」」」
「日本のでは、日本のでは、「日本のでは、「日本のでは、日本のでは、「日本のでは、「日本のでは、日本のでは、「日本のでは、日本のでは、日本のでは、日本のでは、日本のでは、日本のでは、日本のでは、日本のでは、日本

# MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY

RUN XO. FEEDSTOCK	37	9	42	<b>3</b>	\$	1.4	20	51	52
COMDITIONS TEMP *F CCT./OIL	1000 11.6 8.2	1000 5.8 8.3	1050 7.0 6.9	1050 6.9 7.0	950 6.4 7.5	950 3.6 13.1	1050 3.3 14.4	1000 3.1 15.0	1000 3.1 15.0
RECOVERY, WIX CONV. OF 600°F+ TOTAL CONV. WIX	97.9	91.8	95.1		98.2	98.0	99.7	97.5	91.4
to C <sub>5</sub> -400°F, WTX VOLX SELECTIVITY	38.2 44.1 0.46	45.9 52.9 0.52	38.8 44.8 0.47	42.3 48.8 0.51	55.5 63.9 0.68	41.3 47.7 0.63	37.1 42.8 0.55	36.5 42.0 0.60	33.3 38.4 0.56
to 400-600°F, WTA WOLA SELECTIVITY	6.8 7.1 0.07	9.5 9.9 0.11	7.3 7.6 0.08	8.8 9.2 0.10	9.8 10.2 0.11	12.7 13.3 0.18	13.4 14.1 0.18	11.6 12.1 0.17	15.0 15.7 0.23
to C4 MINUS GAS, WTZ *(GAS, WTZ FEED) *(GAS, WTZ RECOVERY)	36.9 49.2 50.3	30.2 37.8 41.2	39.9 53.4 56.1	34.1 42.8 50.2	22.0 39.7 40.5	17.8 30.7 31.3	25.4 37.6 37.7	18.3 25.9 26.6	17.0 25.0 27.3
to COKE, WTX (WTX C on FEED) (WTX C on RECOVERY)	13.7 13.2 13.5	6.6 6.0 6.5	9.3 9.2	9.6	6.9	3.1	2.0	22.9	2.3
UNCONVERTED 600°F+, WIX UNCONVERTED 600°F+, VOLX	4.4	7.8	4.9	5.1 5.0	5.9 6.9	25.0 24.9	22.1	30.7	32.2 32.0
EPRODUCTS, WTZ RECOVERY  C5 - 600°F RECOVERED, VOLZ	100	100	100	100	100.1	99.9	100	100	100

Note: Conversions Determined on a Recovered Weight Bases  $^{\dagger}$  Total Gas, Includes  $C_5^+$  Components

TABLE IV-6

# MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY

RUY KO.	3	5	92	n	11	23	7.	22	2	=	2	2	2	6
Trenstvex	K113 -	+ 000												
TO TOTAL	. 056	920	1000	1000	920	920	1000	1000	1000	1000	1050	1050	1050	1050
CAT. /01L	2.2	2.9	3.1	2.6	3.6	4.3	4.7	3.7	7.7	5.7	3.9	4.3	6.1	4.9
ASTON	21.3	16.0	15.2	16.5	11.1	11.7	10.6	13.6	7.0	9.5	12.8	12.4	7.9	7.5
TEST RESULTS RECOVERY, WIX	92.5	93.2	<b>%</b>	92.7	1.66	. 94.3	90.9	92.4	94.1	97.8	91.6	92.8	92.1	4.68
TOTAL CONF. VIT	41.9	37.5	30.8	57.5	51.7	51.7	62.3	58.6	74.9	9.02	13.9	75.3	83.3	85.2
0 C <sub>5</sub> -400°F, VII	16.1	15.9	18.2	23.6	23.3	23.3	25.1	22.1	29.5	33.0 39.5	31.6	33.8	37.3	37.2
SELECTIVITY	0.45	97.0	0.40	0.47	0.51	0.51	0.46	0.43	0.46	0.55	9. 2	0.53	0.53	0.52
to 400-600%, VIX	11.6	10.3	14.5	16.0	13.9	13.5	16.5	13.3	15.0	14.6	16.9	15.5	12.4	13.6
SELECTIVITY	0.28	0.27	0.29	0.29	0.28	0.27	0.28	0.24	0.21	0.22	0.24	0.22	0.16	0.17
to C, MINUS GAS, WIZ	10.9	9.2	12.5	15.5	11.7	11.3	14.1	14.4	23.6 26.9	15.7	19.8 21.8	19.1	24.5 28.5	23.7
*(CAS, UT. RECOVERY)	13.8	9.5	13.9	19.0	16.9	15.0	17.2	16.8	28.6	25.8	23.6	26.8	30.9	24.0
to COKE, WTA	3.3	2.1	5.7	2.5	2.8	3.7	9.9	•	9.9	7.1	5.7	6.9	9.1	10.6
(WTX C on PEED) (WTX C on RECOVERY)	3.0	1.9	9.0 8.7	2.3	7.0 7.0	ы н п 4	. 0	• •	6.2 6.2	6.5	\$.5 \$.5	 		
UNCOMPERTED 600°F+, VIX	56.1	\$2.5	49.2 50.6	42.5	48.3	46.3	37.7	41.4	25.1	29.4	26.1 26.9	24.7	16.7	14.0
	9	8	100.1	100.1	100	6.6	901	90	100	9.68	100.7	100.1	100	100.1
PRODUCTS, VIA RECOVERY	1./6	3.	•	3.3	9.	7.7		• •	21.5		₹	1:10		
C 600°F RECOVERED, VOLX														

Note: Conversions Determined on a Recovered Weight Bases  $^{\rm A}$  Iotal Gas, Includes  ${\rm C_5}^+$  Components

TABLE IV-6

MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY

RUN NO.	59 12205	09	61	62	63	<b>3</b>	63	92	11	82	62
CONDITIONS TEMP., "F CAT./OIL WHSV	975 5.9 8.2	975 5.3 9.1	1050 6.1 7.9	1050 6.9 7.0	975 \$.9 8.2	975 6.0 8.0	900 5.9 8.1	1000 5.0 10.8	1000 4.6 10.8	950 3.3 14.4	950 4.4 10.6
TEST RESULTS RECOVERY, WIT	94.6	92.1	87.0	69.5	90.4	94.5	6.96	97.6	93.2	92.9	92.4
CONV. OF 600 F+ TOTAL CONY, WIZ	75.9	9.69	84.2	89.5	75.2	0.07	50.7	66.3	66.2	41.9	49.6
to C <sub>5</sub> -400°F, WTZ VOLX SELECTIVITY	35.5 42.3 0.55	32.0 38.2 0.54	31.5 37.5 0.44	29.5 35.3 0.39	29.3 35.0 0.46	32.6 38.9 0.54	22.0	30.3 36.2 0.53	28.7 34.2 0.50	17.7 21.1 0.47	17.5 19.2 0.37
to 400-600°F, WIX VOLI SELECTIVITY	15.2 16.4 0.21	16.3 17.6 0.25	11.8 12.7 0.15	7.9 8.6 0.10	15.5 16.8 0.22	14.6 15.8 0.22	11.0 11.9 0.22	12.7	15.4 16.7 0.25	11.4	13.4 14.5 0.28
to C, Minus Gas, utz "(Gas, utz Peed) "(Gas, utz recovery)	18.1 23.3 24.7	15.9 18.2 19.8	33.7 33.7 38.7	43.4 40.5 45.2	23.2 24.3 26.9	23.9 25.3	12.8 16.4 16.9	16.7 23.0 23.5	15.4 19.5 20.9	9.0	13.7 13.4 14.5
to COKE, WIZ (WIZ C on PEED) (WIZ C on RECOVERY)	7.2 6.2 6.6	8.4.8 8.6.8	7.3 5.8 6.7	8.6 7.1 7.9	7.1 5.9 6.5	8.8 8.0	4.4.4 0.6.2	6.7 5.9 6.1	6.5 6.2 8.2	8.6.6. 8.6.6.	5.0 4.2 4.6
UNCONVERTED 600°F+, WTZ UNCONVERTED 600°F+, WOLZ	24.8	30.4	15.8	10.5 10.8	24.8 25.5	30.0	49.3 50.7	33.7	33.8	58.1	50.4
EPRODUCTS, WIT RECOVERY	100.1	100.1	100.1	6.66	6.66	99.9	100	100.1	100.1	100	100
C <sub>5</sub> - 600°F RECOVERED, VOLX	58.7	55.8	50.2	43.9	51.8	54.7	38.2	6.64	50.9	33.4	33.7

Note: Conversions Determined on a Recovered Weight Bases  $^*$  fotal Gas, Includes  $C_5^+$  Components

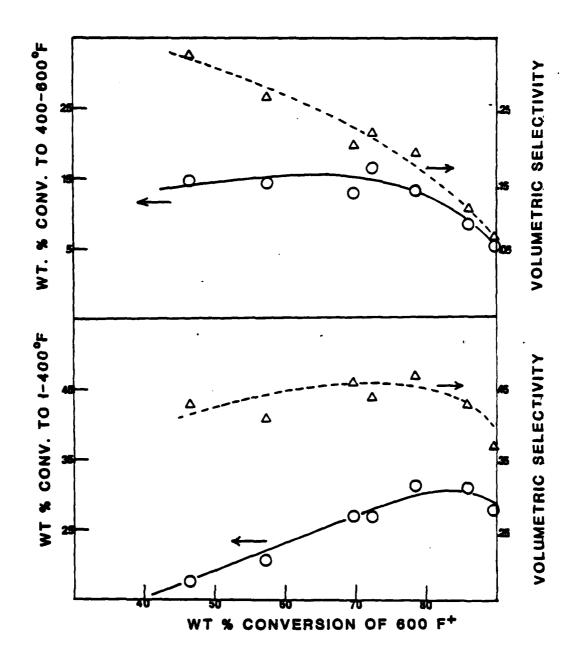


Figure IV-1. Microactivity Test Cracking
Sample 11103

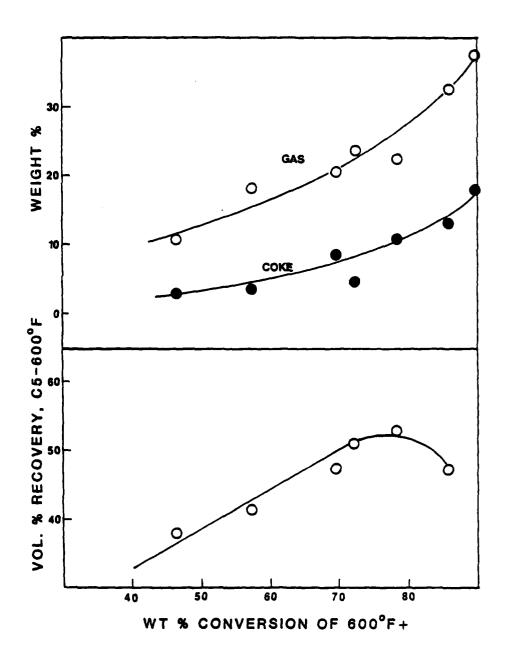


Figure IV-2. Microactivity Test Cracking Sample 11103

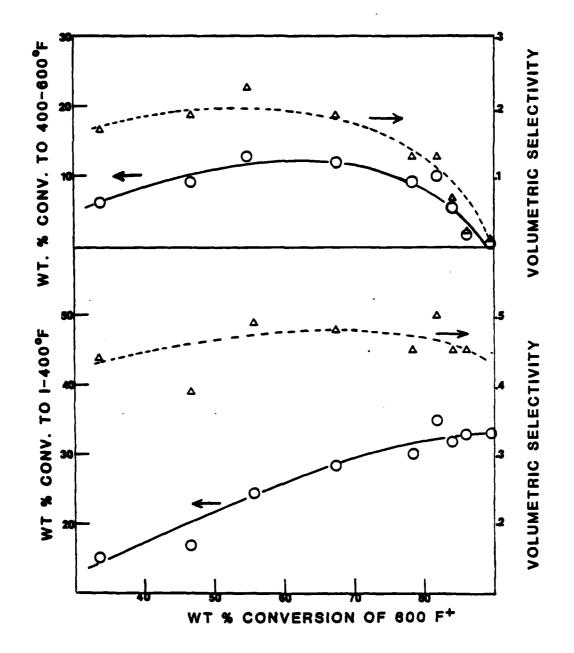


Figure IV-3. Microactivity Test Cracking Sample 11205

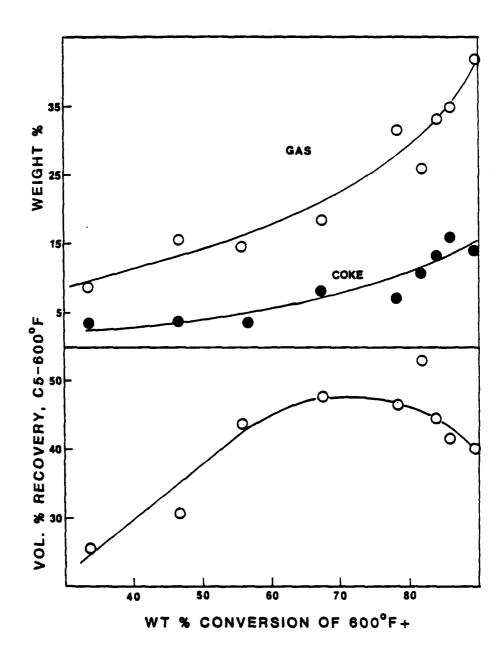


Figure IV-4. Microactivity Test Cracking
\_\_\_Sample 11205

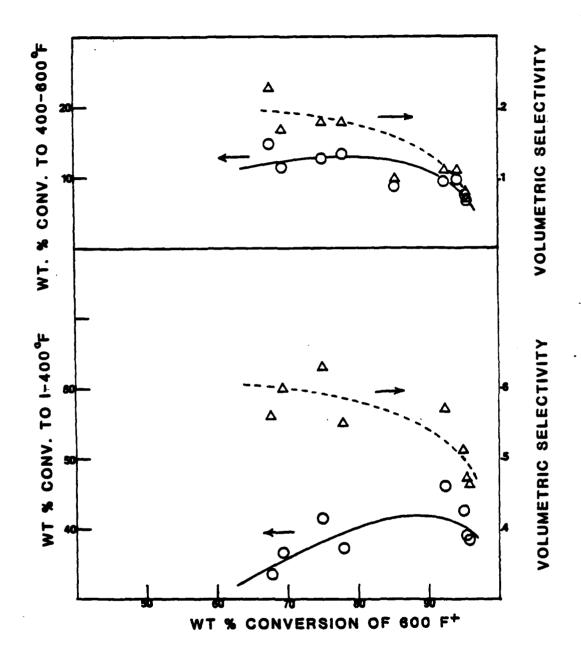
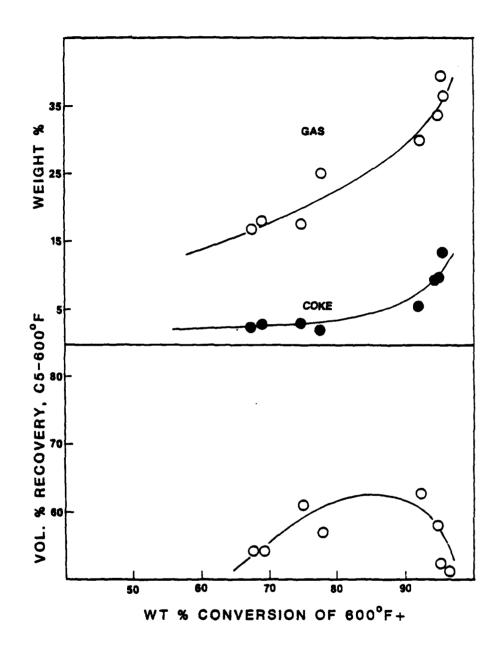


Figure IV-5. Microactivity Test Cracking Sample 11312



\_Figure IV-6. Microactivity Test Cracking \_\_Sample 11312

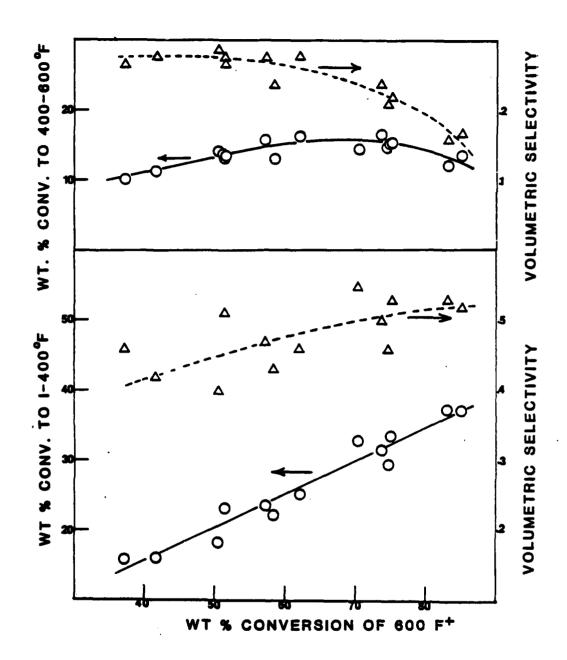


Figure IV-7. Microactivity Test Cracking
Sample M-112 600+

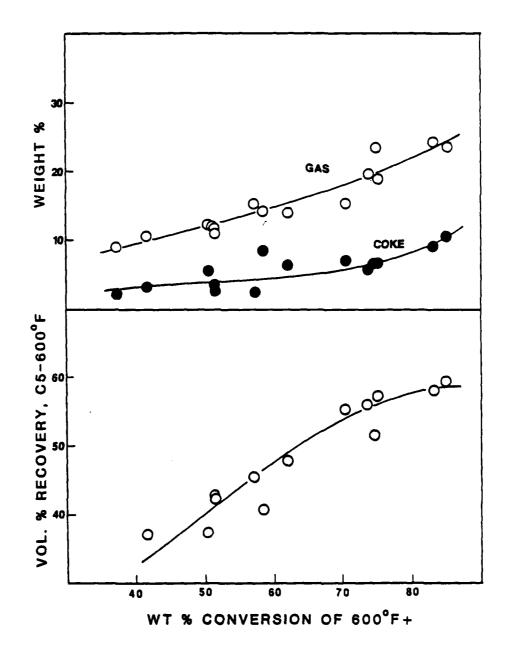
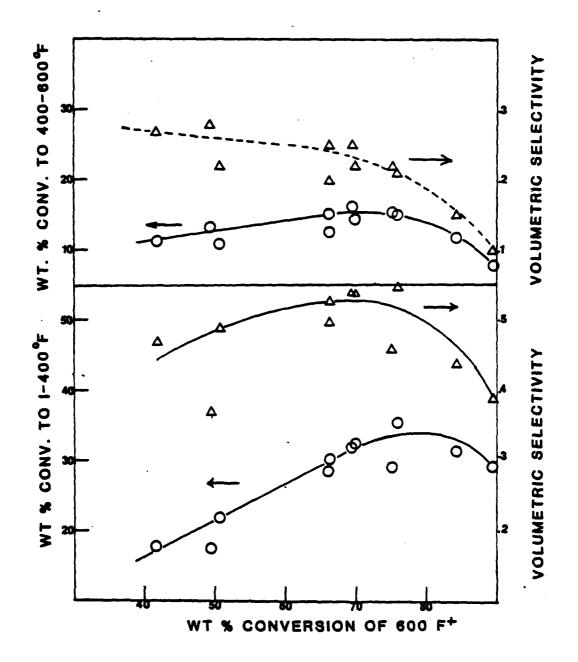
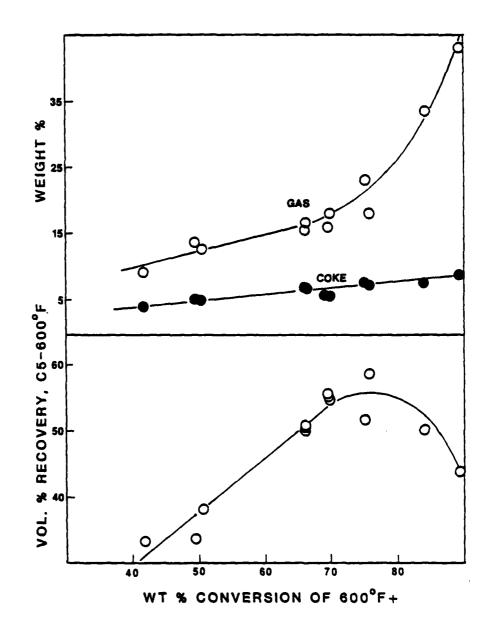


Figure IV-8. Microactivity Test Cracking
Sample M-11-2 600+



\_Figure IV-9. Microactivity Test Cracking \_Sample 12205\_\_\_\_



\_Eigure IV-10. Microactivity Test Cracking \_\_Sample 12205

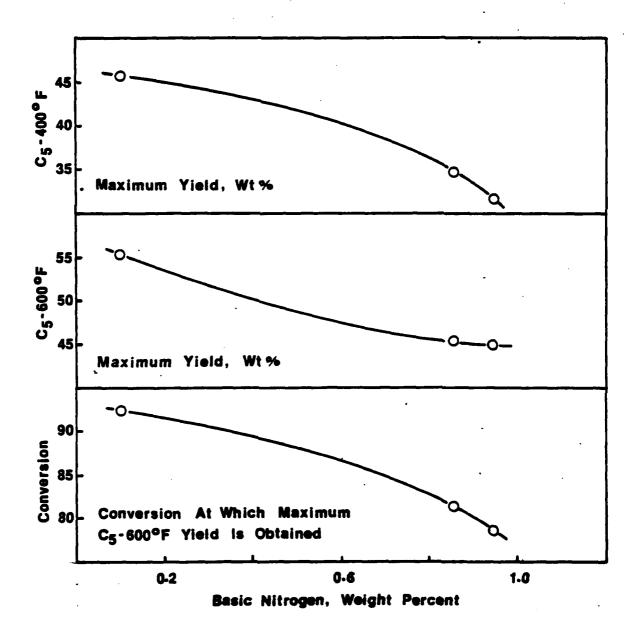


Figure IV-11. FCC Yield Maxima Dependence On Feedstock Basic Nitrogen Content For In Situ Shale Oil

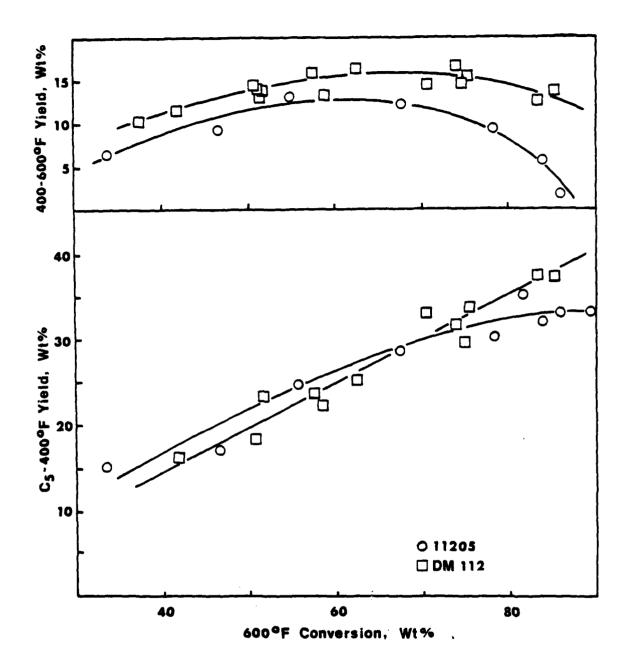


Figure IV-12. Comparison Of Nitrogen Doped
And The Analogous Fresh Feed Yields

(0.4-0.7), the correlation between yields and basic nitrogen content is very good. However, the change to different catalyst, as well as different nitrogen types (DM 112), does not follow this apparent correlation.

The data for the doped (DM 112) sample are somewhat surprising, as shown in Figure IV-12. These initial experiments were allocated to examine the potential for recovery of the nitrogen extract as a high value product by processing through the FCC unit, even though some yield depression was expected. As shown in the graphical analysis, this was not encountered, and in fact, the experimental yields obtained for the doped sample were superior to those obtained for the analogous (although not the same) fresh feed sample. These results are not immediately amenable to explanation, and further work should definitely be allocated to this aspect during Phase III. If, in fact, the same or improved yields are obtained when blending the nitrogen extract to cracker charge, the process itself will provide significantly improved economics.

Of further interest, a comparison of crude shale hydrotreater catalyst type as impacting the FCC fresh feed response shows apparently no significant difference in C5-400°F yields between the two catalysts at analogous operating conditions in the crude shale hydrotreater. Other yield structures appear very similar as well, with the exception that the nickel molybdate based material may produce slightly lower gas yields at equivalent conversion

levels, and a possible higher maximum total distillate production. The primary difference appears to be that the nickel molybdate sample maintains an increasing distillate yield to higher levels of total conversion than the cobalt molybdate sample. These differences are relatively small, and may not actually portray levels of significant difference.

### RECOMMENDATIONS AND CONCLUSIONS

- Higher hydrotreating severity provides better catalytic cracking yields and yield structure as expected.
- Conversions up to 95 weight percent (600°F basis) were obtained.
- Basic nitrogen content is a good correlative parameter for FCC yields only when the sample source parameters are constant.
- Crude shale hydrotreating over nickel molybdate catalyst may provide some yield selectivity advantage over cobalt molybdate catalyst, but only at the highest conversions evaluated.
- Feedstock doped with nitrogen extract demonstrated, on a consistent basis, equivalent or enhanced FCC performance when compared to the analogous undoped fresh feed chargestock.

### FCR PROCESSING - M-SERIES (FRESH FEED)

### SUMMARY

A circulating fluidized bed catalytic cracking unit was utilized in the pilot plant to catalytically crack, under specified conditions, nominal five gallon samples of varying quality shale oil feedstock. Full product yields and quality parameters were measured and adequate quantities of product collected for further processing. Successful processing on an integrated fluid cracking unit was demonstrated.

### **OBJECTIVES**

Objectives for these studies were to: (1) demonstrate the validity of concurrently developed microactivity test data, (2) produce adequate quantities of material for further processing and (3) determine detailed product quality characteristics for varying feedstocks.

### UNIT DESCRIPTION

The R&D fluid catalytic reactor (FCR) is a miniature version of full-scale operating FCC units. Specific details of the unit are proprietary, but in general, the unit is composed of a riser, a catalyst disengaging and stripping section, a regenerator and a regenerated catalyst standpipe. Regenerated catalyst is metered to the riser where it is admixed with oil feed by means of a slide valve. The catalyst/oil mixture flows in dilute phase to the catalyst disengaging section where the catalyst and

product vapors are separated, effectively terminating the reaction. Residual oil content is minimized in the stripping section and the stripped catalyst returned to the regenerator. Carbon content of the spent, stripped catalyst is removed by air combustion in the regenerator and the catalyst returned to the regenerated catalyst standpipe for reuse.

Catalyst inventory and circulation rates are such that the unit can be operated over varying time periods. The unit, due to its small size, is not heat balanced, being provided with external electrical resistance heaters on automatic temperature control. Product vapors are condensed and collected with product fractionation off line.

All flows are measured to provide material balance capabilities and control for the system. Regenerator flue gas composition, spent catalyst carbon content and regenerated catalyst content are monitored for material balance purposes as well.

The unit has been developed over an extensive number of petroleum feedstocks and has been demonstrated to produce results correlatable to those obtained in commercial operations.

### FEEDSTOCK

Two in situ source feedstocks were used in this study. Sample M-112 was produced in crude shale hydrotreater run M-112 over cobalt molybdate catalyst. The hydrotreater effluent was fractionated <u>+</u> 600°F and the +600°F product fed

to the FCR unit. Sample M-121 was produced in an analogous manner, with the exception of the use of nickel molybdate catalyst in the crude shale hydrotreater. Properties of each feedstock are shown in the results table for the appropriate sample.

### EXPERIMENTAL

One charge of equilibrium zeolite FCC catalyst, obtained from Ashland's petroleum refining operations, was used for both M-112 and M-121 evaluations. Two three-hour material balance periods for each sample were performed.

Operations went smoothly in both cases and good material balance closures were obtained. During each run, product gas samples, regenerator flue gas samples, spent and regenerated catalyst samples and composite liquid product samples were collected and analyzed.

The remaining composite liquid samples were collected in five-gallon cans, batch fractionated and analyzed. The distillate product portion was supplied for further processing by extraction and the heavy liquid product retained for further upgrading studies.

### RESULTS

Tables IV-7 and IV-8 summarize results for these samples. Conversions reported are based on actual +600°F feed and product yields obtained, while reported yields are as obtained without correction for -600°F material in the feed.

TABLE IV-7

### FLUID CATALYTIC CRACKING FCR PROCESSING-SAMPLE M-112

Opera	ting		Lic	luid	Proper	rties	
Condit	ions		ltem	Feed	Total Product	– 600° F	+ 600°F
Temper	ature, <sup>o</sup> f	970	OAPI	22.4	29.5	38.5	23.4
Cat/Oil	l Ratio	15.9	Carbon, %w	86.9	-	-	-
C on R	egen Cat, %	0.45	Hydrogen,%w	11.7	-	-	-
Conver	sion, Wt%	57.2	Nitrogen, %	(1.09)	(0.731)	-	-
Closure	e, Wt %	99.1	Nbasic,%w	0.816	-	_	-
	<del></del>		Sulfur, %w	0.17	0.10	-	-
Produc	:†		Oxygen, %w	•	0.176	(0.174)	0.178
Yields,			Rams C,%w	1.34	-	-	-
,			Aromatics, %w	-	-	-	-
H <sub>2</sub>	0.22		GHV, Btu/lb	18,977	-	-	-
c <sub>1</sub>	0.72		Distillation, W	1 % (°F)	by D288	7 - 73	
c <sub>2</sub>	1.96		IBP	-	148	-	-
C <sub>3</sub>	0.87		2	591	188	-	-
c 3=	2.15		10	619	285	-	-
iC4	0.53		30	689	494	· -	-
nC4	0.37		50	755	628	-	-
C4=	2.78		70	822	706	-	-
C <sub>5</sub> - 60	0 34.91		90	913	822	-	-
600+	43.1		98	963	904	-	-
Coke	12.39		EP	980	935	-	-

a - Based on the specified cut point above

( ) - Calculated

TABLE IV-8

# FLUID CATALYTIC CRACKING FCR PROCESSING-SAMPLE M-121

Opera	ting	_ <del>_</del>	Liq	uid	Proper	rties	
Condit			ltem	Feed	Total Product	– 600° F	+ 600°F
Temper	rature, <sup>o</sup> f	975	OAPI	22.9	31.7	42.5	22.5
Cat/Oi	i Ratio	14.7	Carbon, %w	87.4	-	<b>-</b> .	-
C on R	legen Cat, %	0.02	Hydrogen,%w	11.3	-	-	-
Conver	sion, Wt%	64.7	Nitrogen, %	(1.02)	(0.757)	(0.574)	(0.933)
Clasur	e, Wt %	98.0	Nbasic, %w Sulfur, %w	0.74 0.14	0.06	- :	0.12
Produ	ct		Oxygen, %w	-	0.163	(0.106)	0.218
Yields,			Rams C,%w	1.53	] -	-	0.98
,		•	Aromatics, %	-	-	-	-
H <sub>2</sub>	0.15		GHV, Btu/lb	18,938	-	-	-
c <sub>1</sub>	0.0		Distillation, W	1 % (°F)	by D288	17 - <b>73</b>	
C <sub>2</sub>	1.60		189	•	96	-	530
c <sub>3</sub>	0.64		2	-	145	-	548
c <sub>3</sub> =	2.81		10	•	238	-	603
iC4	0.73		30	•	411	-	652
nC4	0.31		50	-	605	-	702
C4=	4.01	•	70	-	690	-	766
C <sub>5</sub> - 60	39.01	•	90	-	814	-	853
600+	35.64		98	_	916	-	921
Coke	15.10		EP	•	940	_	939

a - Based on the specified cut point above

( ) - Calculated

### DISCUSSION

Mechanical limitations presented two major difficulties during these runs. Both M-112 and M-121 required relatively high catalyst-to-oil ratios due to the small quantities of feed available. These excessive ratios led to relatively high coke yields, as demonstrated by the respective 12.4 and 15.1 weight percent coke yields for M-112 and M-121. Further, air flow limitations during processing of M-112 led to a high residual carbon level on regenerated catalyst, which biases the yield distribution and exacerbates the higher coke yield problem caused by high catalyst-to-oil ratios.

In spite of these problems, relatively good conversions (57-64 weight percent, 600°F basis) were obtained for both materials. These results are even slightly higher than the original (estimated) targeted conversion. As demonstrated in Tables IV-7 and IV-8, some heteroatom removal was accomplished, as well as hydrogen redistribution, in addition to the boiling-range conversion objectives.

### RECOMMENDATIONS AND CONCLUSIONS

- Higher than desired catalyst-to-oil ratios, in both cases, and regenerated catalyst carbon level, for M-112, were encountered.
- Slightly higher conversions were attained than expected.

- Heteroatom removal and hydrogen redistribution were attained.
- Adequate products for further processing were collected.
- Detailed product qualities were developed.

### 3. FCR PROCESSING - OXY SAMPLE (FRESH FEED)

### SUMMARY

A circulating fluidized bed catalytic cracking unit was utilized in the pilot plant to catalytically crack, under specified conditions, nominal five gallon samples of various qualities of shale oil feedstock. Full product yields and quality parameters were measured and adequate quantities of product collected for further processing. Successful processing on an integrated fluid cracking unit was demonstrated.

### **OBJECTIVES**

Objectives for these studies were to: (1) demonstrate the validity of concurrently developed microactivity test data, (2) produce adequate quantities of material for further processing and (3) determine detailed product quality characteristics. These objectives are analogous to those delineated in Part 2 for the M-Series samples. Further objectives for these samples, however, were provided to: (1) evaluate the effect of feeding full-range crude shale hydrotreater product and (2) examine a series of samples

from the hydrotreater aging studies to determine the impacts of hydrotreater catalyst age on fluid cracking results.

### EXPERIMENTAL

Experimental procedures for each of these runs were analogous to those utilized during the M-Series studies (Part 2). These runs were again relatively smooth, with reasonable material balance closures obtained in each case. Material balance periods ranged 4 to 5 hours each.

Sampling was also practiced in a manner analogous to the FCR M-Series runs. Each product composite was again fractionated <u>+600°F</u>, evaluated and utilized for further processing.

### UNIT DESCRIPTION

The equipment and catalyst utilized are analogous to that described in Part 2 for the M-Series samples.

### FEEDSTOCK

In situ feedstocks for these runs were composite products from the previously described hydrotreater aging runs. Material composites from hydrotreater aging periods 1-4 (Oxy-1), 9-11 (Oxy-2), 15-20 (Oxy-3) and 25-28 (Oxy-4) were stabilized to 125°F atmospheric overhead temperature and provided to the FCR operators as a full-range, rather than fractionated, feedstock.

Properties of these feedstocks are shown in Tables IV-9 through IV-12.

TABLE IV-9

# FLUID CATALYTIC CRACKING FCR PROCESSING-SAMPLE OXY-1

Opera	ting		Lic	luid	Proper	rties	
Condi	_		ltem	Feed	Total Product	– 600 <sup>0</sup> F	+600°F
Tempe	rature, <sup>o</sup> f	995	OAPI	27.4	33.6	36.6	23.5
Cat/Oi	il Ratio	13.9	Carbon, %w	-	-	-	-
C on F	Regen Cat, %	0.02	Hydrogen,%	-	-	12.45	11.87
Conve	rsion, Wt %	44.5	Nitrogen, %	(1.15)	-	(0.73)	(0.87)
Closur	e, Wt %	96.9	Nbasic,%w	0.77	_	0.356	0.234
			Sulfur, %w	0.14	-	0.11	0.23
Produ	ct		Oxygen, %w	-	-	-	-
Yields,			Rams C,%w	-			1.70
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			Aromatics, %	-		-	44.5
H <sub>2</sub>	0.21		GHV, Btu/lb	-	-	-	-
c <sub>1</sub>	0.56		Distillation, W	/1 % (°F	) by D28	37 - 73	
c <sub>2</sub>	1,18		IBP	333	-	-	-
c <sub>3</sub>	0.53		2	373	-	-	-
c <sub>3</sub> =	1.72		10	453	<b>!</b> -		-
iC4	0.60		30	569	-	-	-
nC4	0.24		50	661	-	-	_
C4=	2.26	·	70	766	-	-	-
C <sub>5</sub> - 60	<b>90</b> 49.71		90	876	-	-	-
600+	31.64		98	943	-	-	-
Coke	11.35		EP	959	-	_	

a - Based on the specified cut point above

( ) Calculated

TABLE IV-10

### FLUID CATALYTIC CRACKING FCR PROCESSING-SAMPLE OXY-2

Opera	ting		Lic	quid	Prope	rties	
Condit	•		Item	feed	Total Product	– 600° F	+ 600°F
Temper	ature, <sup>o</sup> F	1005	OAPI	26.8		35.7	23.3
Cat/Oi	l Ratio	6.54	Carbon, %w	-	-	-	-
C on R	legen Cat, %	0.07	Hydrogen,%w	-	-	11.83	11.62
Conver	sion, Wt%	47.5	Nitrogen, %	-	-	ļ <u>-</u>	_
Closur	e, Wt %	95.3	Nbasic,%w	(0.873)	-	0.387	0.258
	<del></del>		Sulfur, %w	(0.12)	-	0.07	0.10
Produ	ct		Oxygen, %w	-	-	•	-
Yields,	Wt %		Rams C,%w	(0.69)	-	-	0.93
			Aromatics, %	•	-	-	47.4
H <sub>2</sub>	0.14		GHV, Btu/lb	-	-	-	-
c <sub>1</sub>	0.51		Distillation, W	/t % (°F)	by D28	87-73	ı
c <sub>2</sub>	1.13		IBP	328	-	-	-
c <sub>3</sub>	0.41		2	369	-	-	-
c 3=	1.72		10	454	-	-	-
iC4	0.49		30	571	-	-	-
nC4	0.20		50	662	-	-	-
C4=	2.36	·	70	768	-	-	-
C <sub>5</sub> - 60	00 49.27		90	883	-	-	-
600 <del>+</del>			98	955	-	_	-
Coke	12.56		EP	986	-	-	-

a - Based on the specified cut point above

( ) Calculated

### TABLE IV-11

### FLUID CATALYTIC CRACKING FCR PROCESSING-SAMPLE OXY-3

Opera	ting		Lic	Juid	Prope	rties	
Condit	_		ltem	Feed	Total Product	- 600° F	+600°F
Temper	rature, <sup>o</sup> f	990	PAPI	26.8	33.2	36.5	24.2
Cat/Oi	i Ratio	7.76	Carbon, %w	-	-	-	-
C on R	legen Cat, %	0.10	Hydrogen,%	-	-	12.97	11.91
Conver	sion, Wt%	46.0	Nitrogen, %	-		-	-
Closur	e, Wt %	91.2	N <sub>basic</sub> ,%w Sulfur, %w	(0.962) (0.14)	-	0.441 0.08	0.288 0.10
Produ	c <b>t</b>		Oxygen, %w	-	-	-	-
Yields,			Rams C,%w	-	-	-	0.85
,			Aromatics, %	-	-	-	47.8
H <sub>2</sub>	0.18		GHV, Btu/lb	-	-	-	-
c <sub>1</sub>	0.54		Distillation, W	/t % (°F)	by D28	87 - 73	
c <sub>2</sub>	0.93		IBP	318	-	-	-
c <sub>3</sub>	0.38		2	367	-	-	
c <sub>3</sub> =	1.55		10	452	-		-
iC4	0.42		30	570	-	_	-
nC4	0.17		50	662	-	-	-
C4=	2.03		70	768	-	_	-
C <sub>5</sub> - 60	<b>10</b> 44.03		90	877	-	-	-
600+	33.78		98	945	-		-
Coke	11.25		EP	962	-		•

a - Based on the specified cut point above
 ( ) Calculated

TABLE IV-12

## FLUID CATALYTIC CRACKING FCR PROCESSING-SAMPLE OXY-4

Opera	ting		Lic	luid	Prope	rties	
Condit	_		Item	Feed	Total Product	– 600 <sup>0</sup> F	+600°F
Temper	ature, <sup>o</sup> f	1015	OAPI	27.0	34.2	40.9	24.2
Cat/Oi	l Ratio	10.65	Carbon, %w	-	-	-	-
C on R	legen Cat, %	0.03	Hydrogen,%w	-	-	12.50	11.94
Conver	sion, Wt%	50.62	Nitrogen, %	-	-	(0.57)	(0.64)
Closur	e, Wt %	101.0	Nbasic, %w Sulfur, %w	(0.826) (0.13)	-	0.29 0.069	0.25 0.097
Produ	c <b>t</b>		Oxygen, %w	-	-	-	-
Yields,			Rams C,%w	(0.71)	-	-	1.09
,	77.		Aromatics, %w	-	-	-	43.6
H <sub>2</sub>	0.23		GHV, Btu/lb	-	-	-	-
c <sub>1</sub>	0.73		Distillation, W	/t % (°F	by D28	87 - 73	
c <sub>2</sub>	0.99		IBP	304	-	-	-
c <sub>3</sub>	0.54		2	361	-	-	-
c 3=	1.80		10	453	-	-	-
iC4	0.53	•	30	571	-	-	-
nC4	0.27		50	663	-	-	-
C4=	1.96		70	768	-	-	-
C <sub>5</sub> - 60	<b>90</b> 48.08		90	885	-	-	-
600+	30.50		98	960	-	-	-
Coke	13.42		EP	992	-	-	-

a - Based on the specified cut point above

( ) Calculated

### DISCUSSION

These runs were the first in which full-range hydrotreater product was charged to the cracker in an effort to take advantage of both the heteroatom removal capability of the unit, as well as to obtain the previously encountered improvement in extractability of the distillate product. Conversions ranged from 44 to 50 weight percent (600°F basis), which were somewhat lower than had been anticipated. Analogous (though slightly less severe) operations had resulted in 57 to 64 weight percent conversions during the M-Series studies charging >600°F feedstock.

Results from these runs demonstrate that full-range crude shale oil hydrotreater products can successfully be processed through the FCC unit. Yields of liquid products, and in particular distillate liquid products, are reduced even at the lowered severity (conversion) effectively imposed in this operation. However, there are advantages still accruing to this mode of operation, in that total heteroatoms in the extraction module charge are reduced by feeding full-range material to the FCC unit, and further, the nitrogen compounds surviving and/or produced during the FCC operation are more amenable to extraction than the blended hydrotreated straight run plus catalytically cracked product produced during the M-Series runs.

### RECOMMENDATIONS AND CONCLUSIONS

- Full-range stocks are readily processed on the FCC unit.
- Products from full-range cracking have less heteroatoms (primarily nitrogen) and are more efficiently extracted than distillates which are a blend of cat cracked and straight run hydrotreated materials.
- Liquid yields, and in particular distillate yields, suffer when charging a full-range feed to the FCC unit.

### 4. MAT TESTS - RECYCLE

### SUMMARY

FCC bottoms product, hydrotreated for recycle, has been evaluated over a range of conditions on the microactivity test unit. Conversions ranging 43 to 89 weight percent (600°F basis) were obtained for this material. The product distribution is significantly improved over the original fresh feed basis material as expected, with decreased coke, gas and distillate (400-600) yields, resulting in significantly increased C5-400 and total distillate (C5-600) yields at equivalent conversion levels when compared to the analogous fresh feed material. Data from this experimentation confirm the original yield structure projections for this type of recycle material.

### **OBJECTIVES**

The purpose for this study was to determine the yield structure response of hydrotreates recycle material in a rapid manner.

### UNIT DESCRIPTION

The equipment utilized for this study was identical to that used during Part 1, fresh feed studies.

### EXPERIMENTAL

Experimental procedure was analogous to that used in Part 1.

### FEEDSTOCK

The feedstock utilized for this study was M-112 stabilized product from the cycle odd hydrotreater run at 700°F, 1000 psig, 1 LHSV, 3000 SCFB H<sub>2</sub>, (Section VII). Properties of this feedstock and source identification data are shown in Table VI-4 under period 2.

### RESULTS

Table IV-13 and Figures IV-13 and IV-14 present MAT results for the feedstocks processed.

### **DISCUSSION**

M-112 LCO sample, hydrotreated as described in Section III, was processed over a range of conditions resulting in conversions from 43 to 89 weight percent (600°F basis). As

TABLE IV-13

MILITARY JET FUEL FROM SHALE OIL PHASE II MAT SUMMARY

IN NO.	94 M112 -	<b>3</b> 23	*	6	8	16	93	93	35	8	96	97
TEMP., °P CAT./OIL KNSV	950 3.4 13.9	950 3.6 13.1	1000 2.3 22.4	1000 3.4 13.0	950 4.1 12.5	950 5.7 7.9	1000 5.9 7.6	1000 5.6 8.0	1050 5.7 8.1	1050 6.0 7.8	900 3.1 14.2	900 3.0 14.8
EST RESULTS RECOVERY, WTZ CONV. OF 600°F+ TOTAL CONV., WTZ	94.7	96.2	96.0	98.8	95.9	94.8	98.1	96.4	88.5	89.8	92.4	91.9
to C <sub>5</sub> -400°F, 47Z vol3 selectivity	. 37.2 43.4 0.63	38.1 44.6 0.64	28.4 33.2 0.58	35.3 41.2 0.59	41.5 48.5 0.65	50.6 59.1 0.70	50.9 59.5 0.68	48.0 56.1 0.65	47.4 55.4 0.62	45.4 53.0 0.58	25.2 29.4 0.57	28.3 33.1 0.57
to 400-600F, WTZ WOLZ SELECTIVITY	8.1 8.6 0.13	8.0 8.4 0.12	7.7 8.1 0.14	6.6 7.0 0.10	6.8 7.2 0.10	3.9	2.5 2.6 0.03	1.8 1.9 0.02	0.0 0.01	0.01	6.0 6.0 0.12	7.1 7.1 0.13
to C, Minus Gas, Wix *(Gas, Wix Peed) *(Gas, Wix Recovery)	14.6 20.6 21.8	14.7 19.5 20.3	12.1 15.3 16.0	20.1 26.4 26.7	17.1 18.8 19.6	20.8 29.1 30.7	25.2 32.0 32.6	26.1 31.2 32.4	31.6 31.2 35.3	35.4 36.6 37.1	9.3 11.8 12.7	12.2 14.7 16.0
to COKE, WTZ (WTZ C on PEED) (WTZ C on RYCOVERY)	2.73	2.2	2.4	3.0	. 4 . 4 4 . 4 4 . 4	5.4	7.5 6.3 6.4	0.00	8.6 6.4 7.3	9.1	3.1	3.1
NCONVERTED 600°F+, HT. NCONVERTED 600°F+, WOLZ	36.7 37.0	36.0 36.3	49.6	34.5	29.7 29.9	18.3 18.4	13.9	16.0 16.1	11.7	10.8 10.9	56.5 56.9	49.3
PRODUCTS, WIE RECOVERY	6.66	100	100.1	100	99.9	100.1	100	6.66	100.1	100	100.1	100
5 - 600°F RECOVERED, VOLT	52.0	\$3.0	41.3	48.2	55.7	63.0	62.1	58.0	29.5	52.3	35.8	9.0%

Note: Conversions Determined on a Recovered Weight Bases
\* Total Gas, Includes C<sub>5</sub>+ Components

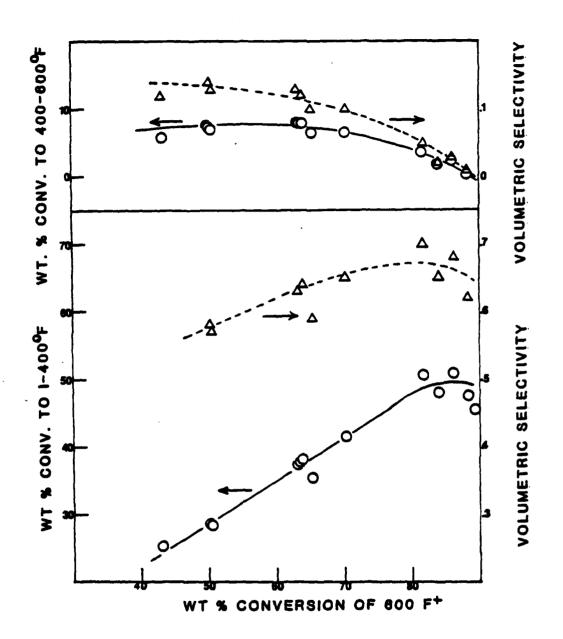
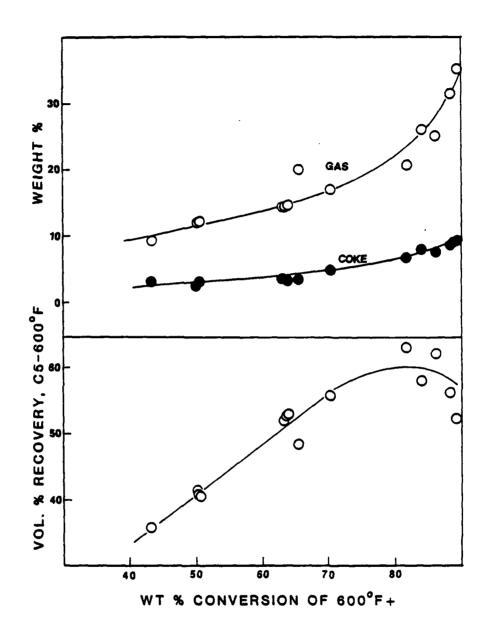


Figure IV-13. Microactivity Test Cracking
Sample M-112 LCO



\_\_Eigure IV-14. Microactivity Test Cracking \_\_Sample M-112 LCO

expected, this recycle material proved to be a superior feedstock to the original fresh feed. Gas, coke and 400-600 distillate product yields were reduced and the C5-600°F product yield increased dramatically when compared to the corresponding fresh feed sample 11205.

### RECOMMENDATIONS AND CONCLUSIONS

 The anticipated increased yield and yield distribution quality were attained from this recycle evaluation.

#### SECTION V

#### EXTRACTION

#### SUMMARY

This work was accomplished using mineral acid as the extraction solvent, and was successful in removing greater than 90 percent of the basic nitrogen content of the feedstock when performed in a counter-current extraction column.

A screening study (parameter variation) was set up to determine the effects of acid strength, ratio of hydrocarbon to mineral acid and number of stages. This was done batch-wise with in situ retorted shale oil, previously hydrotreated at the crude shale hydrotreater conditions. From this study, the recommended conditions for the remainder of the Phase II extraction work were determined.

The M-Series samples were processed on a continuous countercurrent column and good removal of basic nitrogen was accomplished. Also, there were no detrimental effects of the acid such as gum or sludge formation or polymerization of olefins present in the FCC product. The product raffinates were washed with water.

Large volume samples were then processed for the in situ retorted shale source. The samples were subjected to the same continuous extraction technique as were the M-Series samples. The raffinates were washed in a manner

which effectively removed traces of mineral acid contaminants which would have adverse effects on catalyst activity of the succeeding processing step.

#### OBJECTIVE

The purpose of nitrogen extraction is to remove a large portion of the basic nitrogen from the naphtha and mid-distillate range straight run and fluid cat cracked shale oil without the use of hydrogen.

#### MECHANICAL DESCRIPTION

In order to determine the optimal removal of basic nitrogen compounds from the shale oil, a screening study was initiated, varying the acid strength, ratio of hydrocarbon to acid and number of stages involved. This was done in glass separatory funnels. After the conditions were proven to provide the desired results, a continuous countercurrent extraction column was set up for the remainder of the Phase II sample preparations. Residence time and ratio of hydrocarbon to acid were controlled by constant volume bellows feed pumps. Acid strength was controlled by premixing the acid solution feedstock. The column interface level was controlled by the addition of an external standpipe.

After extraction, the raffinate was washed in batch glassware.

#### CHEMICAL DESCRIPTION

The extraction of nitrogen compounds from the shale oil is effected by an aqueous solution of mineral acid. The concentration of the solution was determined in the screening study to be about 40-50 percent for optimum extraction. This is described in the experimental procedure following.

In order to remove residual acid from the hydrocarbon raffinate after extraction, a dilute aqueous solution of sodium hydroxide was also used. This was followed by water washing to remove residual contaminants.

No catalyst was required.

#### EXPERIMENTAL PROCEDURE

#### 1. Screening Study

Removal of basic nitrogen compounds from shale oil was accomplished by mineral acid extraction. In order to determine the acid strength and ratio of hydrocarbon to acid which would give optimal removal of basic nitrogen, a series of batch extractions in laboratory glassware was set up. The number of stages was varied also and the removal of basic nitrogen was determined for all points of the matrix.

Independent variables were as follows:

Acid strength was set at four concentrations, 21.55, 41.9, 63.5 and 85 weight percent mineral acid in water.

The ratio of acid (100 percent basis) to hydrocarbon was determined on a weight basis as 0.05, 0.10 and 0.15 per stage, which gave a range of 0.05 to 0.45 as the number of stages varied.

The number of extraction stages was varied from one to three.

The dependent variables analyzed were the removal of nitrogen, the removal of total hydrocarbon, and subjective observations such as "rag" layer or sludge formation, emulsion formation and phase separation. Also recorded was the temperature change due to mixing.

The availability of feedstock limited the set to be done on 400-600°F boiling range material with a check point on IBP-400°F. The feedstock had been previously hydrotreated at conditions anticipated for the crude shale oil hydrotreater. No FCR product was available at the time. The unavailability of cracked feedstock would, therefore, leave unanswered any questions regarding polymerization or hydration of olefins during extraction.

After optimum conditions were determined for the extractions, a continuous countercurrent column was set up for the remainder of the Phase II extraction work.

#### 2. Continuous Extraction of M-Series Samples

The continuous countercurrent extraction column was set up using constant volume bellows feed pumps and an external siphon breaker interface level control. Acid, at

the strength and dosage found from the screening study to be an efficient extraction medium, was pumped to the top feed port of the column while the shale oil to be contacted was pumped into the lower feed port. The feed material was a material balanced blend of IBP-cutpoint from the CSHT and FCR units. The cutpoint was defined as 500°F or 600°F depending upon the final product desired. For two of the samples the straight run material and cracked material were not blended, but contacted separately to determine the effect of the extraction medium on the Bromine Number of the shale oil. A reduction in the Bromine Number, if encountered, would indicate acid catalyzed polymerization of the olefins in the cracked portion of the feedstock.

The column temperature was controlled to the temperature which would aid phase separation but not evaporate or boil off any hydrocarbon. This was controlled by a heat tape wrapped around the column and fitted with a Variac voltage controller and several thermocouples and temperature readouts. After extraction was completed the product raffinate was water washed to remove residual mineral acid. Recoveries of hydrocarbon averaged 88.4 weight percent.

#### 3. Continuous Extraction of In Situ Retorted Shale Oil

Fuel samples were prepared from in situ retorted shale oil by charging the full range CSHT products to the FCC, followed by nitrogen extraction of the IBP-600°F product

fraction. The extractions were performed in the countercurrent extraction column in much the same manner as the
M-Series samples. After extraction was completed the raffinate was further washed with reagents followed by water
washing. The washing steps were incorporated to remove
traces of mineral acid which might deactivate catalysts in
further processing steps. Washing steps were done in the
countercurrent extraction column at a 1 to 1 ratio of hydrocarbon to aqueous phase. The temperature was controlled
at about 110°F to aid in phase separation.

#### EXPERIMENTAL RESULTS

#### 1. Screening Study

The preliminary screening study had the primary objective of predicting the optimum nitrogen removal in the extraction of in situ retorted shale oil with a mineral acid. The feedstock was previously hydrotreated shale oil in the 400°F-600°F boiling range which contained 0.98 weight percent basic nitrogen. The results of the screening test show that the optimum removal of nitrogen occurs at about 40-50 percent acid strength and a nominal 10:1 ratio of hydrocarbon to acid phase and about 3-4 contact stages. (See Tables V-1 and V-2.)

TABLE V-1

# EXPERIMENTAL DESIGN FOR THE HYDROTREATED IN SITU SHALE OIL DISTILLATE EXTRACTION PARAMETER STUDY

Sequence	Acid Strength	Dosage Percent		
Number	Percent (Wt.)	(Wt.)	Stages	Boiling Range
1	(1) = 21.25	(1)= 5	1,2,3	(2) = 400-600
2	(2) = 41.90	1	1,2,3	2
3	(3) = 63.50	1	1,2,3	2
4	(4) = 85.00	1	1,2,3	2
5	1	(2)=10	1,2,3	2
6	2	2	1,2,3	2
7	3	2	1,2,3	2
8	4	2	1,2,3	2
9	1	(3)=15	1,2,3	2
10	2	3	1,2,3	2
11	3	3	1,2,3	2
12	4	3	1,2,3	2
13	( )	( )	1,2,3	(1) = 1-400

Sample No. corresponds to conditions as a suffix of the hydrotreated sample and will be labeled:

#### EXTRACTION SAMPLE

ie 11104 - 1112

H.T. sequence
Run No. 1 above,
1 stage

TABLE V-2

# DATA SUMMARY - PARAMETER STUDY FOR BATCH EXTRACTION OF IN SITU HYDROTREATED SHALE OIL DISTILLATES

7. N 507D	TOTAL WIT H.C. EXTD	TOTAL ACID: H.C.	MI'L BAL 7 CLOSURE	ACID STREAGTE	STAGES
44.6	7.1	0.05	99.1	21.25	1
65.5	10.7	0.10	99.1	21.25	Ž
78.1	11.7	0.15	100.9	21.25	3
44.6	7.4	0.05	99.8	41.90	ĭ
79.0	11.8	0.10	100.0	41.90	ž
90.2	13.4	• 0.15	99.9	41.90	3
50.0	8.9	0.05	100.0	63.5	i
77.8	13.5	0.10	100.0	63.5	ž
85.0	15.6	0.15	100.0	63.5	3
43.7	7.7	0.05	99.7	85.0	ī
72.9	15.0	0.10	100.0	85.0	2
87.4	19.9	0.15	100.0	85.0	3
61.6	9.4	0.10	99.9	21.25	ī
74.5	11.3	0.20	99.9	21.25	2
79.4	11.7	0.30	99.9	21.25	3
68.5	10.4	0.10	99.9	41.90	ī
82.1	13.3	0.20	99.9	41.90	2
89.2	14.6	0.30	99.9	41.90	3
70.6	11.4	0.10	100.0	63.5	ì
91.2	14.8	0.20	100.0	63.5	2
95.4	16.5	0.30	100.0	63.5	1 2 3
76.2	12.7	0.10	100.0	85.0	1
95.9	18.5	0.20	*****	85.0	2
99.5	22.5	0.30	100.0	85.0	3
70.2	10.3	0.15	99.9	21.25	2 3 1
79.3	11.1	0.30	99.9	21.25	2
82.2	12.2	0.45	99.9	21.25	3
73.0	11.4	0.15	100.0	41.90	1
88.0	13.1	0.30	100.0	41.90	2
95.2	14.2	0.45	100.0	41.90	3
76.5	11.6	0.15	99.9	63.5	1
94.8	14.5	0.30	99.9	63.5	2
97.1	17.6	0.45	99.9	63.5	3
78.1	14.6	0.15	100.0	85.0	1
99.8	17.8	0.30	99.4	85.0	2
99.9	22.6	0.45	100.0	85.0	3

The higher concentrations of acid removed slightly more nitrogen from the oil; however, much more hydrocarbon phase was lost to the acid phase in order to do so. Dilute acid tended not to remove a sufficient amount of nitrogen regardless of the number of stages.

After the original batch extraction matrix was completed an extraction of blended IBP-600°F material from the crude shale hydrotreater and FCC units confirmed the applicability of the process. Acid strengths were varied from 21 to 63 percent and 1 through 3 stages were utilized. The nitrogen extracted followed the same trend as with the straight run material. Bromine numbers did not change significantly, indicating little or no polymerization. Again, nitrogen removal was above 90 percent and the optimum conditions were set for continuous extractions on the countercurrent column. (See Table V-3.)

Two runs were made to test the results of extraction on four stages of the continuous countercurrent column. The IBP-600°F material blended from CSHT and FCR products was used and the results are shown in Table V-4. The basic nitrogen removal was 94 percent.

#### M-Series

Continuous extraction for the M-Series samples was set up according to the same conditions developed in the preliminary screening study. Samples were available separately of straight run and cracked IBP-cutpoint. These were blended together before the extractions were carried out.

TABLE V-3
EXTRACTION OF BLENDED PRODUCT
DATA SUMMARY

Z N EXTD	TOTAL WIZ BC EXTD	STAGES	ACID STRENGTH
84.5	7.3	1 .	21.25
91.2	7.8	2	21.25
93.6	9.1	3	21.25
88.2	8.1	1	41.90
96.4	9.1	2	41.90
98.1	10.6	3	41.90
91.9	8.7	1	63.50
99.0	11.0	2	63.50
99.5	12.9	3	63.50

#### SAMPLE TREATED:

IBP-600°F FROM CSHT BLENDED WITH IBP-600°F PRODUCT FROM FCC

TABLE V-4

## CONTINUOUS COUNTERCURRENT EXTRACTION OF SHALE OIL

SAMPLE: IBP-600°F FROM CSHT BLENDED WITH
IBP-600°F PRODUCT FROM FCC

	<u>121</u>	112
ACID STRENGTH (WT. %)	41.9	41.9
ACID DOSAGE (WT. ACID: WT. HC)	1:10	1:10
NITROGEN REMOVED (WT. %)	94.6	93.6
RATIO HC/N REMOVED (WT.)	20.6	11.8
BROMINE NO. FEED	42.2	42.0
BROMINE NO. PRODUCT	45.0	39.3
ACTUAL STAGES	4	4

Results of these extractions are shown in Table V-5. Products were water washed before further processing.

#### 3. In Situ Retorted Shale Oil

The in situ retorted shale oil, after initial hydrotreatment, FCC processing and appropriate distillation and blending was processed in four separate extractions at conditions determined as optimum for basic nitrogen removal. The continuous extraction column was set up to make use of as many as seven actual stages for mixing and with two for separation of phases. Results are shown in Table V-6. Products were caustic washed with reagent followed by water washing. These steps effectively removed the residual mineral acid to <5 ppm and the other contaminants to <1 ppm in products which were further processed (Section VII).

#### DISCUSSION OF RESULTS

#### 1. Preliminary Screening Study

The screening study consisted of parameter variation runs in batch separatory funnels in order to eliablish response data for basic nitrogen extraction. The acid strengths used were 21.25, 41.90, 63.5 and 85.0 percent (weight). The nitrogen extraction generally was favored by high acid strength, however, a large acid to hydrocarbon ratio and many stages would be required, as contact efficiency dropped off above a strength of about 50 percent. A

TABLE V-5

M-SERIES CONTINUOUS EXTRACTION
RESULTS FOR IN SITU SHALE DISTILLATES

	<u>M-1</u>			112		121
	PEED	RAPY	FEED	RAFF	FEED	RAFF
API	-	•	38.0	39.6	38.5	40.0
BR NO.	55.9	60.0	42.0	39.3	42.2	45.0
S, WT. %	-	•	0.10	0.09	0.08	0.06
N(B), WT. %	0.704	0.125	0.700	0.045	0.623	0.04
n removal Z	-	82.3	•	93.6	•	92.1
HC REMOVAL %	•	•	-	7.7	•	13.6
SIM D, WT. 7	•	•	•	•	•	•
IBP	158		138		128	
2	193	,	189		164	
10	256		298		277	
50	390		478		456	
90	485		574		553	
98	573		605		575	
Ð	698		663		628	

TABLE V-6

## GONTINUOUS EXTRACTION RESULTS FOR GC-1 STUDY-IN SITU SHALE DISTILLATES

		Y-1	0			Y-3	كالمسينية	7-4
	1110	PATTINATE	III	PATTINATE	LEED	MITTIMITE	FEED	MATTERIAL
API	36.6	37.3	36.0	37.6	36.5	38.1	40.9	40.2
BR NO.	38	42	41	43	46	37	52	47
s, wr, %	0.07	0.07	0.07	0.09	0.09	0.09	0.08	0.08
B(B), WT, Z	0.354	0.025	0.353	0.016	0.415	0.010	0.295	0.01
N Removal, %	-	92.9	-	. 95.5	-	97.6	-	95.4
HC Removal,	-	9.9	-	. 5.4	•	3.2	-	0.8
SIM D, WI Z								•
IBP	221	•	223	186	218	211	114	-
2	250	•	258	231	245	242	159	•
10	329	•	307	297	297	299	229	•
50	476	•	456	455	450	449	426	•
90	581	•	574	554	562	551	545	•
96	609	•	596	576	584	577	574	•
EP	682	•	658	582	646	602	586	•

plot of weight percent acid strength versus weight percent nitrogen removed (Figure V-1) shows the greater removal of nitrogen as acid strength increases. Figure V-2, however, shows the dramatic decrease in efficiency of removal of nitrogen to total hydrocarbon removed. This indicates a preferred acid strength of about 50 percent or less.

The preliminary batch work was done on 400-600°F boiling range material with a check run using IBP-400°F naphtha. The results indicate that up to 95 percent of the basic nitrogen could be removed using an acid strength in the neighborhood of 50 percent (Figure V-1). A greater denitrogenation could be realized at higher ratios of acid to hydrocarbon, indicating again that multi-stage extraction would be beneficial.

Continuous countercurrent extractions confirmed both the nitrogen removal and the required acid strength. An acid strength of 41.9 percent weight percent was used and resulted in predictable results both in total hydrocarbon removed and in percent basic nitrogen removed. (See Figures V-3 and V-4.)

The dosage of acid to hydrocarbon required was determined by a set of three single stage extractions, using four acid strengths as seen in Figure V-1. Figure V-3 shows that at 41.9 percent acid strength, a 10 percent dosage removes 90 - 95 percent of the basic nitrogen.

The data from the screening study were used to establish the operating conditions for the continuous runs.

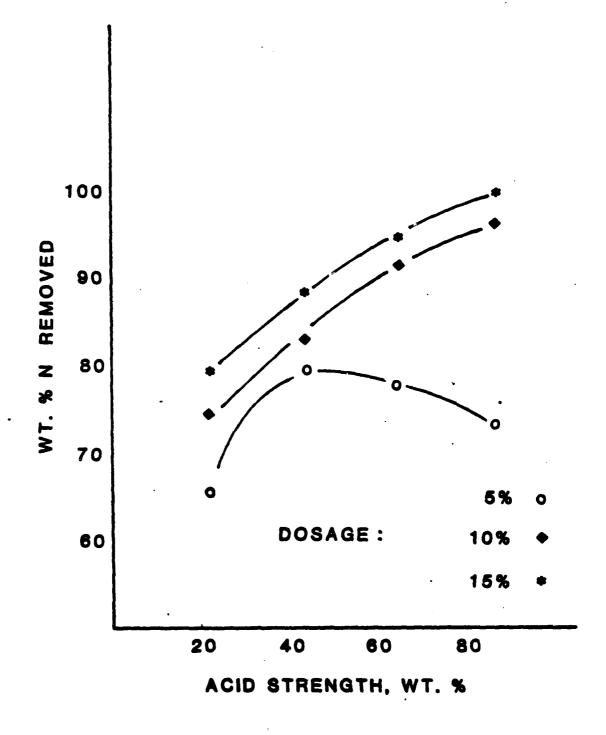
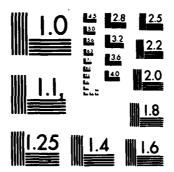


Figure V-1. Batch Extraction Of Basic Nitrogen

A0-A129 031 AÓ-A129 031 REFINING OF MILITARY JET FUELS FROM SHALE ÓIL PART II VOLUME II (IN SITU...(U) ASHLAND PETROLEUM CO KY H RMOORE ET AL. MAR 82 AFWAL-TR-81-2056-VOL-2-PT-2 F/G 21/4 33 . NL END 6 -83 DTIC



MICROCOPY RESOLUTION TEST CHART
NATIONAL BUREAU OF STANDARDS-1963-A

14

~ '

and the state of t

and the second s

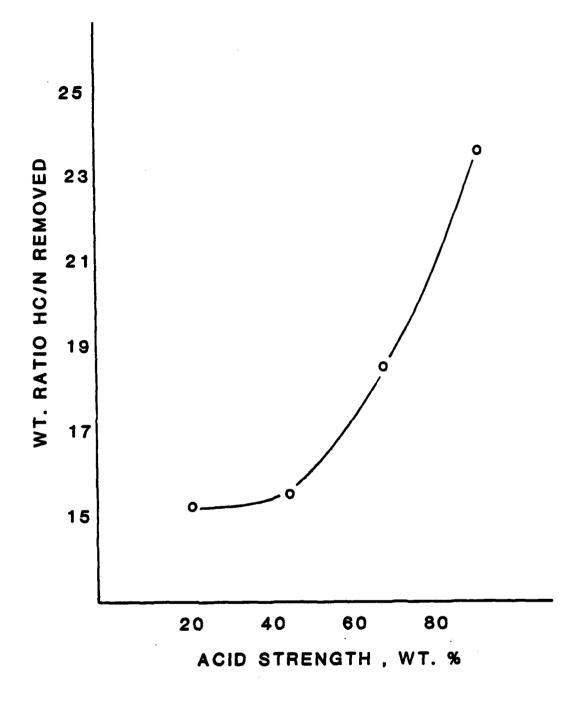


Figure V-2. Ratio Of Hydrocarbon To Nitrogen Removal In Batch Extraction

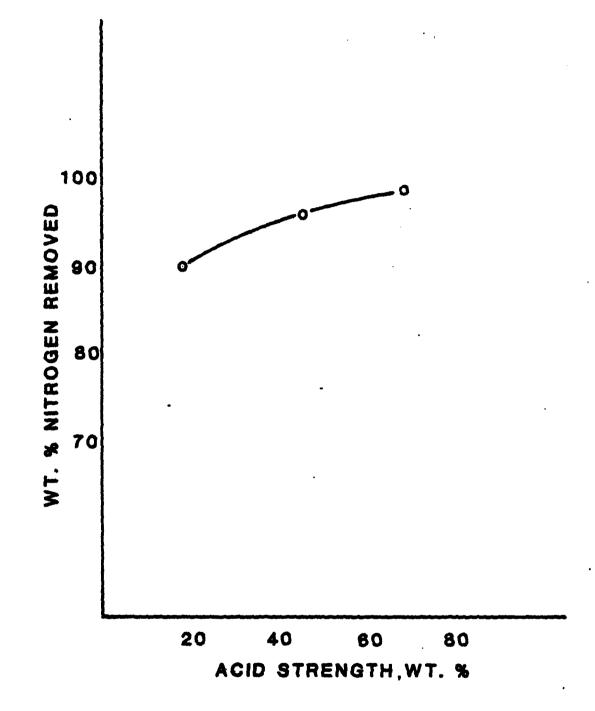
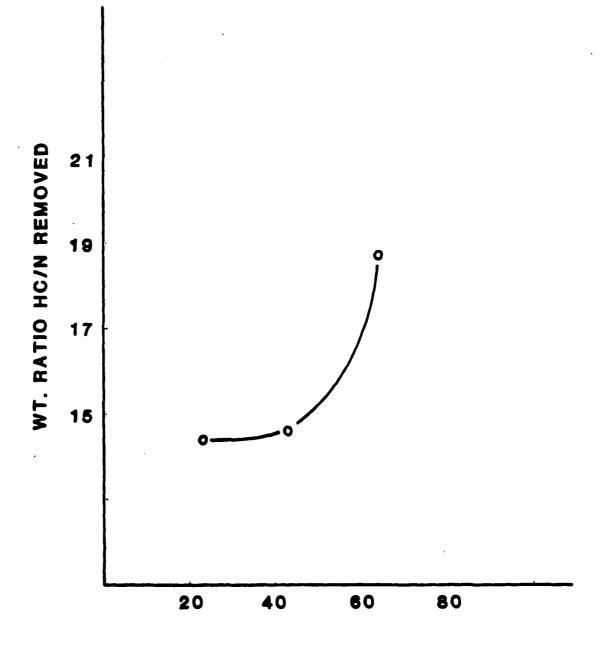


Figure V-3. Continuous Countercurrent Acid Extraction Of Basic Nitrogen



ACID STRENGTH, WT. %

Figure V-4. Ratio Of Hydrocarbon To Nitrogen Removal In Continuous Countercurrent Extraction

#### 2. M-Series

The first series of continuous extractions were conducted on samples in the process scheme. Straight run naphtha (IBP-cutpoint) and FCC naphtha were combined and extracted at conditions determined in the screening study. The continuous extraction utilized up to six actual stages of the countercurrent extraction column and, as a result, showed excellent efficiency in nitrogen removal. The last three lines of Table V-7 show the operating data for the runs and point out that the conditions were held very close to those indicated from the screening runs.

Table V-5 gives the analytical results for the runs showing basic nitrogen removal, Bromine numbers and hydrocarbon recoveries. It should be pointed out that the nitrogen-containing hydrocarbons removed with the aqueous phase are recoverable for further use.

After extraction, the raffinates were washed with water and processed further in the EXTRACTACRACKING scheme.

#### 3. In Situ Retorted Shale Oil

Four samples of in situ retorted shale oil were processed separately on the continuous extraction column. The operating conditions for these runs are shown in Table V-7. The temperature of runs three and four was elevated to effect better phase separation. Acid feed rates were adjusted to give a volumetric ratio close to 10:1 hydrocarbon

TABLE V-7

# OPERATING DATA TABLE FOR CONTINUOUS EXTRACTIONS OF IN SITU SHALE OIL DISTILLATES

RUN NO.	HC FEED	ACID FEED	HC RCVY	AQ RCVY	Volume z Closure	COLUMN TEMP	CONT. IDE. (MIN)	ACID STR	HC: ACID RATIO (WT.)
OXY-1	100	16	90.1	25.1	99.3	85	21	41.9	6.2
OXY-2	100	15.5	94.6	20.9	100.0	87	21	41.9	6.5
OXY-3	100	16.6	96.8	19.8	100.0	125	21	41.9	6.0
OXY-4	100	16.6	99.2	17.5	100.1	145	21	41.9	6.0
M-111	Aborte	d							•
M-112	100	13.0	92.3	20.8	100.1	86	21	41.9	7.7
M-121	100	12.2	86.3	25.9	100.0	86	21	41.9	8.2

<sup>\*</sup> Note: All units are wt% unless otherwise specified

to acid phase. The actual weight ratios are also reported in the operating data. Contact time was approximately 21 minutes.

Removal of basic nitrogen was in the 95 percent range as shown in Table V-6. Hydrocarbon removal was less than 10 percent, indicating optimum conditions. The conditions of acid strength and dosage as set up during the preceding work continued to prove satisfactory. No significant change occurred in Bromine numbers, indicating little or no effect upon olefin content.

The temperature of the raffinate was allowed to go up to 145° in the fourth run, which proved too high. The remaining experiments were conducted at about 87 to 90°F. For good continuous phase separation and higher throughput, it was determined that the temperature should be controlled at about 100°F or slightly above. Care was taken to keep it below the IBP of the feedstock, however.

Products from these runs were subjected to countercurrent washing with reagent and then deionized water to prevent residual contaminants from deactivating the catalyst during guardcase hydrotreating.

#### CONCLUSIONS

From the screening study it was determined that significant basic nitrogen could be extracted from the previously hydrotreated shale oil. Conditions were established for future continuous processing as follows:

Acid Dosage 1:10

Acid Strength 40-60%

Number Stages >2

The M-Series samples were successfully extracted on the continuous column utilizing only four contact stages. Predictable basic nitrogen removal of up to 95.8 percent was accomplished.

The in situ samples were also extracted with up to 99 percent removal of basic nitrogen. These samples were also caustic washed and water washed for removal of trace amounts of mineral acid and sodium.

In the cases involving cracked naphtha, no problems were encountered due to potential polymerization or gum formation.

Operations at temperatures around 100°F suggest better phase separation and no "cuff" layer formation.

Conditions recommended for Phase III Pilot Plant Preparation of samples are as outlined in Table V-8.

TABLE V-8

OPERATING PARAMETERS FOR EXTRACTION

	PHASE II	PHASE III (RECOM)
ACID EXTRACTION		
Column Diameter, in. I. D.	1.0	2.0
Hydrocarbon Feed Rate,ml/min	22.0	115.0
Acid Feed Rate, ml/min	2.2	15.0
Acid Strength, wt. %	41.9	41.9
Residence Time, min.	21	13
Column Temperature, OF	75 - 150	110 - 115
H <sub>2</sub> O Washing		•
Hydrocarbon Feed Rate, ml/min	_*	300
H <sub>2</sub> O Feed Rate, ml/min.	-	300
Column Temperature, OF	-	110

\*Batchwise

#### SECTION VI

#### RECYCLE OIL HYDROTREATING

#### SUMMARY

A total of four nominal twelve hour material balance periods were performed during the bench scale cycle oil hydrotreating of the M-Series in situ shale oil indicating that this product could be effectively hydrotreated. The data collected show that at operating conditions of 1400 psig and 700°F, desulfurization can reach at least 87 percent, while denitrogenation can reach at least 71 percent. The product samples easily met all fuel oil military specifications analyzed except pour point.

#### OBJECTIVES

The main objective of this series of cycle oil hydrotreating runs was to provide parameter response data for optimization of hydrotreating of FCC bottoms produced via the EXTRACTACRACKING process. These studies were intended to verify that the FCC bottoms could be hydrotreated to produce a lowered nitrogen, low-sulfur fuel oil, in addition to providing an upgraded recycle oil stream to the FCC unit if desired.

#### FEEDSTOCKS

The feedstocks for the cycle oil hydrotreating runs on in situ shale oil were coded M-112 and M-121. In the experimental scheme of the EXTRACTACRACKING process the M-112 feedstock had been hydrotreated at 650°F, 1000 psig,

2 LHSV and a 3000 SCFB H<sub>2</sub> rate with a cobalt molybdate catalyst, fractionated into a 600°F+ cut, sent to an FCC unit where it was cracked and finally fractionated into a 600°F+ cut. The M-121 feedstock was hydrotreated at the same conditions although with a nickel molybdate catalyst. This feedstock was then also fractionated into a 600°F+ cut, cracked in an FCC unit and once more fractionated to a 600°F+ cut. The available properties of the respective feedstocks are given in Tables VI-1 and VI-2.

Due to the small equipment sizes used, problems with catalyst fines were encountered.

#### EQUIPMENT

A nominal one-inch trickle flow laboratory reactor was used in the cycle oil hydrotreating studies. The reactor and the equipment set-up corresponded essentially to that used in the M-Series crude shale hydrotreating study and is shown schematically in Figure VI-1. The main differences between this set-up and that used in other studies were:

(1) the absence of a guardbed for catalyst protection, (2) a different method of regulating reactor pressure and (3) the absence of a mist separator in the gas flow stream after liquid collection. Temperature control for the reactor itself was provided in this instance by 3 PID controllers driven by external skin temperature thermocouples. Heat lamps and heating tapes were used to keep the feed flowing to the reactor a 'avoid mp problems.

TABLE VI-1

### FEED CHARACTERIZATION: CYCLE OIL HYDROTREATING - SAMPLE M-112

Properties:		Distillation,			
OAPI	23.4	Wt % By	D-2887		
Carbon, Wt %	-	IBP	600		
Hydrogen, Wt %	-	2	602		
Nitrogen, Wt %	-	10	620		
Basic Nitrogen, Wt %	•	30	661		
Sulfur, Wt %	-	50	711		
Oxygen, Wt %	0.178	70	773		
Ramsbottom Carbon, Wt %	-	90	856		
Aromatics, Wt %	-	98	922		
		FD	935		

TABLE VI-2

# FEED CHARACTERIZATION: GYCLE OIL HYDROTREATING - SAMPLE M421

Properties:		Distillation, °F		
OAPI	22.5	Wt % By	D-2887	
Carbon, Wt %	-	IBP	530	
Hydrogen; Wt %	_	2	548	
Nitrogen, Wt %	(0.93)	10	603	
Basic Nitrogen, Wt %	~	30	652	
Sulfur, Wt %	0.12	50	702	
Oxygen, Wt %	0.218	70	766	
Ramsbottom Carbon, Wt %	0.98	90	853	
Aromatics, Wt %	-	98	921	
	•	EP	939	

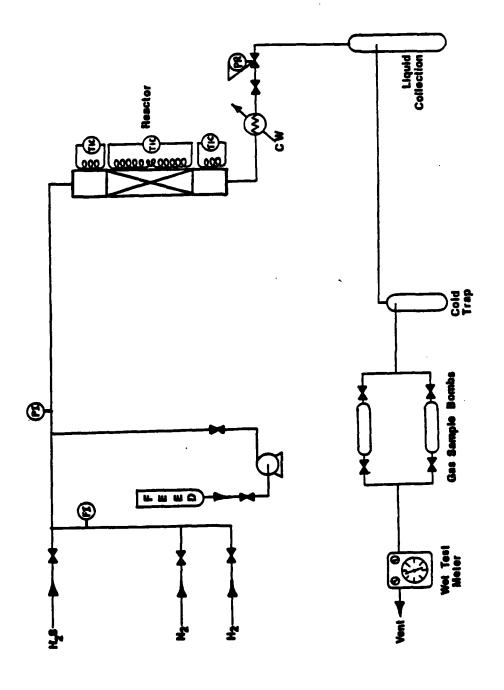


Figure VI-1. Cycle Oil Hydrotreater Configuration

#### EXPERIMENTAL PROCEDURE

The desired nominal operating conditions and scheduling of test periods for the cycle oil hydrotreating studies on an in situ shale oil were as given in Table VI-3. The M-112 cycle oil feedstock was hydrotreated at constant nominal operating conditions of 1000 psig, 1 LHSV, 3000 SCFB hydrogen rate over a catalyst charge of 50cc of a nickel molybdate catalyst and at temperatures of 650°F and 700°F before the feed was exhausted. The M-121 cycle oil feedstock was hydrotreated at 1400 psig, 1 LHSV, 3000 SCFB hydrogen rate over a catalyst charge of 50cc of a nickel molybdate catalyst and at temperatures of 650°F and 700°F before this feed was likewise exhausted.

The reactor for each feedstock was packed first with tabular alumina. Next the catalyst bed, consisting of 50cc of catalyst and 50cc of Ottawa sand, was centered in the reactor with the sand and catalyst added alternately in 10cc increments. The remaining section of the reactor was packed with tabular alumina.

Before each feedstock was fed to the reactor, the reactor was pressure checked to roughly 100 psig above the nominal operating pressure with both nitrogen and hydrogen. The catalyst was presulfided with H<sub>2</sub>S at bottle pressure up to 50°F over the maximum desired run temperature or at 800°F. After the reactor had cooled and the hydrogen flow rate had been set to 3000 SCFB, the reactor was pressurized to the appropriate run pressure. Feed was initiated and

### TABLE VI-3

## M-SERIES OPERATING CONDITIONS FOR RECYCLE HYDROTREATING OF IN SITU OIL

SAMPLE ID	M-112-CO	M-121-CO
API	23.4	22.5
SHALE OIL TYPE	In Situ	In Situ
CONSTANT OPERATING CONDITIONS:		
Pressure, PSIG	1000	1400
LHSV, Hr-1	1	1
H <sub>2</sub> Rete, SCFB	3000	3000
Catalyst	Ni/Mo	Ni/Mo
Catalyst Charge, CC	50	50
PERIOD (HRS) TEMPERATURE,	o <sub>F</sub>	
12	Lineout, 650	o <sub>F</sub>
		0

erating conditions, starting with a reaction temperature of 650°F for each feedstock. Conditions were monitored continuously and recorded hourly to maintain them at the levels delineated above. Twelve-hour lineout periods were allowed between tests as indicated in Table VI-3. Each test period consisted of a nominal twelve-hour material balance, but since the sample quantities were limited, operations were carried through exhaustion of sample rather than end of time sequence. Where end of feed occurred during a lineout, or very early in a material balance, the last full material balance was extended to end of feed.

The samples for each material balance were stabilized on an Oldershaw column. This stripping was performed for two to three hours at atmospheric pressure while maintaining a pot temperature greater than 300°F.

#### RESULTS

The operating results for the cycle oil hydrotreating studies on an in situ shale oil are presented in Table VI-4. These results include material balances based on weight percent of feed for the test lengths given. The operating conditions listed were averaged over the test period.

On the same table are listed the available product properties for the stabilized or stripped full range hydrotreated product. The results for the M-121 feedstock led to the plots of percent desulfurization and percent denitrogenation versus reaction temperature shown in Figures

### TABLE VI-4

### CYCLE OIL HYDROTREATING: IN SITU SHALE OIL

Feed M-	112	112	121	121				
Period / LENGTH.	1/12	2/,	3/12	4/18				
OPERATING CONDITIONS								
Temperature, <sup>O</sup> F	650	700	650	700				
Pressure, PSIG	1000	1000	1400	1400				
LHSV. Hr *1	1	1	1	1				
H <sub>2</sub> Rate, SCFB	3112	3020	2954	2820				
MATERIAL BALANCE Weight Percent of Feed								
Feed	100	100	100	100				
H <sub>2</sub> In	5.2	5.0	5.0	4.7				
Subtotal in	105.2	105.0	105.0	104.7				
H <sub>2</sub> O	(0.20)	•	(0, 25)	(0.25)				
NH <sub>3</sub>	(-)		(0.45)	(0.80)				
H <sub>2</sub> S	(-)	-	(0.10)	(0.11)				
H <sub>2</sub> Out	4.7	•	4,5	3.0				
C <sub>1</sub>	•	•	0.10	0.10				
C <sub>2</sub>	0.03			0.80				
C <sub>3</sub>	0.04		0.10	0.70				
C4	0.03		0.15	0.60				
C <sub>5</sub>	0.04	<u> </u>	0.04	0.50				
Stabilized Liquid	105.04	96.7	99.60	96.70				
Subtotal	-	-	105.29	103.56				
Closure	•	•	100.3	98.9				
Hydrogen Consumption, SCFB	301	-	295	1031				
F	RODUC	T PROP	ERTIES					
PAPI	26.0		26.2	27.0				
Sulfur, Wt %	0.035		0.029	0.016				
Nitrogen, Wt %	0.61	•	0.56	0.28				
Oxygen, Wt %				•				
Viscosity 100°F, cSt			20.47	16.68				
" 210°F, cSt	3.73		3.59	3.27				
Pour Point, OF	60		55	55				

VI-2 and VI-3. Simulated distillations were run on the full range product materials and are also given in Table VI-5.

#### DISCUSSION OF RESULTS

Operation of the bench scale cycle oil hydrotreater to hydrotreat M-Series in situ shale oil went very smoothly. The only problems encountered during the cycle oil hydrotreating step were insufficient quantities of feedstocks as well as insufficient analytical characterization of feedstocks and products.

Even with this shortage of data, the M-121 runs indicate that sulfur and nitrogen removals were quite good. With the reaction carried out at 650°F, 1400 psig, the level of desulfurization was about 76 percent and the level of denitrogenation was 40 percent. With the reactor at 700°F and 1400 psig pressure, desulfurization had increased to roughly 87 percent and denitrogenation had jumped to almost 71 percent.

Since part of the product stream from the cycle oil hydrotreater represents net fuel oil product, examination of the military specifications for fuel oil shows that all analyzed product streams easily surpass the maximum sulfur level of 3.5 weight percent, with the highest sulfur level in any of the product samples at 0.035 weight percent. The API gravity of all product samples is well above the minimum allowed, while the viscosity is well below the maximum

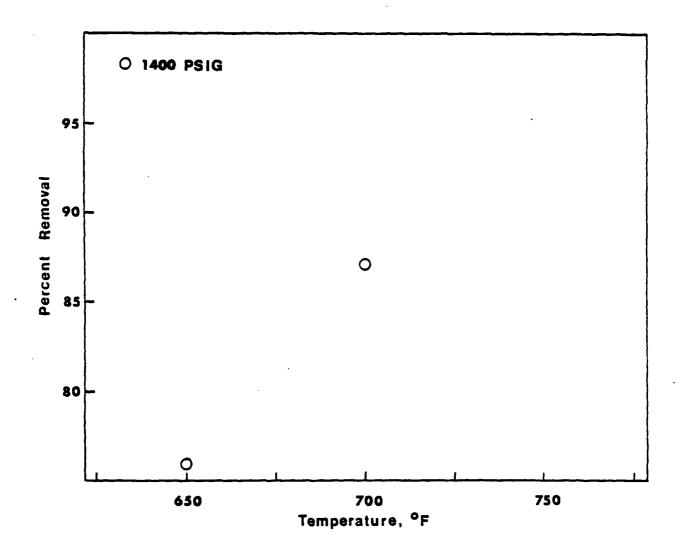


Figure VI-2. Cycle Oil Hydrotreating - In Situ Shale Oil - Temperature Dependence Of Desulfurization At 1 LHSV

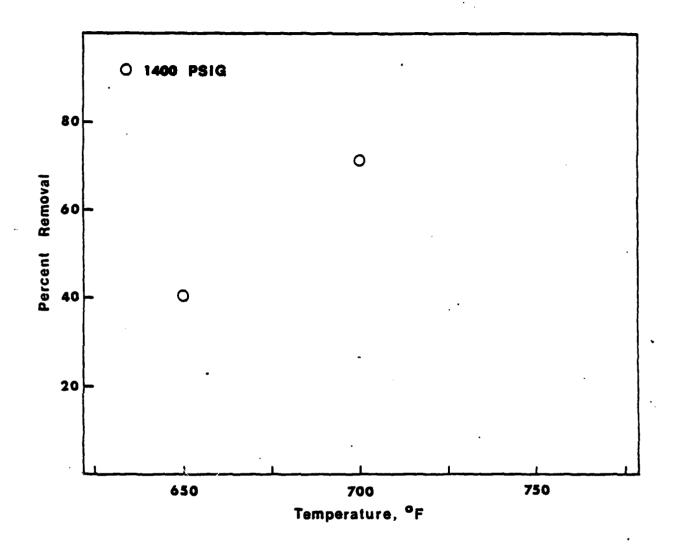


Figure VI-3. Cycle Oil Hydrotreating - In Situ Shale Oil -Temperature Dependence Of Denitrogenation At 1 LHSV

TABLE VI-5

# DISTILLATION OF FULL RANGE CYCLE OIL HYDROTREATED IN SITU SHALE OIL

SAMPLE	M-112	M-121	M-121					
OPERATING CONDITIONS								
Temperature, <sup>O</sup> F	650	650	700					
Pressure, PSIG	1000	1400	1400					
LHSV, Hr <sup>-1</sup>	1	1	1					
H <sub>2</sub> Rate, SCFB	3112	2954	2820					
D2887-73 DISTILLATION Wt% OF OF OF								
Wt %	°F	°F	° <sub>F</sub>					
IBP	431	364	391					
2	513	476	466					
10	592	579	56.5					
. 30	648	645	634					
50	703	697	689					
70	773	761	756					
90	874	854	852					
98	976	929	935					

<sup>\*</sup> Beyond temperature limit of distillation device

allowed. Carbon residue in a fuel oil stream from the cycle oil hydrotreater should also present no problem. The only property evaluated that did not meet specifications was the pour point which was 40 to 45°F over the allowed maximum. This is due to the known high paraffin content of the products. Addition of conventional pour depressants should alleviate this problem.

### CONCLUSIONS/RECOMMENDATIONS

The results of the operation of the bench scale cycle oil hydrotreater on the M-Series in situ shale oil show that this shale, processed as indicated, can be effectively hydrotreated. Data obtained indicate that even at the relatively mild operating conditions of 650°F and 1400 psig, the M-121 samples showed a desulfurization of 76 percent and a denitrogenation of 40 percent. Increasing the reaction temperature increases the levels of heteroatom removal and the maximum levels of desulfurization and denitrogenation obtained were 87 and 71 percent, respectively, at 700°F and 1400 psig.

All military specifications on fuel oil, including sulfur content, carbon residue, API gravity and viscosity, can easily be met of the M-Series in situ shale oil product from the cycle oil hydrotreater. The pour point does not meet specifications due to the high paraffin content of the product. Further processing of the product stream should not be difficult and should be considered.

### SECTION VII

### GUARDCASE HYDROTREATING

The guardcase hydrotreater is provided in the EXTRACT-ACRACKING process for final removal of trace quantities of sulfur, nitrogen, oxygen and/or metallic contaminants.

This process section provides an ultrapure feedstock for any further processing required downstream.

Two sets of studies were provided during Phase II. In the first study, two samples (from the previously described M-Series) were processed at constant (projected) required conditions. In the second, incremental severity increases were provided to define minimum required severity for a 1 ppm sulfur and nitrogen product, followed by roughly 100 hours or more at single conditions to evaluate catalyst stability.

Products from these studies were fractionated, analyze and used for further processing.

### 1. M-SERIES GUARDCASE HYDROTREATING

### SUMMARY

Raffinates of distillate in situ shale oil were hydrotreated at mild conditions to prepare samples for freeze point modification via reforming. The initial mild hydrotreating did not make specification (<1 ppm sulfur and nitrogen) reformer feed. Rehydrotreating the materials at

higher severity produced specification feed. This rehydrotreating requirement was later found to have been caused by contamination of the raffinate.

### **OBJECTIVES**

One objective of these tests was to determine stabilized activity of the catalyst for reducing sulfur, nitrogen and freeze point of pre-jet fuel streams. The second objective was to treat sufficient quantities of shale oil samples for fractionation, reforming and aromatic saturation processing.

### MECHANICAL DESCRIPTION

The equipment used as the guardcase hydrotreater is shown in Figure VII-1. The reactor is 1" I.D., made of stainless steel and 49" long. Liquid is charged to the reactor by a Lapp diaphragm pump and hydrogen flow is controlled with a Brooks electric valve. Needle valves and a rotameter are used for backup manual control of hydrogen flow. Hydrogen sulfide is piped to the unit for catalyst sulfiding and nitrogen is provided for purging and pressure testing. Four controllers adjust electrical heating of the four reactor zones.

A small heat exchanger cools reactor effluent and a Grove backpressure regulator controls system pressure while letting down product to nearly atmospheric pressure. Products are scrubbed of H<sub>2</sub>S by caustic in a downflow tube. The product stream then flows successively into a liquid

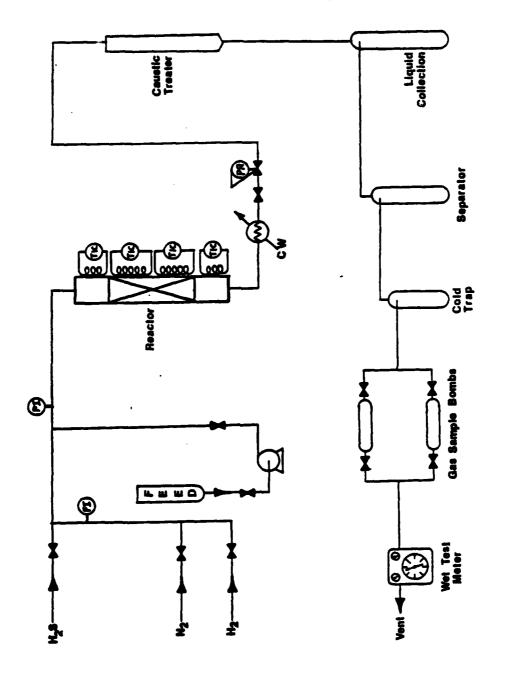


Figure VII-1. Guardcase Hydrotreater Configuration

product collector, a mist separator, a cold trap and gas sample bombs. Gases from the gas sample bombs pass through a water scrubber and a wet test meter before venting to the atmosphere.

### CHEMICAL AND CATALYST DESCRIPTIONS

Feedstocks to these runs were prepared as described in Section VI. Properties of initial and rehydrotreating feedstocks are shown in Tables VII-1 and VII-2, respectively.

TABLE VII-1

### INITIAL HYDROTREATING

Feedstock	<u> PAPI</u>	% S Weight	% N Weight
M-112 Raffinate	39.2	.12	.24
M-121 Raffinate	39.0	.07	.225

The hydrogen used is 99.95+ percent pure grade; the hydrogen sulfide is CP grade; and the sodium hydroxide pellets are electrolytic grade 97.2 percent NaOH, 0.40 percent Na<sub>2</sub>CO<sub>3</sub>. The hydrotreating catalyst is a nickel molybdate on 1/16" extrudate support.

### REHYDROTREATING

Feedstock	*API	<u> </u>	N
M-112 Remixed Product Fractions	41.5	13 ppm	117 ppm
M-121 Remixed Product Fractions	41.9	20 ppm	72 ppm

Other chemicals and catalysts were the same as the original runs.

### EXPERIMENTAL PROCEDURE

The reactors for initial guardcase hydrotreating of M-112 and M-121 samples were packed with 50cc of 1/16" Ni/ Mo catalyst and diluted with an equal volume of Ottawa sand in 10-ml increments. Postheat and preheat zones both were filled with tabular alumina.

For the rehydrotreating runs of M-112 and M-121, the reactors were packed with 250-ml of catalyst and sand each in alternate 10-ml increments. Postheat and preheat were again filled with tabular alumina.

The reactor system was purged with nitrogen and pressure tested with nitrogen and hydrogen successively to 200 psig above run pressure. The catalyst was presulfided by passing hydrogen sulfide through the reactor at 20 psig.

Bed temperature was started at 200°F and increased 50°F every half-hour until attaining 50°F greater than the highest run temperature. The unit was blocked in for two hours

at final sulfiding temperature and full H<sub>2</sub>S bottle pressure. The reactor was allowed to cool to 400°F and purged for ten minutes.

Catalyst break-in was accomplished as follows:

- 1. Begin  $H_2$  flow to cold reactor at the specified rate
- 2. Set temperature controls to 300°F
- 3. Begin oil flow to reactor at the specified rate
- 4. Hold at 300°F for one hour, then raise temperatures to 400°F
- 5. Hold at 400°F for 30 minutes, then raise to 450°F
- 6. Thereafter raise temperature to final temperature in 50°F increments every 30 minutes

The first 12 hours of operation were break-in and lineout. Break-in and lineout materials were saved. Each run continued until the feed was nearly exhausted (50-ml left), then the break-in and lineout products were added to the feed reservoir and run until this feed was also exhausted.

The target run conditions for initial hydrotreating of both samples were:

2 LHSV (100cc/hr)
3000 SCFB H<sub>2</sub> (1.88 SCFH)
675°F
1000 psig
50cc of Ni/Mo 1/16" catalyst

The target conditions for rehydrotreating were:

1 LHSV (250cc/hr)
3000 SCFB H<sub>2</sub> (4.72 SCFH)
700°F
1200 psig
250cc of Ni/Mo 1/16" catalyst

Reactor effluent was cooled and scrubbed by in-line caustic pellets to remove dissolved H<sub>2</sub>S from the liquid. The collected liquid was water-washed three times in a separatory funnel with equal volumes of deionized water to remove traces of caustic and ammonia. All liquid products were also stabilized at total reflux for two hours at 300°F pot temperature on an Oldershaw column.

### EXPERIMENTAL RESULTS

### INITIAL HYDROTREATING

The results for initial hydrotreating of M-112 and M-121 are presented in Tables VII-3 and VII-5. Table VII-3 displays the M-112 material balances and heteroatom contents. Table VII-5 displays similar data for M-121 initial hydrotreating products.

# INITIAL GUARDCASE HYDROTREATING SAMPLE M-112

Period	1	2	3	Total				
	RATI	NG C	ONDIT	TIONS				
Temperature, <sup>o</sup> F	675	675	675					
Pressure, PSIG	1015	1011	1013					
LHSV, Hr -1	1.93	1.97	1.80					
H2 Rate, SCFB	3109	3036	3389					
M	ATERI	AL B	ALAN	CE				
Weight Percent of Feed								
Feed	100.00	100.00	100.00					
Há In	5.66	5.53	6.17					
Subtotal In	105.66	105.53	106.17					
H <sub>2</sub> O	•	•	•					
NH <sub>3</sub>	0.28	0.26	0.03					
H <sub>2</sub> S	0.09	0.09	0.01					
H <sub>2</sub> Out	5.44	5.28	5.93					
C <sub>1</sub>	0.13	•	•					
C2	0.07	0.42	0.27					
C <sub>3</sub>	0.11	0 37	1.05					
C4	0.12	0.39	1.02					
C <sub>5</sub>	0.16	0.41	0.70					
Stabilized Liquid	91.80	99.43	109.59					
Subtotal	98.20	106.65	118.6					
Closure	92.9	101.1	111.7					
Hydrogen Consumption, SCFB	121	137	132					
PR	ODUC.	T PR	OPER	TIES				
OAPI .	41.9	41.6	42.0					
Sulfur, ppm	39	16	9					
Nitrogen, ppm	111	223	62					
Oxygen, pem	•	•	•					

# REHYDROTREATING OF SAMPLE M-112 THROUGH GUARDCASE HYDROTREATER

Period	1	2	3	4	Total				
OPI	ERATII	NG C	ONDIT	TIONS					
Temperature, <sup>O</sup> F	706	706	706	705					
Pressure, PSIG	1212	1212	1212	1212					
LHSV, Hr -1	.87	. 98	. 99	.65					
H2 Rate, SCFB	3434	3063	3028	4632					
M	ATER	AL B	ALAN	CE					
Weight Percent of Feed									
Feed	100.00	100,00	100.00	100.00					
H <sub>2</sub> In	6.33	5.64	5.58	8.64					
Subtotal In	106.33	105.64	105.58	108.64					
H <sub>2</sub> O		-							
NH <sub>3</sub>	0.01	0.01	0.01	0.00					
H <sub>2</sub> S	0.00	0.00	0.00	0.00					
H <sub>2</sub> Out	3.11	5.01	5.14	7.96					
C <sub>1</sub>	0.07	0.11	0.11	0.17					
C <sub>2</sub>	0.04	0.07	0.07	0.11	ļ				
C <sub>3</sub>	0.09	0.15	0.15	0.23					
C4	0.10	0.16	0.17	0.26	L				
C5	0.06	0.08	0.09	0.14	<u> </u>				
Stabilized Liquid	101.66	93.93	117.55	97.52	<u></u>				
Subtotai	105.14	99.53	123.30	106.41					
Closure	98.88	94.21	116.79	97.94					
Hydrogen Consumption, SCFB	1743	343	237	364					
PR	ODUC	T PR	OPER	TIES					
OAPI	45.5	44.4	44.0	45.0					
Sulfur, ppm	2	1	1	1					
Nitrogen, ppm	1	1	1	1					
Oxygen, Wt %									

# INITIAL GUARDCASE HYDROTREATING SAMPLE M-121

Period	1	2	3	4			
L	RATII	NG C	ONDIT	TIONS			
Temperature, <sup>0</sup> F	679	674	674	672			
Pressure, PSIG	999	1000	1000	1000			
LHSV, Hr <sup>-1</sup>	2.02	2.04	2.04	1.97			
H2 Rate, SCFB	3120	3025	3096	3046			
M	ATERI	AL B	ALAN	CE			
Weight Percent of Feed							
Feed	100.00	100.00	100 00	100.00			
H <sub>2</sub> In	5.71	5.54	5.66	5. 57			
Subtotal in	105.71	105.54	105.66	105.57			
H <sub>2</sub> O	-						
NH <sub>3</sub>	.27	. 26	.25	-			
H <sub>2</sub> S	.07	.07	.07				
H <sub>2</sub> Out	4.77	5.12	4.97	5.56			
C <sub>1</sub>	.07	0.25		0.03			
C2	.05	0.17	0.63	0.03			
C <sub>3</sub>	.12	0.44	1.28	0.06			
C4	-05	0.14	0.19	0.03			
C <sub>5</sub>	11	0.36	0.22	0.02			
Stabilized Liquid	99.83	96.85	98.10	101.27			
Subtotal	105.07	103.67	105.71	107.00			
Closure	99.4	98.2	100.05	101.4			
Hydrogen Consumption, SCFB	514	228	380	7			
PR	ODUC	T PR	OPER	TIES			
PAPI	42.1	42.6	42.1	41.9			
Sulfur, <del>ppm</del>	33	36	52	34			
Nitrogen, ppm	21	122	166	24			
Oxygen, ppm	•		•	-			

### REHYDROTREATING

The results for rehydrotreatment of M-112 and M-121 are presented in Tables VII-4 and VII-6. The M-112 retreat feed contained 13 ppm sulfur and 117 ppm nitrogen, with an API Gravity of 41.5°. The M-121 retreat feed contained 20 ppm sulfur and 72 ppm nitrogen, with an API gravity of 41.9°. Table VII-4 shows the material balances for the rehydrotreating of M-112, while Table VII-6 shows the material balances for the rehydrotreating of M-121.

The M-112 product at 44.8°API was fractionated into IBP-520°F (1565gr and 48.1°API) and 520°F+ (523gr and 42.7°API) cuts before reforming; the 520°F+ diesel stream was not further processed. The M-121 product at 45.3°API was not fractionated before reforming.

### DISCUSSION OF RESULTS

### INITIAL HYDROTREATING

The only major problem encountered during these runs was a water-soluble precipitate which formed in the unit feed buret. The precipitate was found to be unremoved extraction compounds. Results from these runs are somewhat unexpected in that: (1) high heteroatom concentrations in the product were encountered and (2) apparent extreme catalyst deactivation was observed. The deactivation was most probably due to the aforementioned sample contamination.

# REHYDROTREATING OF SAMPLE M-121 THROUGH GUARDCASE HYDROTREATER

Period	1	2	3	4	5	Total			
OPE	RATI	NG C	ONDIT	IONS					
Temperature, <sup>O</sup> F	706	703_	701	702	700				
Pressure, PSIG	1218	1218	1216	1218	1218				
LHSV, Hr <sup>-1</sup>	0.99	0.99	1.00	1.00	0,77				
H <sub>2</sub> Rate, SCFB	3036	3033	2996	2996	3107				
M	ATERI	AL B	ALAN	CE					
Weight Percent of Feed									
Feed	100.00	100.00	100.00	100.00	100.00				
H <sub>2</sub> In	5.61	5.60	5.53	5.53	5.85				
Subtotal in	105.61	105.60	105.53	105.53	105.85	6			
H <sub>2</sub> O	-	-	-	-	-				
NH <sub>3</sub>	0.01	0.01	0.01	0.01	0.00				
H <sub>2</sub> S	0.00	0.00	0.00	0.00	0.00				
H <sub>2</sub> Out	4.48	4.47	4.27	4.34	4.96				
C <sub>1</sub>	0.09	0.09	0.08	0.08	0.10				
C <sub>2</sub>	0.15	0.15	0.14	0.15	0.17				
C <sub>3</sub>	0.39	0.39	0.38	0.38	0.43				
C4	0.25	0.25	0.24	0.24	0.27				
C <sub>5</sub>	0.16	0.16	0.16	0.16	0.18				
Stabilized Liquid	97.15	94.09	96.33	102.68	115.87				
Subtotal	102.67	99.61	101.59	108.05	121.99				
Closure	97.23	94.33	96,27	102.39	115.25				
Hydrogen Consumption, SCFB	607	613	687	643	473				
PR	ODUC	T PR	OPER	TIES					
OAPI	45.2	45.2	44.8	44.6	45.9				
Sulfur, ppm	1	1	1	1	1				
Nitrogen, ppm	1	1	1	1	1				
Oxygen, ppm	•			<u>.</u>	<u> </u>				

Since the sulfur and nitrogen contents of the products were greater than 1 ppm, two further runs were made to rehydrotreat the M-112 and M-121 samples.

### REHYDROTREATING

Rehydrotreating the M-112 and M-121 samples achieved the objective of reducing the sulfur and nitrogen contents to <1 ppm.

A JP-4 fraction and a diesel fraction were made from the M-112 sample, while the M-121 sample became a JP-8 turbine fuel.

The freeze points of the JP-4 fraction of M-112 and the JP-8 fraction of M-121 both were not yet low enough to meet jet fuel specifications of -72°F for a JP-4 fuel and -58°F for a JP-8 fuel. The M-112 JP-4 fraction had a -56°F freeze point and the M-121 JP-8 fraction had a -33.7°F freeze point.

### CONCLUSIONS

- The tested M-112 and M-121 materials were slightly contaminated with unremoved extraction compounds.
- The contaminated M-112 and M-121 materials rapidly deactivated the Ni/Mo hydrotreating catalyst at mild conditions.
- The contaminated M-112 and M-121 material were successfully rehydrotreated to <1 ppm sulfur and nitrogen.</li>

- It is recommended that additional cleanup steps be provided between extraction and guardcase hydrotreating.
- The freeze points of the JP-4 and JP-8 guardcase hydrotreated fractions do not meet turbine fuel specifications.

### 2. GUARDCASE HYDROTREATING

### SUMMARY

The most important aspect of the GC-1 test was to define the required guardcase severity for in situ raffinate. Since the M-Series raffinate had been contaminated with unremoved extraction compounds the GC-series raffinates were processed under additional cleanup steps. The GC-1 tests confirmed that an uncontaminated shale oil raffinate can easily be upgraded to reformate feed specifications. The GC-1 catalyst showed no apparent deactivation.

The GC-1 products were separated into two categories:

(1) main portion "ON SPEC" material for further processing and final samples, (2) pseudocomponent fractionation stocks. Various pseudocomponents were analyzed for "API, Reid vapor pressure, simulated distillation, viscosity, FIA and freeze point. The pseudocomponent analyses allowed calculation of fractionation and blending to make JP-4, JP-5, JP-8, diesel and gasoline samples from the fully processed GC-1 sample.

### **OBJECTIVES**

There are four objectives of the GC-1 guardcase hydrotreating tests:

- Determine the least severe operating conditions required for guardcase hydrotreating to <1 ppm nitrogen and sulfur.
- 2. Operate the reactor for several days at conditions that achieve <1 ppm nitrogen and sulfur.</p>
- Treat enough shale oil product to use in later freeze point modification tests and pseudocomponent distillations.
- 4. Determine whether extracted shale oil that is not contaminated can be hydrotreated less severely to specification than the contaminated M-Series shale oils.

### MECHANICAL DESCRIPTION

Figure VII-1 shows the reactor scheme used for GC-1 hydrotreating. The scheme is the same as described for the M-Series guardcase hydrotreater.

### CHEMICAL AND CATALYST DESCRIPTION

Feedstock for this study was prepared as described in Section V for samples OXY-1, OXY-2, OXY-3 and OXY-4. These raffinates were blended, resulting in the properties shown below:

	<u> </u>	<u>N</u>	•API
Blend of Occidental Shale Oil Raffinates for #1, #2, #3 and #4	.0634%	.20%	38.2°

The hydrogen used in the guardcase hydrotreater was 99.95+ percent pure; the hydrogen sulfide was CP grade from Linde; the sodium hydroxide was electrolytic pellet 97.2 percent NaOH and 0.40 percent Na<sub>2</sub>CO<sub>2</sub>; and the catalyst was nickel molybdate on 1/16\* support.

### DESCRIPTION OF EXPERIMENTAL PROCEDURE

The GC-1 study used a reactor packed with 80cc of 1/16" Ni/Mo catalyst diluted with 160-ml of Ottawa sand in 20-ml increments. Postheat and preheat zones were filled with tabular alumina.

The units were purged with nitrogen, pressure tested first with nitrogen, then hydrogen to 200 psig greater than run pressure. Sulfiding the catalyst was accomplished by passing hydrogen sulfide through the reactor at 20 psig. The bed temperature was started at 200°F and increased 50°F every half-hour until attaining 50°F greater than the highest run temperature. The unit was blocked-in for two hours at the final sulfiding temperature and full H<sub>2</sub>S bottle pressure. After the two hours blocking, the reactor was allowed to cool to 400°F and purged with nitrogen for ten minutes.

Catalyst break-in was performed as follows:

- 1. Set H<sub>2</sub> flow to cold reactor at 1.88 SCF/Hr
- 2. Set temperature controls to 300°F
- 3. Set oil flow to reactor at 100cc/hr
- 4. Held at 300°F for one hour, then raised temperatures to 400°F
- 5. Held at 400°F for thirty minutes, then raised to 450°F
- 6. Thereafter raised temperature to 675°F in 50°F increments every thirty minutes

The first 18 hours of operation were break-in and lineout. Break-in and lineout material were saved. The run continued until the original feed nearly ran out. The break-in and some off-spec product were then recycled to the feed reservoir and run until the feed was exhausted.

The conditions for GC-1 were as follows:

Test	1	650°F		LHSV ml/hr)	800	psig	4000 SCFB H <sub>2</sub> (2.01 SCFH)
Test	2	650°F		LHSV ml/hr)	1000	psig	4000 SCFB H <sub>2</sub> (2.01 SCFH)
Test	3.	650°F		LHSV ml/hr)	1200	psig	4000 SCFB H <sub>2</sub> (2.01 SCFH)
Test	4	675°F		LHSV ml/hr)	1200	psig	4000 SCFB H <sub>2</sub> (2.01 SCFH)
Test	5	700°F	_	LHSV ml/hr)	1200	psig	4000 SCFB H <sub>2</sub> (2.01 SCFH)

Since the reactor products were scrubbed by caustic to remove dissolved  $H_2S$  from the liquid, the liquid collected was water-washed at least three times in a separatory

funnel with an equal volume of fresh deionized water. As an added precaution against leftover contamination from dissolved NH<sub>3</sub> and H<sub>2</sub>S, all liquid products were stabilized at total reflux for two hours at 300°F pot temperature on an Oldershaw fractionation column.

### EXPERIMENTAL RESULTS

The experimental results of GC-1 are presented in Table VII-7.

Four pseudocomponent distillations were performed on GC-1 materials. The distilled materials were:

- GC-1 feed
- Lineout #1 product
- Lineout (L.O.) #2 + L.O. #3 + L.O. #4 + L.O. #5 + Test
  #1 product
- Test #2 + Test #3 + Test #4 + all periods from Test #5
   ("ON SPEC")

Each of the four groups has different sulfur and nitrogen contents. The pseudocomponent boiling ranges were IBP-290°F, 290-400°F, 400-510°F, 510-570°F and 570°F+. A Todd fractionation column operated at 5:1 reflux was used to fractionate the material. Tables VII-8 through VII-11 display the pseudocomponent properties.

The main portion of "ON SPEC" product was later used for freeze point modification processing.

TABLE VII-7

# GUARDCASE HYDROTREATING OF SAMPLE GC-1

OPI 00 10 10 10 10 10 10 10 10 10 10 10 10	7	-	0	3	4	5.1	5.5	5.3	5.4
Color   Colo	Period	•	Į .				140121	ú	
Signate   651   650   649   675   700   700   700   650   650   650   610			0	PERA	SNE			O	
Signate   Signate   1217   1210   1209   1212   1201   1101   1101   1202   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0.98   0.95   0	30	159	650	649	675	200	700	700	669
MATERIAL BALANCE   1.00	Temperature, r	12	1010	1217	1210	1209	1212	1201	1185
MATERIAL BALANCE	Pressure, Polo		1 02	0.98	0.95	0.98	0.95	0.98	0.92
Weight Percent of Feed           Weight Percent of Feed           Lin         100,00	- 0	3979	3925	9807	4215	4070	4193	4086	4341
Weight Percent of Feed         Weight Percent of Feed         Weight Percent of Feed         100,00<	חפופי			MATE		BALAN	1CE	÷	
ogen In         100,00			•	Weig	ht Perc	ent of F	eed		
ogen In         107.19         7.10         7.39         7.61         7.36         7.36         7.39         7.00           olal In         107.19         107.10         107.39         107.61         107.36         107.36         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.30         107.39         107.30		100 001	100 001	100 00	100.00	100.00	100.00	100.00	100.00
ogen In         107.19         107.10         107.39         107.61         107.36         107.36         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.39         107.30         107.39         107.30         10.07         0.07		20.70	7 10	7.39	7.61	7.36	7.58	7.39	7.85
ogen Out	٩.	107 19	107.10	107.39	107.61	107.36	107.58	107.39	107.85
ogen Out         0.07         0.00	- {					,	-	•	
ogen Out         0.24         0.20         0.00	H20	0 07	0.07	0.07	0.07	0.07	0.07	0.07	0.07
ogen Out         6.48         6.17         5.68         6.77         6.27         6.31         6.45           0.00         0.00         0.16         0.41         0.00 <t< th=""><th>H2S</th><th>76.0</th><th>0.74</th><th>0.24</th><th>0.24</th><th>0.24</th><th>0.24</th><th>0.24</th><th>0.24</th></t<>	H2S	76.0	0.74	0.24	0.24	0.24	0.24	0.24	0.24
drogen Out         0.01         0.00         0.16         0.41         0.00         0.00           drogen Out         0.30         0.00         0.16         0.41         0.47         0.69           0.56         0.82         0.30         0.08         0.35         0.47         0.69           0.56         0.82         0.39         0.11         0.44         0.62         2.54           0.27         0.43         0.39         0.10         0.35         0.57         0.63           abilized Liquid         101.60         103.8         1.00         0.29         0.63         0.85         0.96           abilized Liquid         101.60         103.33         104.25         97.48         103.05         104.32         97.95         10           0 sure, %wt         101.60         103.33         104.25         97.48         103.05         104.35         104.35         104.35         102.02         10           1 plotal Out, %wt         41.5         41.9         41.7         42.2         42.9         42.8         42.6         4           April or         101         1         1         1         1         1         1           April or<	,	87.9	6.17	5.68	6.77	6.27	6.31	6.45	5.68
abilized Liquid  by the consumption, SCFB  ou solve a consumption is a consumption of the	drogen	00 0	0.00	0.0	0.16	0.41	0.00	0.00	0.00
abilized Liquid  0.27 0.43 0.39 0.11 0.44 0.62 2.54  abilized Liquid  10.96 0.58 1.00 0.29 0.63 0.85 0.96  0.96 0.58 1.00 0.29 0.63 0.85 0.96  0.96 101.60 103.33 104.25 97.48 103.05 104.32 97.95 10  osure, %wt 110.49 112.20 112.33 105.31 111.84 113.46 109.55 11  Iffur, ppm  trogen, ppm ssic Nitrogen, %wt 5.8 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	5	02.0	0.56	0.30	0.08	0.35	0.47	0.69	0.61
biotal Cut, %wt 101.60 103.33 104.25 97.48 103.05 0.57 0.63 0.96 blotal Out, %wt 110.49 112.20 112.33 105.31 111.84 113.46 109.55 11 sure, %wt 103.07 104.76 104.61 97.85 104.17 107.58 102.02 110 ltr., ppm 5.8 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	C2	92.0	0.82	0.39	0.11	0.44	0.62	2.54	2.24
bilized Liquid         0.96         0.58         1.00         0.29         0.63         0.65         0.96           biotal Dout, %wt         101.60         103.33         104.25         97.48         103.05         104.32         97.95         10           sure, %wt         110.49         112.20         112.33         105.31         111.84         113.46         109.55         1           pl         41.5         41.9         41.7         42.2         42.9         42.8         42.6         4           pl         1.8         1         1         1         1         1         1         1           rogen, ppm         5.8         1         1         1         1         1         1         1           sic Nitrogen, %wt         -         -         -         -         -         -         -         -         -           sic Nitrogen Consumption, SCFB         393         513         943         468         601         700         518	E .	0.27	0.43	0.39	0.10	0.35	0.57	0.63	0.56
ized Liquid     101.60     103.33     104.25     97.48     103.05     104.32     97.95     102.02       ital Out, %wt     110.49     112.20     112.33     105.31     111.84     113.46     109.55     113.46       ire, %wt     103.07     104.76     104.61     97.85     104.17     107.58     102.02     104.61       ir, ppm     41.5     41.9     41.7     42.2     42.9     42.8     42.6     42.6       gen, ppm     5.8     1     1     1     1     1     1       c Nitrogen, %wt     5.8     1     1     1     1     1     1       c Nitrogen, consumption, SCFB     393     513     943     468     601     700     518     119	2.0	96 0	0.58	1.00	0.29	0.63	0.85	0.96	0.84
ite, %wt     110.49     112.20     112.33     105.31     111.84     113.46     109.55     113.46       ire, %wt     41.5     41.5     41.6     104.61     97.85     104.17     107.58     102.02     104.       ir, ppm     5.8     1     1     1     1     1     1     1       gen, ppm     5.8     1     1     1     1     1     1       c Nitrogen, %wt     -     -     -     -     -     -     -     -       c Nitrogen Consumption, SCFB     393     513     943     468     601     700     518     113		101	103.33	104.25	97.48	103.05	104.32	97.95	102.88
Inch. %wt     Out. 76 Min.     103.07     104.76     104.61     97.85     104.17     107.58     102.02     104.02       Ire, %wt     %wt     41.5     41.9     41.7     42.2     42.9     42.8     42.6     42.6       gen, ppm     5.8     1     1     1     1     1     1     1     1       c. Nitrogen, %wt     -     -     -     -     -     -     -     -       ogen Consumption, SCFB     393     513     943     468     601     700     518     119		110.49	112.20	112.33	105.31	111.84	113.46	109.55	
Ire, 7eWI     John     41.5     41.9     41.7     42.2     42.9     42.8     42.6       Ir, ppm     5.8     1     1     1     1     1     1       c Nitrogen, %wt     -     -     -     -     -     -       ogen Consumption, SCFB     393     513     943     468     601     700     518		103.07	104.76	104.61	97.85		107.58	102.02	
gen, ppm c Nitrogen, %wt ogen Consumption, SCFB 393 513 943 468 601 700 518	e i	41.5	6 17	41.7	42.2		42.8	42.6	42.6
8n, %wt	Iden		• •	-	-	1	1	1	-
sn, %w/l sumption, SCFB 393 513 943 468 601 700 518	Sulfur, ppm	0.1	1-	-	-	1	-	1	-
SCFB 393 513 943 468 601 700 518		3.0					-	٠	
SCFB 393 513 943 400 001					037	109	002	518	1196
		393	513	943	400	100			

# TODD PSEUDOCOMPONENTS

# GC-1 FEED

	IBP-290°F	290-400°F	400-510°F	510-570°F	570+°F
Wt%-yield	11.05	21 75	41.66	22.95	2.58
*API	53.6	42.9	36.1	31.9	-
RVP	•	0.4 1b	0.2 1b	-	-
Vis 100°F (cst) 210°F	-	1.03 0.71	1.97 0.99	3.59 1.36	-
FREEZE ,*F	-90	<b>-90</b> .	-40	+6.8	•
FIA Saturates	-	•	-	-	•
Olefins Aromatics	•	•		-	-
SIM-0 IBP D~2887	101°F	250°F	357 <sup>0</sup> F	469 <sup>0</sup> F	395 <sup>0</sup> F
2%	133	267	376	482	536
10%	174	291	404	504	561
50%	254	347	459	547	590
90%	323	399	516	578	730
98%	358	425	543	590	861
ΕP	484	445	561	620	907

# TODD PSEUDOCOMPONENTS

# GC-1 LINEOUT 1

	1BP-290°F	290-400°F	400-510°F	510-570°F	570+°F
Wt% yield	8.86	24.62	46.01	16.57	3.93
*API	57.4	45.0	37.9	37.2	37.4
RVP	-	0.35 lb	0.25 1ъ	•	-
Vis 100 °F (cst) 210°F	-	0.94 0.55	1.93 0.91	3.64 1.35	~
FREEZE , °F	•	-90	-41.8	•	
FIA Saturates	82.2	68.5	58.3	•	•
Olefins	1.3	1.7	8.0	-	-
Aromatics	16.5	29.8	33.0		
SIM-O IBP D-2887	133 <sup>0</sup> F	229 <sup>°</sup> F	370°F	500 °F	556 °F
2%	144	252	384	505	560
10%	193	286	408	517	571
50%	249	345	458	546	591
90%	302	394	507	574	751
98%	333	418	521	584	760
EP	352	441	538	607	818

# TODD PSEUDOCOMPONENTS

# GC-1 LINEOUT 2,3,4,5,& TEST 1

	18P-290°F	290-400°F	400-510°F	510-570°F	570+°F
Wt% yield	15.65	14.44	44.02	23.84	2.06
*API	55.6	43.8	38.9	38.4	35.8
Ryp	1.75 lb	0.05 lb	0.05 1b	-	-
Vis 100 °F		0.99	1.75	3.45	8.48
(cst) 210°F	-	0.57	0.86	1.31	2.39
FREEZE . F	-90	-90	-48.i `	+6.8	•
FIA Saturates	82.6	68.8	67.5	76.4	•
Olefins	0.8	1.0	1.6	8.0	
4	16.6	30.2	30.9	15.6	
Arometics SIM-D IBP D-2887	95 <sup>o</sup> f	295°F	342 <sup>0</sup> F	475°F	499 °F
2%	140	311	356	484	530
10%	194	325	385	505	572
50%	272	351	447	541	597
90%	314	396	495	579	754
98%	355	439	513	588	818
EP	445	466	521	598	867

# TODD PSEUDOCOMPONENTS

# GC-1 ONSPEC

	18P-290°F	290-400°F	400-510°F	510-570°F	570+ <sup>®</sup> F
Wt% yleid	15.33	25.61	37.58	15.40	6.08
"API	56.6	44.3	38-8	38.6	40.3
RVP	1.9 lb	0.5 lb	0.0 1ь	-	_
Vis 100°F (cst) 210°F	1.02 0.61	1.99 0.93	3.22 1.24	5.37 1.83	-
FREEZE , F	-90	-90	-40	+10.3	+29.3
FIA Saturates	83.8	73.7	73.5	81.2	•
Olefins  Aromatics	1.1 15.1	1.0 25.3	1.4 25.1	2.3 16.5	•
SIM-0 IBP D-2887	97 <sup>0</sup> F	245 <sup>0</sup> F	389 <sup>°</sup> F	447 <sup>°</sup> F	520°F
2%	140	271	401	476	552
10%	193	300	419	493	567
50%	259	357	457	529	586
90%	311	400	505	563	606
98%	343	415	523	575	729
EP	381	427	544	583	759

### DISCUSSION OF RESULTS

All objectives of the run were met. Most important, under relatively mild conditions of Test \$2, the shale oil raffinate was successfully treated to <1 ppm nitrogen and sulfur. Thus contaminant-free raffinate hydrotreats much easier than did the raffinate from the M-Series runs.

It was planned to stop increasing reactor severity when the product nitrogen and sulfur levels reached <1 ppm, but data turnaround was such that the minimum severity required was exceeded inadvertently.

### CONCLUSIONS

All four objectives for the run were met.

- The least severe conditions for hydrotreating GC-1 shale oil raffinate to <1 ppm sulfur and nitrogen were 650°F; 1 LHSV and 1000 psig at 4000 SCFB hydrogen circulation.
- The reactor was operated at constant conditions for 3-1/2 days and operated for over 6-1/2 days during the run.
- Sufficient "ON SPEC" GC-1 shale oil was prepared for later processing tests and final samples.
- The uncontaminated shale oil raffinate was successfully hydrotreated to <1 ppm sulfur and nitrogen at a
  much lower severity than the contaminated M-Series
  raffinates required.

### SECTION VIII

### FREEZE POINT MODIFICATION

Yet another unique application of the EXTRACTACRACKING process is the ability to selectively modify the freeze point of shale oil derived jet fuels. While freeze point can be attained in some cases by distillation corrections, modification of the freeze point allows an increased yield of jet fuel under otherwise constant conditions.

Studies performed during Phase II encompassed a simulated prescreening, processing of two M-Series samples at conditions defined by the prescreening, and an accelerated aging run to determine if the high endpoint feedstock would cause excessive catalyst deactivation.

Products from these studies were fractionated into appropriate boiling ranges and the targeted jet fuel products supplied to aromatic saturation.

1. SIMULATION OF FREEZE POINT MODIFICATION BY BLENDS OF JET FUEL AND NORMAL PARAFFINS

### SUMMARY

Studies were performed to demonstrate the effectiveness of freeze point modification on normal paraffins and
normal paraffin-doped jet fuels. Experimentation was performed on cetane (n-hexadecane) for scoping purposes; parameter variation studies provided for a JP-7 fuel doped
with 10 percent dodecane, 10 percent tetradecane and 10

percent hexadecane; and finally both reforming and aromatic saturation were carried out with intermediate and final distillation analysis of the doped JP-7 feedstocks. Freeze point depressions up to 100°F were accomplished and a series of correlative curves developed.

### **OBJECTIVES**

The purpose of this work was threefold: (1) determine the effectiveness of Ashland's freeze point modification process, (2) determine freeze point, product property and product yield response to process parameters, and (3) determine the interactions between freeze point modification and aromatic saturation for JP-4 and JP-8 fractions.

### EQUIPMENT AND EXPERIMENTAL

These experiments were performed in a 316 stainless steel 1" I.D. Universal reactor heated by a radiant heat furnace. The temperatures were controlled through the use of Thermo Electric Selectrol switches and measured by means of a Honeywell temperature indicator and Iron-Constantan thermocouples located throughout the reactor (preheat, catalyst bed, postheat). The feed was metered from a capped (air bleed equalizer) graduate via a Lapp pump. The hydrogen was metered to the reactor through a Brooks continuous flow electronic equalizer. The effluent gas was separated from the liquid product in a Jerguson liquid level gauge and depressured to atmospheric pressure through a Grove

pressure regulator. The effluent gas and any light gases from the liquid (after discharging to atmospheric pressure in a closed system) were passed through two acetone-dry ice traps and then measured by means of a wet test meter. The material collected in the dry ice traps and the liquid product were combined and stabilized to a 5°C overhead to yield a C5+ reformate. Spot gas samples were taken during various periods of the test cycle, submitted for gas chromatographic analysis and the values averaged and normalized to yield a light gas value for each experiment.

The catalyst utilized in this test series was a bimetallic extrudate containing platinum and rhenium on a chlorided alumina. The catalyst was charged to the reactor in a pseudo-logarithmic bed configuration employing six bed sections (catalyst plus 35-ml of tabular alumina) with catalyst concentration increasing as one progresses from the top of the reactor to the bottom. The direction of hydrocarbon and gas flow was downflow. The catalyst was prereduced by treatment with flowing hydrogen (1-2 cu.ft. per hour), starting at room temperature and continuing up to and including operating temperature and pressure.

From this study and correlation, two sets of process conditions were picked to give the highest liquid yield, lowest freeze point and substantial aromatics in the product (greater than 20 weight percent). These two sets of conditions were picked so that extended runs could be made to obtain enough product to fractionate and yield JP-4 and

JP-8 jet fuel fractions for product properties evaluation (yield, freeze point, aromatic content). The JP-4 and JP-8 fractions were then hydrogenated to lower aromatic content and the effect on freeze point and the ease of hydrogenation were evaluated.

### FEEDSTOCKS

Feedstock for the initial cetane study was laboratory grade n-hexadecane (cetane) measuring 98.8 percent saturates by FIA analysis. Properties are shown in Table VIII-1.

Feedstock for all remaining runs was doped JP-7, also shown in Table VIII-1.

### RESULTS

The purpose of this study was to reform a n-paraffin containing petroleum based jet fuel (simulating a shale oil jet fuel fraction) under varying conditions of temperature, pressure and space velocity. This feedstock would have a high freeze point (off specifications), contain little or no aromatics which, upon reforming, would yield a total product meeting freeze point requirements and contain aromatics in the 20-40 weight percent range. Then through interpretation and correlation of this data, some insight could be gained into process conditions which would convert a high freeze point n-paraffin containing jet fuel to a low freeze point aromatic containing jet fuel. From this test series the best range of conditions that produced the improved jet fuel could be determined. Then an extended test

# FREEZE POINT MODIFICATION SIMULATION - FEEDSTOCK PROPERTIES

DOPED		CETANE
70% JP-7 10% Dodeca 10% Tetrad 10% Hexade	ecane	
FREEZE PT.	$^{\circ}$ F = $+9.5^{\circ}$ F	+68 <sup>0</sup> F
s -	97.1	98.8
0 -	1.0	0.4
A -	1.9	0.08
SIM-D CUMU	LATIVE WT.7	
C5-130	- 0	0
130-518 (J		1.7
330-572 (J	P8) - 100	95.3
572+		100.0
IBP 34	49	513
10 3	86	542
	01	•
	16	551
	23	-
	27	556
	41	•
	73	560
	90	-
	46	564
100 56	62	579

run could be made to produce sufficient quantity of product that would be fractionated to yield the JP-4 and JP-8 jet fuel fractions for more detailed analysis. Should one or more of the individual fractions not meet the freeze point specifications (although as total product they may have exceeded freeze point requirements), a study would follow of the hydrogenation of these fractions to convert the aromatics to naphthenes and to determine to what extent the freeze point had been lowered.

The process conditions and results of this study are given in the attached Tables VIII-2 through VIII-5 and Figures VIII-1 through VIII-13. The feedstock utilized in this reforming study was a JP-7 (-46°F freeze point, 2.5 weight percent aromatics) which contained 10 volume percent each of dodecane, tetradecane and hexadecane. The inclusion of 30 volume percent n-paraffins raised the freeze point to +10°F, but had little effect on the distillation range of the original jet fuel. Tables VIII-3 through VIII-5 give the results obtained by reforming this feedstock at three different temperatures (850, 880, 910°F), three different pressures (150, 250, 500 psig) and three different space velocities (2, 4, 8 LHSV). With the liquid product measured and all freeze points, aromatic contents, and SIM-D distillations reported, JP-4 and JP-8 yields and all weight balances were corrected to a 100 weight percent basis.

TABLE VIII-2

# REFORMING OF CETANE

RUN	1	2	3	
Conditions				
Temp. °F	750	825	900	
Pressure PSIG	250	250	250	
LHSV	8	8	8	
н <sub>2</sub> /н с	5/1	5/1	5/1	
Product Analyses				
roddet maryses				
Freeze Pt-OF	51.8	27.5	-8.5	
Saturates	98.3	89.6	72.3	
Olefins	0.4	1.4	1.9	
Aromatics	1.3	9.0	25.8	
SIM-D				
IBP	126	100	74	
10	299	285	208	
30	516	477	363	
50	544	524	486	
70	552	538	530	
90	559	552	556	
EP	568	613	667	
Liquid Recovery Wt.Z Feed	99.4	95.8	85.3	

TABLE VIII-3

# REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL

RUN	66-1	66-2	67-1	<u>67-2</u>
Catalyst - Gms.	35		35 -	<del></del>
Conditions Temp. OF Pressure-PSIG LHSV	850 150 8	910 . 150	880 150	910 150 4
H <sub>2</sub> /H C	5/1			·
Liq. Prod. Analysis Freeze Pt (F) S O A	-46 67.0 2.6 30.4	-71 53.6 4.0 42.4	-82 42.6 6.8 50.6	-90 48.1 1.9 50.0
SIM-D-Wt.7 C5-130 130-518 (JP4) 330-572 (JP8) 572+	1.7 89 89 2.0	83.0 72 3	3 86 76 3	5 83 70 4
Wt. Balance H2+C1+C2 C3+C4 C5+ Total	3.20 4.73 95.83 103.76	3.88 7.30 90.30 101.48	4.70 7.91 88.72 101.33	5.98 8.69 86.85 101.52
Wt. Balance Corrected to 100% H2+C1+C2 C3+C4 C5+ C5+Vol.% Yield (100% Basis)	3.1 4.6 92.3 95.2	3.8 7.2 89.0 90.8	4.6 7.8 87.6 01.0	5.9 8.5 85.6 89.0

#### \_\_\_TABLE VIII-4\_\_\_

# REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL

RUN	60-1	60-2	<u>61-1</u>	61-2	62-1	<u>62.2</u>
Catalyst - Gms.	35		<b>→</b> 17.5		<b>→</b> 17.5	
Conditions Temp. OF Pressure-PSIG LHSV H2/H C	850 250 2 5.1	910 250 2	880 250 4	880 250 8	850 250 8	910 250 8
Liq. Prod. Anal Freeze Pt (OF) S O A	-90 62.0 0.5 37.5	-73 47.6 0 52.4	-72 61.1 0.8 38.1	-62 66.7 2.1 31.2	-45 69.3 3.5 27.2	-85 57.0 2.6 40.4
SIM-D-Wt.% C5-130 130-518 (JP4) 330-572 (JP8) 572+	4 89 69 2	9 83 58 3	5 86 71 3	3 87 82 2	0 91 88 2	5 87 71 2
Wt. Balance H2+C1+C2 C3+C4 C5+ Total	7.04 6.12 87.23 100.39	11.51 10.26 79.06 100.83	5.53 6.46 88.03 100.02	3.67 4.26 91.68 99.61	3.06 2.09 94.23 99.38	5.06 10.16 85.06 100.28
Wt. Balance Corrected to 10 H2+C1+C2 C3+C4 C5+ C5+Vol.% Yield (100% Basis)	7.0 6.1 86.9 86.5	11.4 10.2 78.4 78.0	5.5 6.5 88.0 86.7	3.7 4.3 92.0 92.2	3.1 2.1 94.8 95.1	5.0 10.1 84.9 83.7

TABLE VIII-5

# REFORMING OF n-PARAFFIN-DOPED JP-7 JET FUEL

RUN	<u>63-1</u>	<u>63-2</u>	64-1	64-2	65-1	<u>65-2</u>
Catalyst - Gms.	35 -		17.5 -	······································	17.5 -	
Conditions						
Temp. <sup>O</sup> F Pressure-PSIG	850 500	910 500	880 500	880 500	850 500	910 500
LHSV	2	2	4	8	8	8
H <sub>2</sub> /H C	5/1 —				<del></del>	<del></del>
Liq. Prod. Analy Freeze Pt (OF)	<u>rsis</u> -90	<del>-</del> 90	-79	-78	-45	-90
S	76.1	56.6	65.4	68.0	73.0	60.1
0 <b>A</b>	1.5 22.4	1.6 41.8	1.4 33.2	1.9 30.1	2.3 24.7	2.4 37.5
SIM-D-Wt.%		72.0	33.1		<b></b> + • •	37.3
C5-130	6	10	8	6	1	7
130-518 (JP4) 330-572 (JP8)	91 70	87 49	88 57	89 73	91 90	87 65
572+	0	0	1	1	0	1
Wt. Balance	4 60	. 70		2 01	0.00	/ 00
H <sub>2</sub> +C <sub>1</sub> +C <sub>2</sub> C <sub>3</sub> +C <sub>4</sub>	4.60 7.58	6.79 16.80	6.20 7.90	3.81 5.69	2.83 4.27	4.20 9.58
C5+	90.04	74.37	86.32	93.26	95.72	84.12
Total Wt. Balance	102.22	97.96	100.42	102.76	102.82	97.90
Corrected to 100	)%					
H2+C1+C2	4.5	6.9 17.1	6.2	3.7	2.7	4.3
C3+C4 C5+	7.4 88.1	76.0	7.8 86.0	5.5 90.8	4.1 93.2	9.8 85.9
C5+Vol.% Yield (100% Basis)	91.5	75.1	87.0	93.2	95.5	88.0

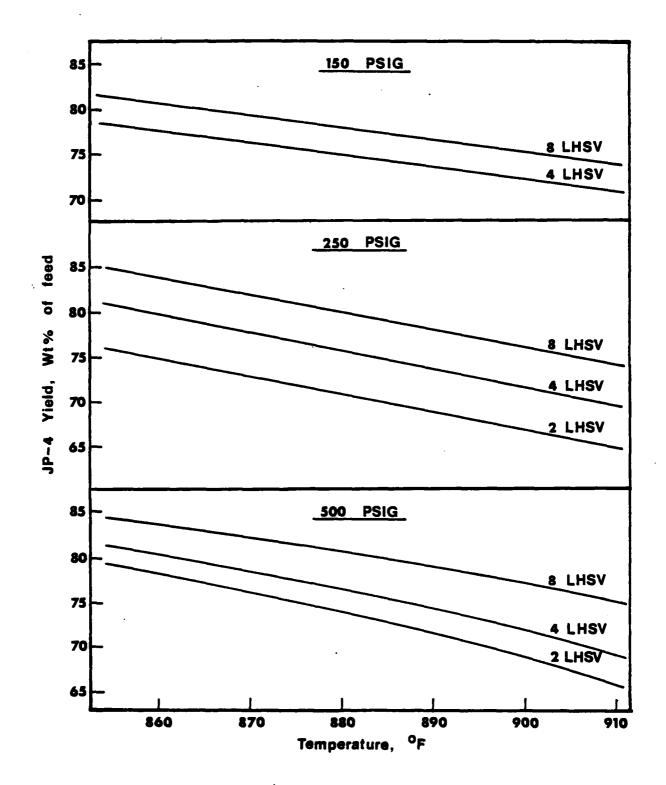


Figure VIII-1. Effect Of Temperature On JP-4

Yield During Reforming Of n-Paraffin-Doped JP-7

Jet Fuel

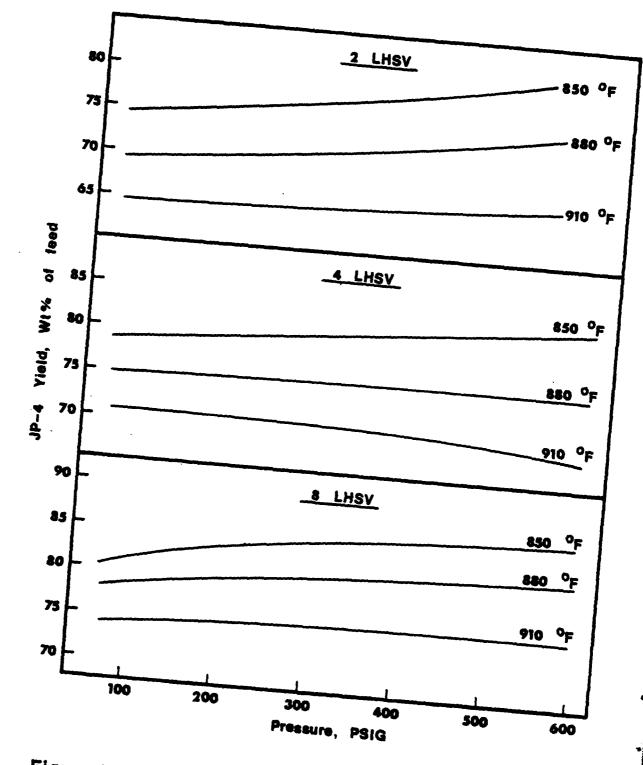


Figure VIII-2. Effect Of Pressure On JP-4 Yield

During Reforming Of n-Paraffin-Doped JP-7

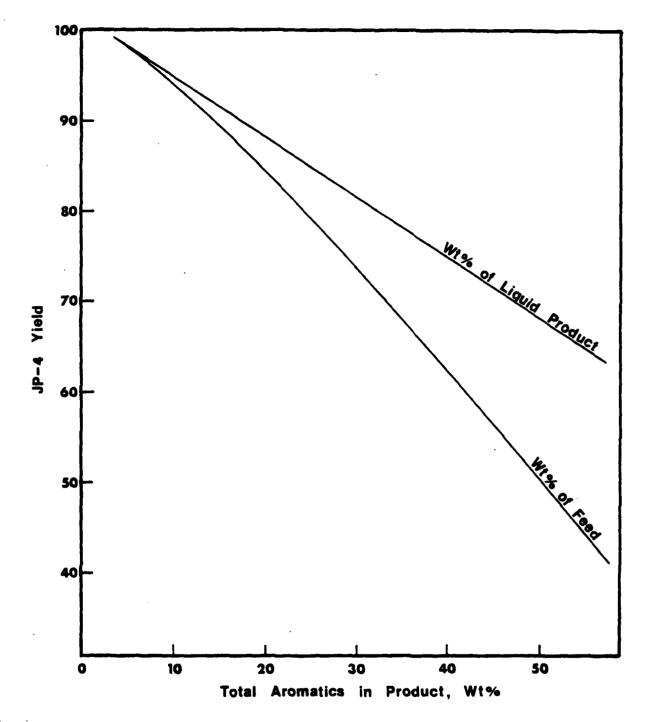


Figure VIII-3. Dependence Of JP-4 Yield - As Both
Wt. % Of Liquid Product And As Wt. % Of Feed - On
Product Aromatic Content During Reforming Of
Paraffin-Doped JP-7 Fuel

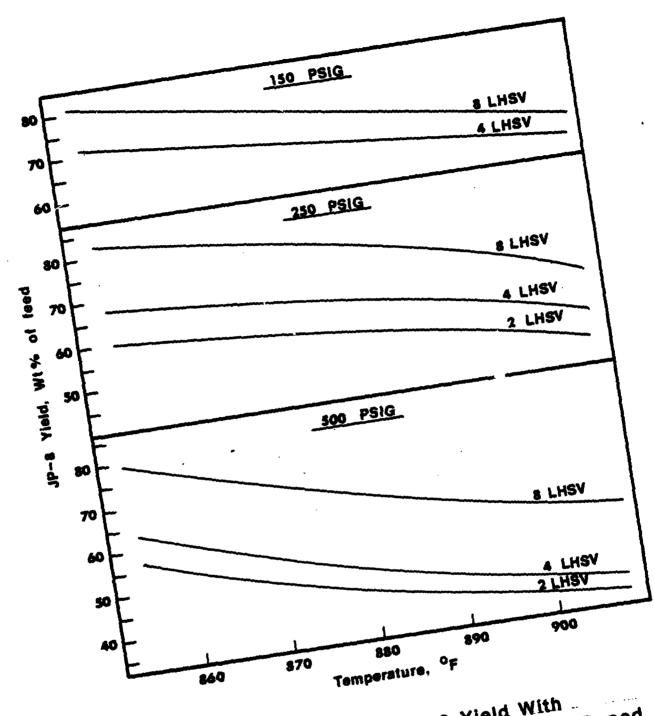
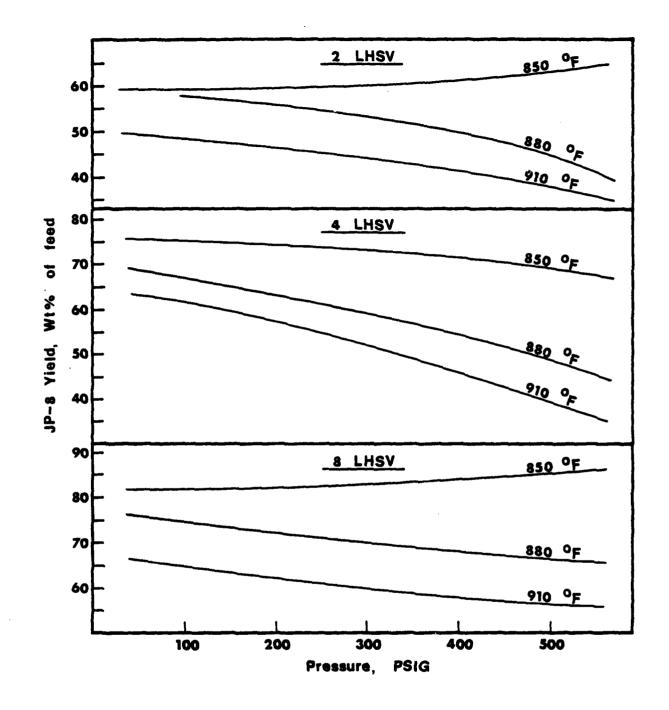


Figure VIII-4 Variation Of JP-8 Yield With

Temperature During Reforming Of n-Paraffin-Doped

JP-7 Jet Fuel



\_\_ Figure VIII-5. Variation Of JP-8 Yield With Pressure \_\_ During Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

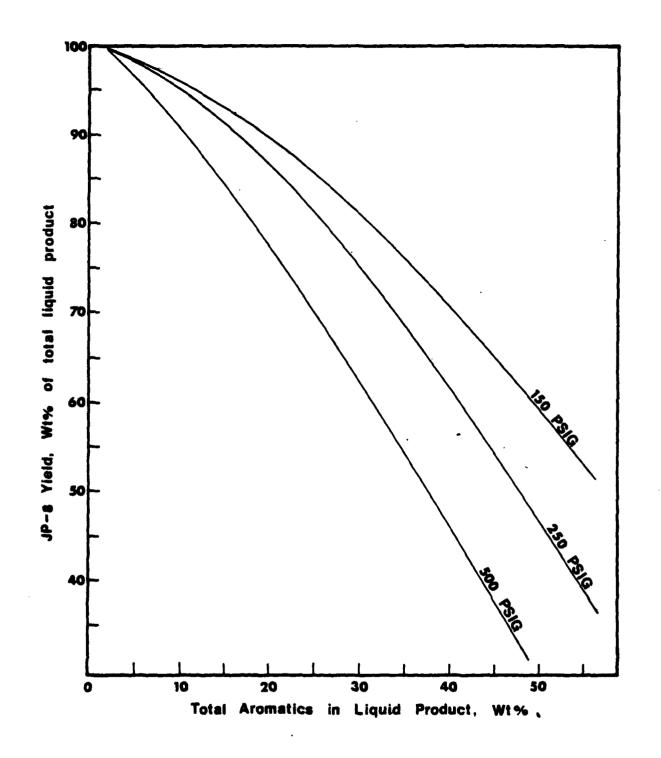


Figure VIII-6. Dependence Of JP-8 Yield - As Wt. %
Of Liquid Product - On Product Aromatic Content During
Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

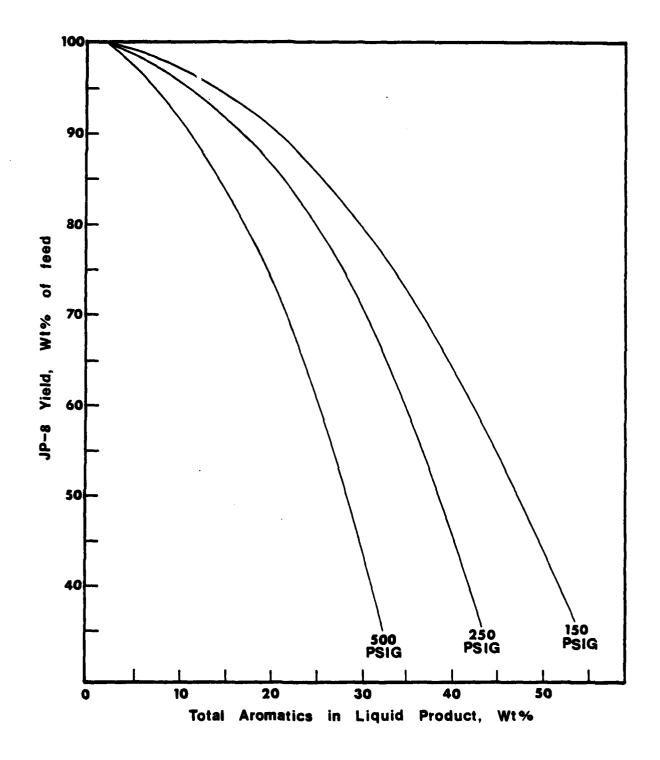


Figure VIII-7. Dependence Of JP-8 - As Wt. % Of Feed - On Product Aromatic Content During Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

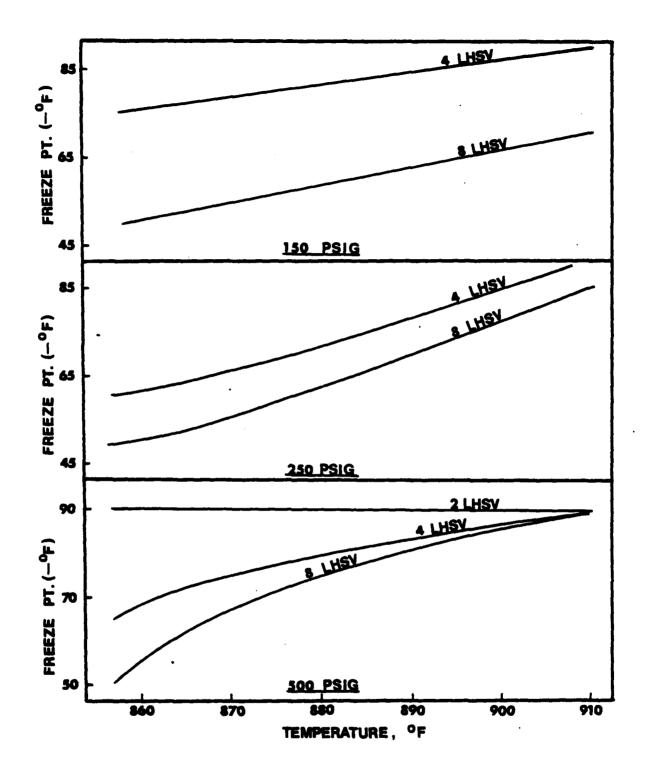


Figure VIII-8. Effect Of Temperature On Product Freeze Point During Reforming Of n-Paraffin-Doped JP-7

Jet Fuel

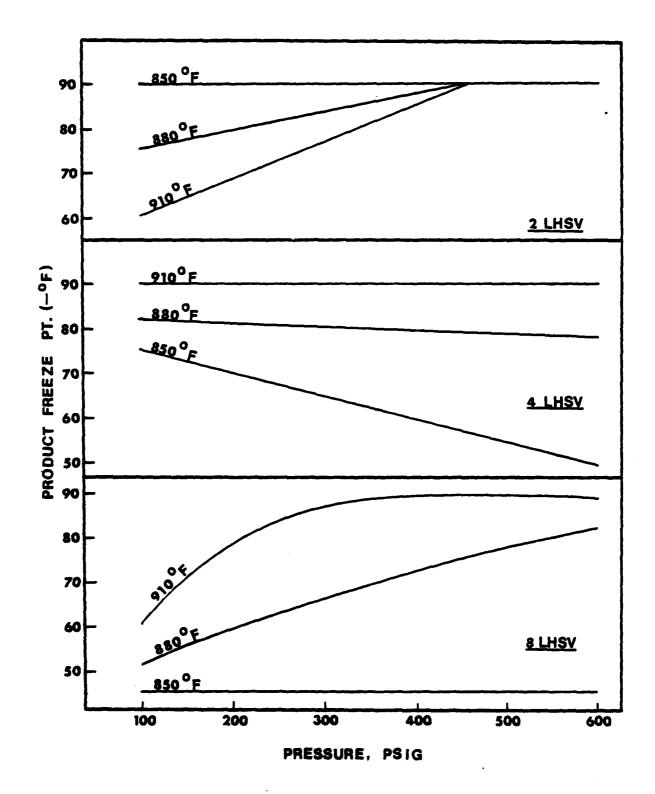


Figure VIII-9. Effect Of Pressure On Product Ereeze
Point During Reforming Of n-Paraffin-Doped JP-7
Jet Euel

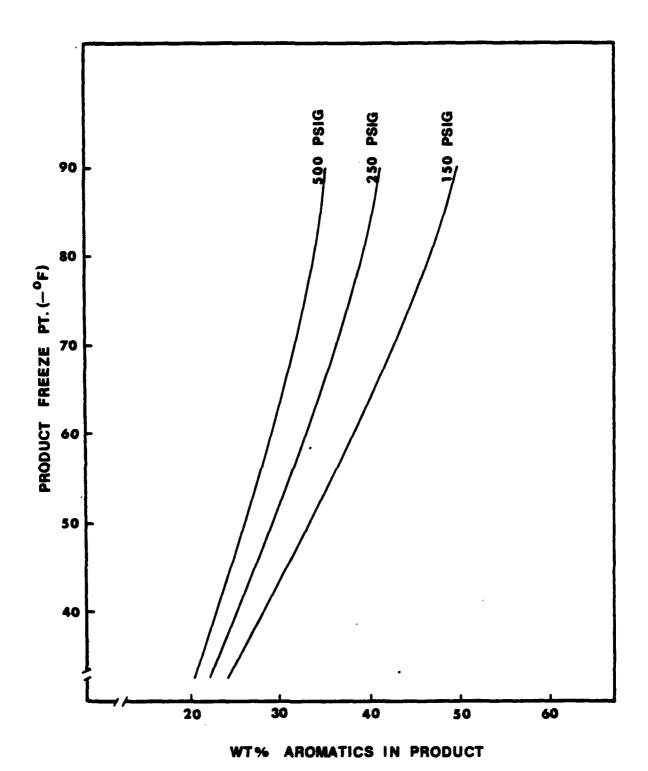


Figure VIII-10. Dependence Of Product Freeze
Point On Product Aromatic Content During
Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

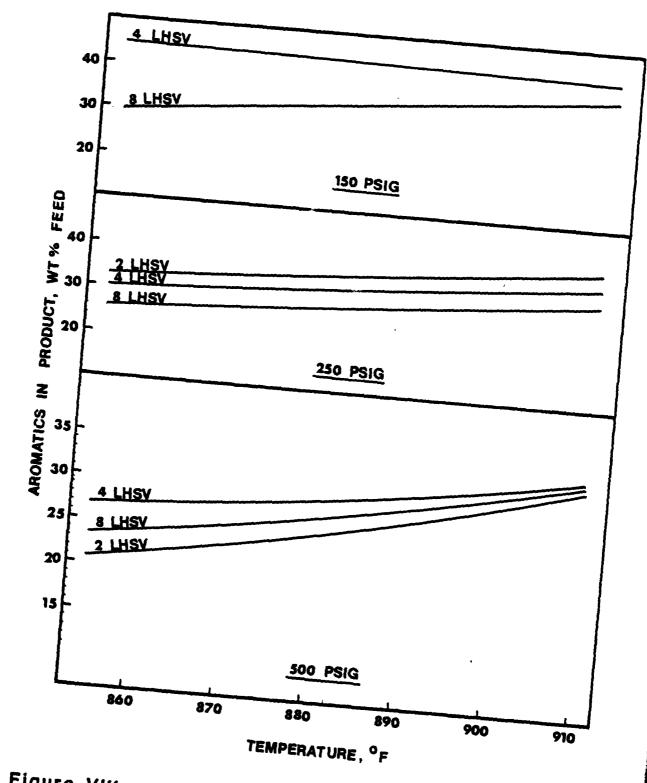


Figure VIII-11. Effect Of Temperature On Product Aromatic Content - As Wt. % Of Feed - During Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

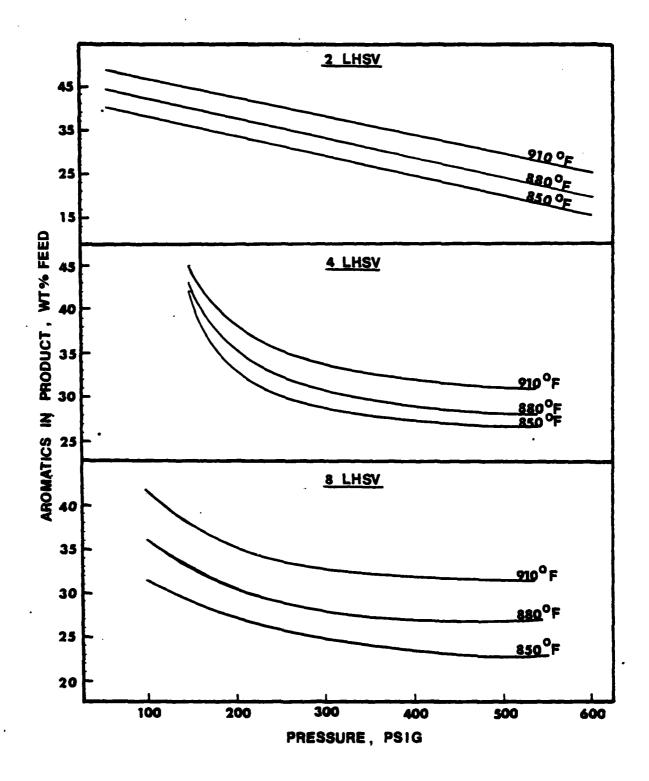


Figure VIII-12. Effect Of Pressure On Product Aromatic Content - As Wt. % Of Feed - During Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

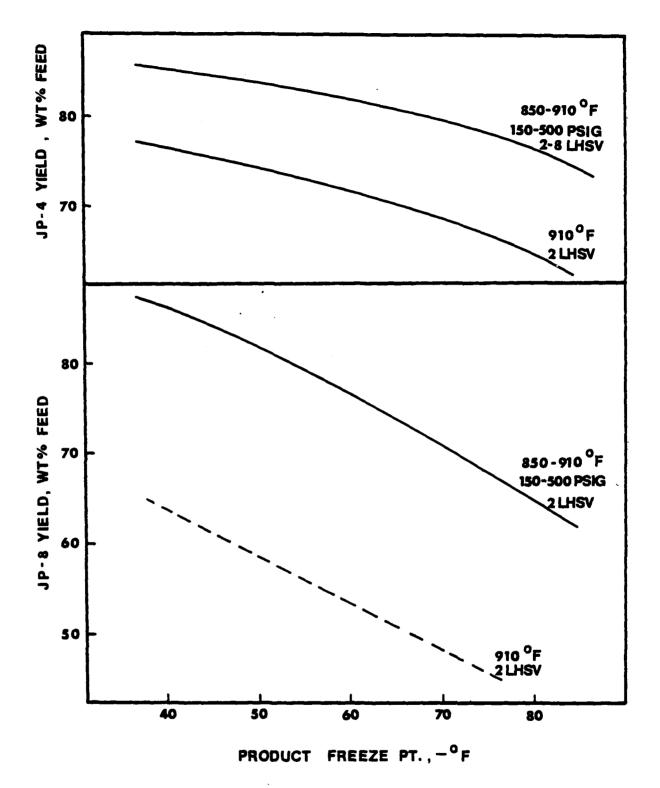


Figure VIII-13. Variation Of JP-4 And JP-8 Yield As Wt. % Of Feed - With Product Freeze Point
During Reforming Of n-Paraffin-Doped JP-7 Jet Fuel

The reforming of a n-paraffin containing jet fuel at two sets of process conditions was also accomplished. The results are given in the attached Tables VIII-6 and VIII-7. Table VIII-6 lists the results obtained by reforming at 850°F, 500 psig, and 8 LHSV over a Pt/Re catalyst. The properties (freeze point, aromatic content) of the reformate were measured and the total reformate then fractionated to yield a JP-4 jet fuel (130-518°F) and a JP-8 fuel (330-527°F). The properties of each fraction were measured. The jet fuel fraction was then subjected to mild hydrogenation and the properties remeasured.

A second reforming test (Table VIII-7) was performed at 880°F, 250 psig, and 8 LHSV over the same catalyst and the same sequence of property measurement and hydrogenation was followed as outlined above. The best results considering yield of total product, aromatic content, and lowered freeze point for both JP-4 and JP-8 fractions were obtained at the second set of conditions listed in Table VIII-7. The JP-4 fraction contained 34 weight percent aromatics and had a freeze point of -78°F which, upon mild hydrogenation, yielded a JP-4 fraction containing 21 weight percent aromatics with a freeze point of -70°F. The JP-8 fraction contained 45 weight percent aromatics and had a freeze point of -67°F which, upon mild hydrogenation, yielded a fraction containing 29 weight percent aromatics with a freeze point of -65°F.

TABLE VIII-6

# REFORMING OF JET FUEL CONTAINING n-PARAFEIN

Run 416	<u>68</u>	<u>68A</u>	<u>74</u>	68B	<u>75</u>
<u>Operation</u>	Reforming	Distillation of Product from Run 68 to Yield JP-4	Hydrog. of 68A	Distillation of Product from Run 68 to Yield JP-8	Hydrog. of 68B
CONDITIONS					
Temp. OF	850	130-518°F	500	330-572 <sup>0</sup> F	500
Pressure PSIG	500	130-318 F	200	330-372 F	200
LHSV	8		2		2
н <sub>2</sub> /н'с	5/1	•	10/1		10/1
PRODUCT					
Freeze Point oF	-58	-61.6	-59.8	~50.8	-54.4
<u>FIA</u>					
\$	74.9	72.0	98.3	69.5	86.9
0	1.1	1.9	0.7	1.4	0.8
<b>A</b>	24.0	26.1	1.0	29.1	12.3
CORRECTED TO 100%					
WT. BALANCE-WTZ FEEL	<u> </u>				
$H_2 - C_1 - C_2$	4.4				
c <sub>3</sub> - c <sub>4</sub>	6.2				
c <sub>5</sub> +	89.4				
C, + VOLZ YIELD  CORRECTED TO 100%	<b>90.3</b>		106		106

TABLE VIII-7
REFORMING OF JET FUEL
CONTAINING n-PARAFFIN

Run 416	<u>69</u>	<u>69A</u>	<u>76</u>	<u>698</u>	<u>77</u>
<u>Operation</u>	Reforming	Distillation of Product from Run 69 to Yield JP-4	Hydrog. of 69A	Distillation of Product from Run 69 to Yield JP-8	Hydrog. of 698
CONDITIONS					
Temp. OF	880	130-518 <sup>0</sup> P	500	330-572 <sup>0</sup> F	. 500
Pressure PSIC	250	. 130-318 F	200	330-372 F	200 .
- LHSV	8		2	•	2
H <sub>2</sub> /H'C	5/1		10/1		. 10/1
PRODUCT					
Freeze Point <sup>O</sup> F	-74.2	-77.8	-70.6	-67	-65.2
FIA					
S	66.4	63.3	77.2	\$0.9	70.2
0	2.2	2.4	1.3	4.3	0.8
A	31.4	34.3	21.5	44.8	29.0
CORRECTED TO 1002					
VT. BALANCE-WTZ FEEL	2				
$H_2 - C_1 - C_2$	3.6	,			
c <sub>3</sub> - c <sub>4</sub>	5.2				
c <sub>5</sub> +	91.2		•		
C + VOLZ YIELD  CORRECTED TO 100Z	93		103		103

#### DISCUSSION

The reforming carried out in this study was atypical reforming or high-endpoint reforming since the IBP of the feedstock used in this study is the endpoint for typical naphthas processed in petroleum catalytic reformers. Many of the correlations follow the typical graphs that are representative for petroleum reforming. However, there are a few isolated areas that appear abnormal. These can be explained on the basis of: (1) the data developed for these points may be erroneous, however, it should be remembered that the data correlated in other areas, or (2) atypical reforming reactions were occurring due to the high boiling character of the feed; these reactions would include partial dehydrogenation of the bicyclic naphthenes, dehydrogenation of one ring of a bicyclic followed by hydrocracking of the other naphthenic ring or vice versa to yield an alkylbenzene, dehydrocyclization of n-paraffins to yield a bicyclic aromatic or naphthene, dehydroisomerization of a naphthene to an indan, hydrocracking of a hydrindane to yield an alkylcyclopentane, selective crackingisomerization of the n-paraffins, etc. All of these, plus other reactions, could lead to the abnormalities seen in some of the graphical interpretation of these data.

The main purpose of this work, as mentioned earlier, was to define a set or sets of conditions for reforming a n-paraffin containing petroleum based jet fuel to yield

sufficient quantities of product for more detailed analysis. This large sample of reformate would be fractionated to yield JP-4 and JP-8 jet fuel fractions. These two fractions would contain a high percentage of aromatics (20-30 weight percent versus less than 5 percent in JP-7) and, hopefully, have a freeze point below -45°F. If these fractions of jet fuel do not meet freeze point specifications while containing a high concentration of aromatics, then hydrogenation of the aromatics to naphthenes would be utilized to attain this end. It should be remembered that in the case of the reforming step an attempt is made to maximize the selective cracking and isomerization properties of the catalyst for the conversion of n-paraffins in addition to its cyclization properties.

The parameters that will determine the best process conditions from this study and correlation are as follows: maximize JP-8 yield, achieve a freeze point below -45°F, and maintain an aromatic content in the product of 20-30 weight percent. Maximum JP-8 yield was determined as follows:

FIGURE	TEMPERATURE °F	PRESSURE PSIG	LHSV	JP-8 YIELD
IX-4	850	150	8	82
IX-4	850	250	8	83
IX-4	850	500	8	84
IX-5	Substanti	lates above	values	

JP-8 yield versus weight percent aromatics in product (Figure VIII-12) indicates the following range of conditions to yield 20-30 weight percent aromatics: 150-250 psig at 8 LHSV. The process conditions required to yield a freeze point of -45°F or lower are found in Figure VIII-8. A broad range of conditions exist that yield a freeze point of -45°F or lower, thus this parameter will be correlated with the aromatics and JP-8 yield. An examination of the data indicated five sets of conditions exist that deserve consideration in the EXTRACTACRACKING scheme and two sets were selected for further study as indicated below.

	CONDITIONS		JP-8		PERCENT
TEMPERATURE °F	PRESSURE	LHSV	YIELD WT. % FEED	F.P.	AROMATICS IN PRODUCT
850	500	8	84	-45	21
880	500	8	66	<del>-</del> 75	29
850	250	8	83	-45	24
880	250	8	75	-62	30
850	150	8	82	-46	30

The two sets of conditions selected for further study are as follows:

850°F - 500 psig - 8 LHSV

In the PV study this set yielded a JP-8 fraction of 84 weight percent containing 20 weight percent aromatics with a freeze point of -45°F

880°F - 250 psig - 8 LHSV

This set was chosen because aromatic content of
product is increased to
30 weight percent and the
freeze point is lowered
to -62°F.

#### 2. M-SERIES

#### SUMMARY

Studies were performed, at single conditions determined from the simulation studies described in Part 1 of this section, on freeze point modification of one JP-4 and one JP-8 sample derived from in situ shale oils. Freeze point reductions of 23 and 12°F were obtained at comparable operating conditions, signifying that the heavier JP-8 fraction may require higher severity operations for comparable freeze point reduction effectiveness.

#### **OBJECTIVES**

Objectives for these studies were to: (1) modify the freeze point of potential Air Force fuel samples, (2) determine the comparison between simulated and actual sample processing response and (3) derive a complete freeze point modification yield structure for each sample.

#### EQUIPMENT AND EXPERIMENTAL

Procedures and equipment utilized for these studies were identical to those described in Part 1 of this section.

#### FEEDSTOCK

Feedstocks for these studies were rehydrotreated samples M-112 and M-121 from the previously described guard-case studies. Guardcase product from M-112 was fraction-ated to a nominal 510°F atmospheric still overhead temperature in preparation for reforming. The M-121 sample was stabilized and reformed as a broad-range (C<sub>6</sub>-600°F) feed. Properties of both the as-charged feedstock and pertinent fractions are shown in Table VIII-8.

#### RESULTS

Pertinent conditions and material balance data are shown in Table VIII-9. Excellent closures, liquid recovery and operating stability were attained.

#### DISCUSSION

These runs demonstrated the feasibility of modifying turbine fuel freeze point by these methods. Surprisingly, these materials demonstrated essentially equivalent C5+ liquid yields. As might be expected, there was a slight hydrogen yield advantage (0.3 weight percent of feed) to the lighter M-112 sample.

Of particular interest to this study, properties of the pertinent fractions for both feed and product are shown

TABLE VIII-8

# MILITARY JET FUEL FROM SHALE OIL SAMPLE PREPARATION SEQUENCE

		M-112			H-121			- <del>2</del> 5	
	ଞା	III.	AR. SAT	81	IR.	AR. SAT.	81	EF.	AL. SAT.
ADDIATICS, VIX		45.8			39.9	20.4	23	42.3	11.9
GLAVITY, OAPI		39.8			37.0	46.6	42.4		44.2
		12.7			12.3	13.9	•		14.2
		-78.7			-46.3	-46.3	-3		-48.1
		İ			350	346			240
		ļ			380	384			300
20		1			396	904	1		330
					456	445	1		403
8	797	1			550	523	į		492
2	<b>79</b>	İ			586	260	ļ		522
D-2867 DIST., OP @ 13P	ļ	٤.			109	254	3		112
10 47 2	;	244			241	340	285		238
20	1	295			302	375	333		287
<b>S</b>	ł	403			430	447	433		387
<b>&amp;</b>	ì	205			199	926	3		<b>5</b> 05
	i	169	619	!	111	199	<b>56</b>		631
VAPOR PRESSURE, PSI	0.85	1			;	{	ļ		i
PLASH POINT, "P		+			.}	i	}		1
HITHOGEN, PPM		ł			i	1	2		1
		ļ			ļ	ł	₽		1
		ł			į	į	77.0		1
		1			}	į	0.3		1

## TABLE VIII-9

# M-SERIES FREEZE POINT MODIFICATION

Sample	M 112	M 121						
OPERA'	OPERATING CONDITIONS							
Temperature, <sup>O</sup> F	850	850						
Pressure, PSIG	500	500						
LHSV, Hr <sup>-1</sup>	8.0	8.0						
H2/HC Molar Ratio	. 5.0	5.0						
MATERIAL BALANCE								
Weig	ht Percent of F	eed						
H <sub>2</sub> , NET	1.42	1.15						
C <sub>1</sub>	0.38	0.38						
C <sub>2</sub>	0.54	0.57						
C <sub>3</sub>	1.03	1.09						
C4	0.87	0.90						
C <sub>5</sub>	0.39	0.54						
Stabilized Liquid	95.52	95.40						
Closure	100.15	100.93						

in Table VIII-8. These properties represent the entire product fraction for M-112 and the nominal 310°F+ fraction of M-121. The JP-4 target (M-112) sample demonstrates a tremendous 23°F freeze point depression with a +20 percent aromatics gain. The resulting freeze point of -79°F is even more impressive in view of the low volatility (0.8 RVP) of the feed sample. The JP-8 target (M-121) demonstrated a lesser freeze point depression (12°F), indicating that these higher boiling materials may require higher severity operations than the JP-4 products. The aromatics gain of +10 percent was also less than the JP-4 sample.

While these data are not of sufficient breadth and quantity to allow correlation, there is an apparent good agreement between these two samples for aromatics and freeze point. As noted above, a 20 percent aromatics gain provided a -23°F freeze point change; the 10 percent aromatics gain provided a -12°F change. On this basis, for these samples, the apparent correlation would then be a freeze point depression of 1.2°F for each 1 percent gain in aromatic content through the reformer. This compares to a 2.4°F lowering of freeze point for each 1 percent gain in aromatics during the simulation studies (without fractionation).

#### RECOMMENDATIONS AND CONCLUSIONS

Freeze point modification was demonstrated for both
 JP-4 and JP-8 turbine fuels.

- Complete module yield structures were defined, with both samples demonstrating >95 weight percent yield.
- Freeze point reduction for actual shale samples, on a
   F per percent aromatics change basis, is roughly half of that demonstrated for simulated (non-fractionated)
   materials.
- Extremely low freeze point for the JP-4 sample was obtained, particularly impressive since the sample also demonstrated low volatility.

#### 3. ACCELERATED AGING

#### SUMMARY

GC-1 guardcase product was reformed under accelerated aging conditions to examine preliminary aging parameters for a selected catalyst system. Results, which would nominally correspond to roughly two months of operation for petroleum naphtha, demonstrated satisfactory stability, although aging was encountered during the last half of the run.

#### **OBJECTIVES**

Objectives for this study were: (1) to examine, on a preliminary basis, initial catalyst deactivation rates and (2) to produce material for further processing.

#### EQUIPMENT AND PROCEDURE

Equipment utilized for this study was identical to that used for Parts 1 and 2 of this study. Experimental

procedure was changed so that a smaller catalyst charge could be used at higher throughputs and effective severity to accelerate the aging process. Conditions selected were:

910°F

500 psig

24 LHSV

5:1 Hydrogen: Hydrocarbon Ratio

Under petroleum naphtha reforming conditions, the severity differential provided by these parameters would correspond to approximately two months of commercial operation after 32 hours on stream for this experiment. This correlation is not, of necessity, valid for the shale feedstock used, but does serve to indicate the magnitude of acceleration involved.

#### FEEDSTOCK

Feedstock for this run was guardcase product from Run GC-1. The full-range guardcase product was fractionated to a nominal 580°F overhead to remove excessive heavy ends which could cause erroneously high deactivation rates in this study. Properties of the feedstock are shown in Table VIII-8.

#### RESULTS

Results for this study are shown in Table VIII-10, as obtained by experimental measurement.

# TABLE, VIII-10

# GC-1 ACCELERATED AGING

Period	1	2	3	4	5	Total		
OP	ERATIN	NG C	ONDIT	IONS				
Temperature, <sup>O</sup> F		910						
Pressure, PSIG			500	0				
LHSV, Hr <sup>-1</sup>	1	24 5.0						
H <sub>2</sub> /HC Ratio								
N	IATERI	AL B	ALAN	CE				
Weight Percent of Feed								
H <sub>2</sub>	1.52	1.49	1.36	1.02	0.99	1.29		
C <sub>1</sub> .	0.78	0.93	1.36	1.69	1.07	1.17		
C <sub>2</sub>	1.84	1.48	2.81	2.88	0.79	2.00		
C <sub>3</sub>	2.89	2.36	0.03	0.75	1.32	1.48		
C 4	1.99							
C <sub>5</sub>	1.13	1.65	2.66	0.67	0.96	1.58		
Stabilized Liquid	94.08	96.13	94.23	94.28	91.49	94.13		
Closure	104.2	105.1	104.5	101.5	97.3	102.7		
Hours on Stream	6	12	18	24	29	-		
Hydrogen Production, SCFB								
Pi	RODUC	T PR	OPER	TIES				
Saturates	54.5	51.1	54.4	58.3	56.3	-		
Olefins	0.6	0.9	0.7	0.5	0.8	-		
Aromatics	44.9	48.0	44.9	41.2	42.9			

#### DISCUSSION

Some deactivation was observed during the run. Figure VIII-14 presents (normalized) data for hydrogen production, C5+ liquid yield and liquid product aromatic content. As the aging occurred, hydrogen yields decreased, liquid product yield increased and aromatics content decreased as expected. There may have been some experimental problem during the period ending in hour 24, as hydrogen yield and aromatics content improved during the next period. Unfortunately, the supply of feed was exhausted in hour 29 and no further data could be obtained.

The deactivation rates demonstrated appear to be quite acceptable though higher than encountered in conventional petroleum reforming. Assuming the freeze point/aromatics correlation developed during Part 2 of this section pertains, the loss in freeze point depression (hour 6 to hour 29) is only 1°F.

#### RECOMMENDATIONS AND CONCLUSIONS

- Aging behavior of the freeze point modification module was shown to be moderate and gradual.
- Sample was produced for further processing.

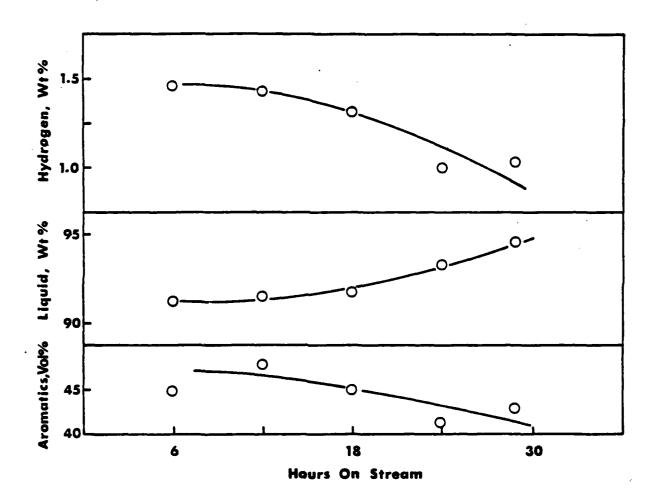


Figure VIII-14. Accelerated Aging Response Of Sample GC-1 (Normalized Data)

#### SECTION IX

#### AROMATIC SATURATION

#### SUMMARY

Three potential turbine fuel samples were processed through an aromatic saturation unit to produce aromatic specification materials. For feed aromatic contents of 45.8, 39.9 and 42.3 volume percent, aromatic saturation products of 18.3, 20.4 and 11.9 volume percent were obtained. Freeze point changes were minor, ranging from +1 to -3°F. Some loss of catalyst activity was noted in the longer runs.

#### OBJECTIVES '

These studies were designed to: (1) determine the effectiveness of Ashland's aromatic saturation process, (2) determine freeze point, product property and product yield response to process parameters and (3) determine the interactions between freeze point modification and aromatic saturation for JP-4 and JP-8 fractions.

#### EQUIPMENT

These experiments were performed in a 316 stainless steel 1" I.D. Universal reactor heated by a radiant heat furnace. The temperatures were controlled through the use of Thermo Electric Selectrol switches and measured by means of a Honeywell temperature indicator and Iron-Constantan thermocouples located throughout the reactor (preheat, catalyst bed, postheat).

The feed was metered from a capped (air bleed equalizer) graduate via a Lapp pump. The hydrogen was metered to the reactor through a Brooks continuous flow electronic equalizer. The effluent gas was separated from the liquid product in a Jerguson liquid level gauge and depressured to atmospheric pressure through a Grove pressure regulator. The effluent gas and any light gases from the liquid (after discharging to atmospheric pressure in a closed system) were passed through two acetone-dry ice traps and then measured by means of a wet test meter. The material collected in the dry ice traps and the liquid product were combined and stabilized to a 5°C overhead to yield a C5+ reformate. Spot gas samples were taken during various periods of the test cycle, submitted for gas chromatographic analysis and the values averaged and normalized to yield a light gas value for each experiment.

#### FEEDSTOCKS

The second secon

Feedstock for these runs were reformer products from runs M-112, M-121 and GC-1. All except M-121 were whole reformer product, while M-121 was a nominal 310°F+ reformate. Properties of these materials are shown in Table VIII-8.

#### **PROCEDURE**

The Universal 1" I.D. reactor was packed with 35 grams (50-ml) of Pt/Al catalyst in a normal downflow pseudolog-arithmic manner. Tabular alumina was used for preheat and postheat sections.

Hydrogen was allowed to flow over the catalyst overnight while the temperature was maintained at 750°F. After reduction of the catalyst, feed was introduced and the reaction zone was controlled at 500°F. Hydrogen flow was maintained at 5.2 SCFH. After a two hour lineout period, product was collected and data taken for eight hour material balance periods until all feed material was used. Refractive index was monitored to confirm activity. The conditions of the runs were as follows:

Temperature: 500°F

Pressure: 200 psig

LHSV: 2.0 (100cc/hr)

H<sub>2</sub>: 10/1 (5.2 SCFH)

#### RESULTS

The M-Series samples were saturated at the conditions outlined in the procedure. An average material balance closure for the five periods was 99.69 weight percent overall and a liquid yield of 103.6 volume percent was obtained. Hydrogen consumption was 906.5 SCFB for Run M-112 and 837 SCFB for Run M-121. Freeze point was lowered from -78.7 to -81.4°F for M-112 with no change occurring for M-121. The GC-1 run consumed an average of 780 SCFB H2 and its freeze point was lowered from -48 to -48.1°F. Analytical results are given in Table IX-1.

TABLE IX-1

# ANALYTICAL RESULTS FOR AROMATIC SATURATED PRODUCTS

SAMPLE	M-112	M-121	GC-1				
ANA	LYTICAL F	RESULTS					
API GRAVITY	46.7	46.6	44.2				
SATURATES, Vol%	80.3	78.6	87.9				
OLEFINS, Vol%	1.4	1.0	0.2				
AROMATICS, Vol%	18.3	20.4	11.9				
HYDROGEN, Wt%	14.6	13.9	14.2				
FREEZE POINT, OF	-81.4	-46.3	-48.1				
D	6 DISTILL	ATION					
Wt%	°F	°F	°F				
IBP	218	346	240				
10	287	384	300				
20	314	400	330				
50	379	445	403				
90	454	524	492				
EP	490	560	522				
D2887 DISTILLATION							
Wt%	0F	-OF	O <sub>F</sub>				
IBP	151	254	112				
10	244	340	238				
20	290	375	287				
50	391	447	387				
90	482	556	504				
EP	619	661	631				

#### DISCUSSION

Sample M-112 demonstrated a 60 percent aromatic saturation level with an actual reduction in freeze point. This run was particularly successful in that no measurable gas production was observed, resulting in a total weight gain of product equivalent to the quantity of hydrogen consumed. Results for Sample M-112 are shown in Tables IX-2 and IX-3.

Sample M-121 attained a 49 percent aromatic saturation level at conditions equivalent to that used for M-112. This sample, targeted for JP-8, is significantly heavier than M-112, which probably accounts for the lower saturation level attained. No change in freeze point was noted in the unit, and again, no measurable quantities of gas were produced. Results for Sample M-121 are shown in Tables IX-3 and IX-4.

Sample GC-1 was run for several material balance periods due to the larger quantities of feedstock available. Aromatic saturation was very good at 72 percent, with a 1°F increase in product freeze point. Measurable quantities of gas were, however, produced during this run. Some deactivation of the catalyst was apparent, as shown by the decreases in gravity and hydrogen consumption. A pressure increase may be required for this module to correct this deactivation situation. Results for Sample GC-1 were shown in Tables IX-3 and IX-5.

## TABLE IX-2

# MATERIAL BALANCE RESULTS FOR AROMATIC SATURATION OF SAMPLE M-112

Period	1		2
OPERATING	CON	DITIO	NS
Temperature, <sup>O</sup> F	504		500
Pressure, PSIG	200		200
LHSV, Hr <sup>-1</sup>	2.00		2.03
H <sub>2</sub> Rate, SCFB	8214		8210
MATERIA	L BA	LANCE	<b>:</b>
Weight Pe	rcent of	Feed	
H <sub>2</sub> Consumed	1,70		1.70
C <sub>1</sub>	-		-
C <sub>2</sub>	-		
C <sub>3</sub>	-		
C4			-
C <sub>5</sub>	-		-
Stabilized Liquid	98.25		101.52
Closure	97.01		99.86
Hdea a a			
Hydrogen Consumption, SCFB	911		904
PRODUCT	PROPE	RTIE	S
OAPI	52.1		51.5

TABLE IX-3

### PERTINENT JET FUEL SPECIFICATIONS FOR AROMATIC SATURATE SAMPLES

	<u>M-112</u>	<u>M-121</u>	<u>GC-1</u>
AROMATICS, WIZ	18.3	20.4	11.9
GRAVITY, OAPI	46.7	46.6	44.2
HYDROGEN, WTZ	14.6	13.9	14.2
FREEZE POINT, OF	-81.4	-46.3	-48.1
D-86 DIST, OF @ IBP	218	346	240
10 VOL Z	287	384	300
. 20	314	400	330
50 .	379	445	403
90	454	523	492
EP	490	560	522
D-2887 DIST., OF @ IBP	151	254	112
10 WTZ	244	340	238
20	290	375	287
50	391	447	387
90	482	556	504
EP	619	661	63!

# TABLE IX-4

# MATERIAL BALANCE RESULTS FOR AROMATIC SATURATION OF SAMPLE M-121

Period	1 2		3				
OPERATING CONDITIONS							
Temperature, <sup>O</sup> F	504	506	506				
Pressure, PSIG	200	200	200				
LHSV, Hr <sup>-1</sup>	2.07	2.04	1.80				
H <sub>2</sub> Rate, SCFB	9901	10,226	11,750				
MATERIAL BALANCE							
Weight Percent of Feed							
H <sub>2</sub> Consumed	0.19	0.39	2.28				
C <sub>1</sub>	_	-	-				
C <sub>2</sub>	-	-	-				
<b>□</b>	_	-	-				
C4	_	-					
C <sub>5</sub>	-	-	-				
Stabilized Liquid	101.80	101.94	101.92				
Closure	101.40	101.34	98.83				
Hydrogen Consumption, SCFB	246	246	246				
PRODUCT PROPERTIES							
OAPI	46.1	43.7	43.6				

# TABLE IX-5

# MATERIAL BALANCE RESULTS FOR AROMATIC SATURATION OF SAMPLE GC-1

Period	1	2	3	4	5		
OPERATING CONDITIONS							
Temperature, <sup>O</sup> F	518	520	515	515	515		
Pressure, PSIG	200	200	200	200	200		
LHSV, Hr <sup>-1</sup>	2.04	2.02	2.04	2.00	2.00		
H <sub>2</sub> Rate, SCFB	9800	9900	9730	10,040	10,000		
MATERIAL BALANCE							
Weight Percent of Feed							
H <sub>2</sub> Consumed	1.46	1.18	1.21	0.99	1.04		
C <sub>1</sub>	-	•	-				
C <sub>2</sub>	•	-	-		-		
C <sub>3</sub>	0.29	0.30	0.19	0.05	0.05		
C4	0.85	0.95	0.42	0.26	0.26		
C <sub>5</sub>	1.81	1.27	0.80	0.60	0.60		
Stabilized Liquid	98.57	103.73	100.89	101.50	101.67		
Closure	100.06	104.40	101.10	101.50	101.67		
Hydrogen Consumption, SCFB	966	781	801	655	688		
PRODUCT PROPERTIES							
OAPI	50.9	54.5	56.4	46.5	46.		

#### CONCLUSIONS

- The aromatic saturation module and conditions provided were adequate to produce specification levels of turbine fuel aromatics.
- Limited quantities of gas are produced.
- There is very little impact on freeze point in this module.
- Slightly higher pressures may be required to counteract a slight deactivation trend encountered for Sample GC-1.

